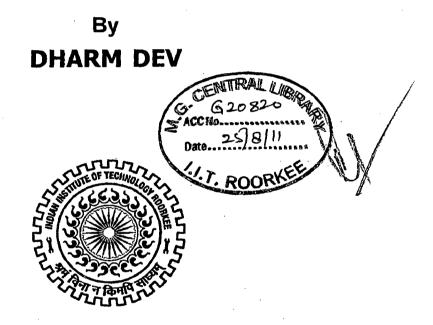
SYNTHESIS AND CHARACTERIZATION OF 2-(SUBSTITUTED ARYL)-3, 3a-DIHYDRO-8H-PYRAZOLO [5,1-a] ISOINDOL-8-ONES VIA CHALCONE BASED N-FORMYL-PYRAZOLINES

A DISSERTATION

Submitted in partial fulfillment of the requirements for the award of the degree of MASTER OF TECHNOLOGY in ADVANCED CHEMICAL ANALYSIS



DEPARTMENT OF CHEMISTRY INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ROORKEE -247 667 (INDIA) JUNE, 2011



Dr. Naseem Ahmed Assistant Professor

INDIAN INSTITUTE OF TECHNOLOGY, ROORKEE DEPARTMENT OF CHEMISTRY ROORKEE-247667, UTTRAKHAND, INDIA Tel: + 91-1332-285745(Off) 285010(Res) E-mail: nasemfcy@iitr.ernet.in

CERTIFICATE

This is to certify that the thesis entitled "Synthesis and characterization of 2-(substituted aryl)-3,3a-dihydro-8H-pyrazolo [5,1-a]isoindol-8-ones via chalcone based N-formyl-pyrazolines" submitted by Dharm Dev, who has registered as M.Tech student in Advanced Chemical Analysis in chemistry department, Indian Institute of Technology Roorkee since July 2009. This work is done by him under my supervision and neither this thesis nor any part of it has been submitted for any degree / diploma or any other academic award any where before.

Date: 27/6/2011

(Dr. Naseem Ahmed) Signature of Supervisor with official seal Dr. NASEEN Frotessor Assistant Protessor Department of Chemistry Department of Technology Indian Institute of Technology ROORKEE



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ROORKEE

CANDIDATE'S DECLARATION

I hereby certify that the progress report presented in the dissertation in the entitled "Synthesis and characterization of 2-(substituted aryl)-3,3a-dihydro-8H-pyrazolo [5,1-a]isoindol-8-ones via chalcone based N-formyl-pyrazolines" for the award of the degree of Master of Technology submitted to the Indian institute of Technology Roorkee is an authentic record of my won work carried out by me during the period from July 2010 to June 2011 under the supervision of Dr. Naseem Ahmed, Assistant Professor, Department of Chemistry, Indian Institute of Technology Roorkee, Roorkee.

The matter presented in this thesis has not been submitted by me for the award of any other degree of this or any other Institute.

Place: Roorkee

Date: 27-6-204

This is to certify that the above statement made by the candidate is correct to the best of our knowledge.

Date: 27/6/2011

HC

Dr. NASEEM AHMED Assistant Professor Department of Chemistry IIT, Roorkee.

Abstract

N-formyl-pyrazoline is the central intermediate in the biogenesis of naturally occurring pyrazoloisoindole derivatives and a useful material in the laboratory synthesis. These derivatives (N-formyl-pyrazoline and pyrazoloisoindoles) were displayed many interesting biological properties, such as hypoglycemic, antimicrobial, amoebicidal, antibacterial, antipyretic, and analgesic activities. An alternate synthetic approach of 2-(substituted aryl)-3,3a-dihydro-8H-pyrazolo [5,1-a] isoindol-8-ones via chalcone based N-formyl-pyrazolines is described. N-formyl-pyrazolines (**1b-15b**) were prepared in excellent yield (81-96%) by the reaction of chalcones (**1a-15a**) with hydrazine hydrate in presence of formic acid, which undergoes intramolecular Friedal-Craft acylation in the presence of trifluoroacetic acid (TFA) as a catalyst to afford functionalized 2-(substituted phenyl)-3,3a-dihydro-8H-pyrazolo[5,1-a]isoindol-8-one (**3c-15c**) in good to excellent yield (74-94%) at refluxed in acetonitrile. Our synthetic route avoids expensive reagents and significantly improved the yield.

Acknowledgement

As I reflect upon the years gone by, I deeply feel the need to acknowledge my gratitude to many wonderful people who have helped me reach this day.

In the first place, I would like to express my deep and sincere gratitude to my supervisors, Dr. Naseem Ahmed for their supervision, advice and guidance from the very early stage of this research as well as giving me extraordinary experiences throughout the work. Above all and the most needed, they provided me unflinching encouragement and support in various ways. Their scientific intuitions have made them as a constant oasis of ideas and passions in science, which inspired and enriched my growth as a student, a researcher and to become a scientist. I am indebted to them for their valuable guidance.

I would like to express my heartiest thankful to Dr. V.K.Gupta, Professor and Head, Department of Chemistry, Indian Institute of Technology, Roorkee for extending various facilities during the course of this work.

I am especially indebted Dr. R. K. Dutta (Coordinator of M.Tech Programme), Assistant Professor, all the faculty members and staff of Department of Chemistry, IIT Roorkee, for their support during the entire period of M.Tech programme.

I wish to express my thanks to my lab mats B. Venkata Babu, Naveen kumar, Praveen, Sunita Dey, friends, the staff of Institute Instrumentation Centre, Central Library and Computer Centre, who are the part to the completion of my work.

Where would I be without my family? In my family special thank to my BADE PAPA Mr. Man Raj and BADI MAMMA Mrs. Prabha Baudh support throughout my life; this work is simply impossible without them. I am indebted to my father, Mr. Radhe Shyam, for his care and love. I cannot ask for more from my mother, Mrs. Chanarama Devi, as she is simply perfect. I have no suitable word that can fully describe them everlasting love to me and I remember, most of all for her delicious dishes.

Last but not least, thanks to God for granting me the faith and will to obtain this academic achievement that has culminated in this thesis.

Place: Roorkee

DHARM DEV M.Tech (A.C.A.)

Department of Chemistry, IIT Roorkee

iii

CONTENTS

Candidate's Declaration i			
Ab	Abstract		
Ac	Acknowledgement		
Co	Contents		
List of Tables		· · ·	ix
Ch	apter No.	Name of chapter	
1		Introduction	2
	1.1	Introduction of pyrazoline and pyrazoloisoindole	2
	1.2	Literature survey	5
		Aim and scope of the work	7
		References	8
2		Synthesis and characterization	12
	2A.	Synthesis of chalcone derivatives	13
	2A.1	Instruments and Chemicals used	13
	2A.2	General procedure for the synthesis of Chalcone	13
	2A.3	Synthesis of chalcone(1a)	14
	2A.4	Synthesis of (E)-3-(4-chlorophenyl)-1-phenylprop-2-en-1-one(2a)	14
	2A.5	Synthesis of (E)-1, 3-bis(4-chlorophenyl)prop-2-en-1-one (3a)	15
	2A.6	Synthesis of (E)-3-(4-bromophenyl)-1-phenylprop-2-en-1-one(4a)	15
	2A.7	Synthesis of (E)-1-(4-bromophenyl)-3-phenylprop-2-en-1-one (5a)	16
	2A.8	Synthesis of (E)-1,3-bis(4-bromophenyl)prop-2-en-1-one (6a)	16
	2A.9	Synthesis of (E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one (7a)	17
	2A.10	Synthesis of (E)-1-(4-bromophenyl)-3-(3,4-dimethoxy- phenyl)prop-2	17
		1-one(8a)	
	2A.11	Synthesis of (E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl)	18
		prop-2-en-1-one(9a)	
	2A.12	Synthesis of (E)-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-	18

	one (10a)	n
2A.13	Synthesis of (E)-3-(4-chlorophenyl)-1-(2-hydroxyphenyl)prop-	19
	2-en-1-one (11a)	
2A.14	Synthesis of (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one	19
	(12a)	
2A.15	Synthesis of (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-	20
	one(13a)	
2A.16	Synthesis of (E)-1-(2-hydroxyphenyl)-3-(4-nitrophenyl)prop-2-	20
	en-1-one (14a)	
2A.17	Synthesis of (E)-1-(4-chlorophenyl)-3-(3-nitrophenyl)prop-2-en-1-	21
	one(15a)	
2B.	Synthesis of N-formyl - pyrazoline derivatives	22
2B. 1	General procedure for the synthesis of N-formylpyrazoline	22
2B.2	Synthesis of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde (1b)	22
2B.3	Synthesis of 5-(4-chlorophenyl)-3-phenyl-4,5dihydropyrazole-	23
	1-carbaldehyde(2b)	
2B.4	Synthesis of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-	23
	carbaldehyde (3b)	
2B.5	Synthesis of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole	24 .
	1-carbaldehyde (4b)	
2B.6	Synthesis of 3-(4-bromophenyl)-5-phenyl-4,5dihydropyrazole-	24
	l-carbaldehyde (5b)	
2B.7	Synthesis of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1-	25
	carbaldehyde (6b)	
2B.8	Synthesis of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-	25
	carbaldehyde (7b)	
2B.9	Synthesis of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-	26
	dihydropyrazole-1-carb -aldehyde (8b)	
2B.10	Synthesis of 3-(4-bromophenyl)-5-(3,4,5-imethoxyphenyl)4,5-	26
	dihydropyrazole-1-carbaldehyde (9b)	
2 B .11	Synthesis of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-	27

.

,

v

	4,5dihydro pyrazole-1-carbaldehyde (10b)	
2B.12	Synthesis of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-	27
	dihydropyrazole-1-carbaldehyde (11b)	
2B.13	Synthesis of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydro-	28
	pyrazole-1-carbaldehyde e (12b)	
2B.14	Synthesis of 5-(3-hydroxyphenyl)-3-phenyl-4,5-dihydro-	28
	pyrazolecarbaldehyde (13b)	
2B.15	Synthesis of 3-(2-hydroxyphenyl)-5-(4-nitrophenyl)-4,5-	29
	dihydropyrazole-1-carbaldehyde (14b)	
2B.16	Synthesis of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydro	29
	pyrazole-1- carbaldehyde(15b)	
2C.	Synthesis of pyrazoloisoindole	30
2C. 1	General procedure for the synthesis of pyrazoloisoindole	30
2C.2	Synthesis of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one	30
	(1c)	
2C.3	Synthesis of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]	31
	isoindol-8-one (2c)	
2C.4	4Synthesis of 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo	31
	[5,1-a]isoindol-8-one (3c)	
2C.5	Synthesis of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]	32
	isoindol-8-one (4c)	
2C.6	Synthesis of 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]	33
	isoindol-8-one (5c)	
2C.7	Synthesis of 6-bromo-2-(4-bromophenyl)-3,3a-	33
	dihydropyrazolo[5,1-a]isoindol-8-one (6c)	
2C.8	Synthesis of 2-(4-bromophenyl)-6-methyl-3,3a-dihydro	34
	pyrazolo[5,1a]isoindol-8-one (7c)	•
2C.9	Synthesis of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydro	34
	pyrazolo[5,1-a]isoindol-8-one (8c)	
2C.10	Synthesis of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3-dihydro	35
	pyrazolo [5,1-a]isoindol-8-one (9c)	

2C.11	Synthesis of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydro pyrazolo[5,1-	35
	a]isoindol-8-one (10c)	
2C.12	Synthesis of 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydro-	36
	pyrazolo[5,1-a]isoindol-8-one (11c)	
[•] 2C.13	Synthesis of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]	37
	isoindol-8-one(12c)	
2C.14	Synthesis of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]	37
	isoindol-8-one(13c)	
2C.15	Synthesis of 2-(2-hydroxyphenyl)-6-nitro-3,3a-	⁻ 38
	dihydro-pyrazolo[5,1-a]isoindol-8-one(14c)	
2C.16	Synthesis of 2-(4-chlorophenyl)-7-nitro-3,3a-	38
	dihydropyrazolo[5,1-a]isoindol-8-one(15c)	
	Results and Discussion	42
3.1	Synthesis and Characterization of synthesis and characterization of	42
5.1	2-(substituted aryl)-3,3a-dihydro-8h-pyrazolo [5,1-a]isoindol-8-	42
	<i>2-(substituted dryt)-3,3d-dinydr0-8n-pyrd2010 [3,1-d]tsolndol-8-</i> ones derivatives	
3.2		42
3.3	Characterization of Compound: 1 Characterization of Compound: 2	42 44
3.4	Characterization of Compound: 3	44 46
3.5 3.6	Characterization of Compound: 4	47
•	Characterization of Compound: 5	48
3.7	Characterization of Compound: 6	49
3.8	Characterization of Compound: 7	50
3.9	Characterization of Compound: 8	52
3.10	Characterization of Compound: 9	53
3.11	Characterization of Compound: 10	54
3.12	Characterization of Compound: 11	56
3.13	Characterization of Compound: 12	57

3

3.14 Characterization of Compound: 13

vii

	3.15	Characterization of Compound: 14	59
	3.16	Characterization of Compound: 15	61
4		Conclusion	73
		Supporting Information	74 ·
		Spectra	79-154

List of tables

Table	No. Name of table	Pase No.
1.	Construction of chalcones by a variety of acetophenones and a variety of	64
·	aromatic aldehydes	
2.	Formation of N-formylpyrazoline from chalcone.	66
3.	Formation of Pyrazoloisoindole from N-formylpyrazoline.	68

. . e.

Chapter - 1

Introduction

CHAPTER 1

INTRODUCTION

1.1 Introduction:

The plant chemistry is a wide and distinct field, which concerned with enormous variety of organic substances accumulated by plants, for example alkaloids, amino acids, pyrazoloisoindole, pyrazoline, etc. In literature, derivatives of nitrogenated heterocyclic aromatics of five members have been described with the inhibition of prostaglandin biosynthesis. Some of these azole derivatives are pyrrols^{1–3}, imidazoles⁴, pyrazoles^{5,6} and pyrazolines, which are pyrazole derivatives^{7,8}. It has been suggested that biological evaluation of new bioactive molecules containing pyrazol nucleus is important for the creation of promising new analgesic agents^{9–10}. Some pyrazoline-derived compounds including dipyrone (metamizol) have been shown to possess analgesic activities mediated by peripheral mechanisms such as inhibitions of cyclooxygenase enzyme activity, arachidonic acid cascade and prostaglandin biosynthesis^{1–8}. However, some other pyrazoline-derived compounds have been reported as centrally acting analgesic agents^{9–11}. Decrease of on/off cell firing in the periaqueductal gray¹⁰, activation of endogen opioid mechanisms about this centrally mediated analgesia.

Pyrazoloisoindolone derivatives (Figure 2) are known for their plant growth regulating properties¹². These structures can be regarded as aza-pyrroloindolones or aza-analogues of tripentones which possess anti-cancer properties and which are widely studied in laboratorys¹³. Pyrazoloisoindole, Pyrazoline and its derivatives have attracted considerable attention due to the wide variety of biological activities (Figure 3). They exhibit hypoglycemic, antimicrobial, amoebicidal, antibacterial, antipyretic, and analgesic activities¹⁴. Specifically, 5-hydroxy-4,5-dihydro-1*H*-pyrazoles are known to possess anti-inflammatory and analgesic activity¹⁵. Pyrazoles exhibit analgesic¹⁶, antimicrobial¹⁷, anti-inflammatory¹⁸, antihypertensive¹⁹ and hypoglycemicactivities²⁰. Also as potential antiprotozoal, cytotoxic agents²¹, and CB₁ cannabinoid receptor antagonists as appetite supp``ressants for the treatment of obesity²². Even today, the pyrazole core concontinues to emerge as a central candidate for pharmaceutical and agricultural applications²³. The pyrazoline function is quite stable, and has inspired chemists to

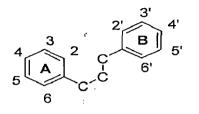
utilize this stable fragment in bioactive moieties to synthesize new compounds possessing biological activities²⁴. Pyrazoloisoindole and pyrazoline derivatives synthesized based on moiety of chalcone.

Chalcones are well known natural compounds and in synthetic chemistry a well known intermediates for synthesizing various heterocyclic compounds bearing the 1, 3- diphenylprop-2-en-1-one framework that have displayed a wide biological activities such as antimicrobial, anti-inflammatory, analgesic, antiplatelet, antiulcerative, antimalarial, anticancer, antiviral, antileishmanial, antioxidant, antitubercular, antihyperglycemic, immunomodulatory inhibition of chemical mediators, release inhibition of leukotriene B_4 , inhibition of tyrosinase and inhabition of aldose reductase activities²⁵⁻²⁹.

The majority of the naturally occurring chalcones contain either hydroxyl (OH) or methoxy (OCH₃) substituents on the two aromatic rings³⁰. However, the number of different chalcones that are theoretically possible in nature is extremely high since each of the groups can be substituted by one or more sugars, which in turn can be acylated with different phenolic of aliphatic moieties. Synthetic and naturally occurring hydroxychalcones and methoxychalcones are of particular interest as they display a wide range of biological properties and exert diverse pharmacological activities. In fact, because of their chemical structures, these compounds can promote both antioxidant and pro-oxidant effects and, as a consequence, have been shown to be effective chemopreventive agents as well as to exert bactericidal, antifungal, anticarcinogenic, and anti-inflammatory actions^{31,32}. The mechanism(s) responsible for this pleiod opism remain to be fully understood, but it is becoming evident that more than one specific cellular target is implicated in the pharmacological actions mediated by chalcones. For instance, the antiinflammatory properties of 4-dimethylamino-3',4'-dimethoxychalcone and 2',5'-dihydroxy-4chloro-dihydrochalcone in murine macrophages were initially reported to involve a direct scavenging effect on superoxide anion production and inhibition of inducible nitric oxide synthase (iNOS) expression^{33,34}.

-3

Many studies of chalcone related to cancer have demonstrated. The presence of a reactive α,β unsaturated keto function in the chalcone is found to be responsible for their microbial activity. The chalcone are acyclic polyphenolic compounds possessing 15carbon atoms(Figure 1) ; two benzene rings joined by a linear three carbon chain³⁵. They constitute one of the most characteristic classes of compounds in higher plants.



The skeleton above, can be represented as the

 $C_6 - C_3 - C_6$ system.

Figure1:

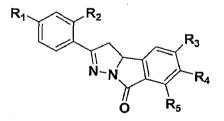
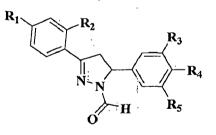


Figure 2: General Structure of Pyrazoloisoindole



 $R_1 = H$, OH, Br, Cl; $R_2 = H$, OH; $R_3 = H$, OCH₃; $R_4 = H$, CH₃, OCH₃, OH, Cl, Br, NO₂; $R_5 = H$, OCH₃, NO₂.

4

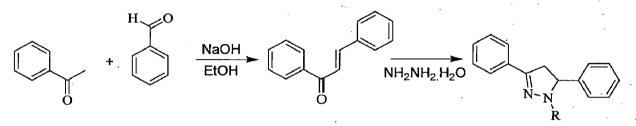
3

Figure 3: General Structure of Pyrazoline



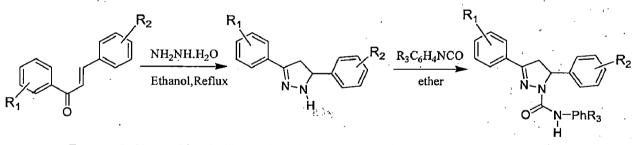
1.2 Literature survey:

Chalcone was synthesized using Claisen-Schmidt reaction. In chalcone, more important moiety is α,β unsaturated ketone, which has more important for biological activity. The α,β -unsaturated ketones can play the role of versatile precursors in the syntheses of the corresponding pyrazoline derivatives³⁶⁻⁴¹. Pyrazoline are very important intermediate in organic chemistry and can serve as versatile precursor in synthesis of many natural products and drug moieties. Some reagent including, hydrazine hydrate, phenylhydrazine, and methyl hydrazine, have used in cyclization of chalcone to form pyrazoline⁴².



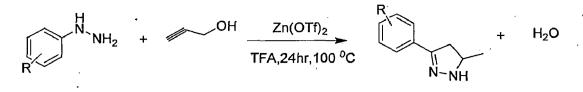
Route: 1 (R=-H, Ph, CH₃)

Rolf van Hes, Kobus Wellinga, and Arnold C. Grosscurt was Synthesis and Insecticidal Properties of 3,5-Diphenyl-1-phenylcarbamoyl-2-pyrazolines. They are Synthesis these product by chalcone derivatives⁴³.



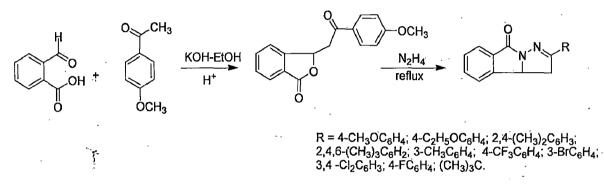
Route: 2 (R_1 = -Cl, -C₂ H_5 , -N(CH₃)₂; R_2 = H, -Cl, CH₂; R_3 =-Cl, -Br, CF₃).

Karolin Alex, Annegret Tillack, Nicolle Schwarz, and Matthias Beller have been developed a method for novel regioselective synthesis of aryl-substituted pyrazolines. Substituted phenylhydrazines react with 3-butynol in the presence of a catalytic amount of zinc triflate to give pyrazoline derivatives. The resulting products are easily oxidized in a one-pot procedure to the corresponding pyrazoles⁴⁴.



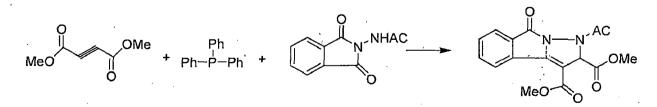
Route: $3(R = Cl, Br, CH_3)$.

Alexander L. Johnson and Philip B. Sweetser have described the synthesis of 2-(4-methoxyphenyl)-8H-pyrazolo[5,l-u]isoindol-8-one from its 3,3a-dihydro derivative, which was prepared from phthalaldehydic acid, 4-methoxyacetophenone, and hydrazine⁴⁵. The reverse cyclisation of 2-[3(5)-arylpyrazol-3(5)-yl] benzoic acid proceeded with thionyl dichloride, phosphoric trichloride or acetic anhydride⁴⁶.





Islami, Mohammad Reza and Abedini-Torghabeh was synthesized fused pyrazole systems as biological important molecules. An efficient two-step protocol has been developed to maked molecules of this family via the reaction between di-Methylacetylenedicarboxylate, triphenylphosphine and N-aminophthalimide in the presence of benzoic acid or N-aminophthalimide derives. This reaction occurred easily at ambient temperature to give stable phosphorus ylides, which upon heating undergoes an intramolecular Wittig reaction to afford the pyrazolo[5,1-a]isoindole derives⁴⁷.

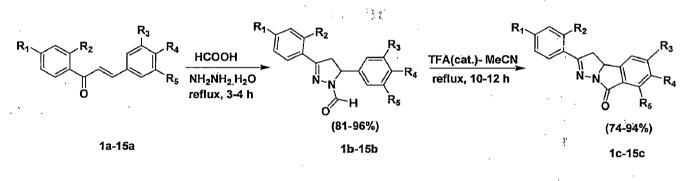




·6

Aim and scope of the present work

Chalcone is a very special α,β -unsaturated ketone system. The α,β -unsaturated ketones play the important role of versatile precursors in the synthesis of the many natural product and biological active compounds. Chalcone are convenient and versatile materials for the synthesis of pyrazolines. Some reagent including, hydrazine hydrate, phenylhydrazine, and methyl hydrazine, have used in cyclization of chalcone to form pyrazoline. The present study deals with the synthesis of N-formylpyrazoline as a new representative of this compound class by reaction of an α,β -unsaturated ketone with hydrazine hydrate and formic acid and report an alternate method for the preparation of functionalized 2-(substituted aryl)-3,3a-dihydro-8H-pyrazolo[5,1-a]iso-indol-8-one (**1c-15c**) via chalcone based N-formyl-pyrazolines using trifluoroacetic acid as catalyst at refluxed in acetonitrile (Scheme 1).



 $R_1 = H$, OH, Br,CI , $R_2 = H$, OH , $R_3 = H$, OCH₃, $R_4 = H$, CH₃ ,OCH₃ , OH , CI, Br. NO₂ , $R_5 = H$, OCH₃ , NO₂.



: : · · ·

7

. .

References:

- 1. W.W. Wilkerson, W. Galbraith, K. Gans-Brangs, M. Grubb, W.E. Hewes, B. Jaffee, J.P. Kenney, J. Kerr, N. Wong, J. Med. Chem. 1994, 37, 988–998.
- 2. W.W. Wilkerson, R.A. Copeland, J.M. Trzaskos, J. Med. Chem. 1995, 38, 389-391.
- I.K. Khanna, R.M. Weier, Y. Yu, P.W. Collins, J.M. Miyashiro, C.M. Koboldt, A.W. Veenhuizen, J. Med. Chem., 1997,40, 1619–1633.
- 4. I.K. Khanna, R.M. Weier, Y. Yu, X.D. Xu, F.J. Koszyk, P.W. Collins, C.M. Koboldt, J.L. Masferrer, P.C. Isakson, J. Med. Chem. 1997,40, 1634–1647.
- 5. K. Tsuji, K. Nakamura, N. Konishi, T. Tojo, T. Ochi, H. Senoh, M. Matsuo, Chem. Pharm. Bull. 1997, 45, 987–995.
- 6. K. Tsuji, N. Konishi, G.W. Spears, T. Ogino, K. Nakamura, T. Tojo, T. Ochi, F. Shimojo, H. Senoh, M. Matsuo, *Chem. Pharm. Bull.* **1997**, *45*, 1475–1481.
- 7. E. Bansal, V.K. Srivastava, A. Kumar, Eur. J. Med. Chem. 2001, 36, 81-92.
- F. Mana, F. Chimenti, A. Bolasco, M.L. Cenicola, M.D.C. Parrillo, F. Rossi, E. Marmo, Eur. J. Med. Chem. 1992, 27, 633–639.
- 9. J. Milano, S.M. Oliveira, M.F. Rossato, P.D. Sauzem, P. Machado, P. Beck, N. Zanatta, H.G. Bonacorso, *Eur. J. Pharmacol.* 2008, 54, 86–96.
- 10. Z. Tabarelli, M.A. Rubin, D.B. Berlese, P.D. Sauzem, T.P. Missio, M.V. Teixeira, A.P. Sinhorin, M.A.P. Martins, J. Med. Biol. Res. 2004, 37, 1531–1540.
- M.C. Godoy, M.R. Fighera, F.R. Souza, M.A. Rubin, M.R. Oliveira, N. Zanatta, M.A. Martins, H.G. Bonacorso, C.F. Mello, *Eur. J. Pharmacol.* 2004, 96, 93–97.
- 12. (a) E. Toja; A. Omodei-Sale; C. Cattaneo,; Galliani, Eur. J. Med. Chem. 1982, 17, 223–227; (b) Johnson, A. L.U.S. Patent 1974, 3, 436; (c) A. L. Johnson; Sweetser, P.B. DE 1972, 2219702; (d) E. M. Beyer; Johnson, A. L.Sweetser, P. B. Plant Physiol. 1976, 57, 839–841; (e) G. F. Katekar; A. E Geissler. Plant Physiol. 1980, 66,1190–1195; (f) G. F. Katekar; Nave; Geissler, E.Plant Physiol. 1981, 68, 1460–1464.
- (a) A. R. Katritzky; Pozharskii, A. F. Handbook of Heterocyclic Chemistry: 2000, 233-235; (b) E. J. Noga; G. T. Barthalmus; M. K. Mitchell Cell Biology Int. Rep. 1986, 10, 239. (c) P. N. Craig In Comprehensive Medicinal Chemistry; C. J. Drayton, Ed.; Pergamon Press: New York, 1991, 8; (d) T. Kodama; M. Tamura, T. Oda; Y.

Yamazaki; M. U.S. Patent 983928, 2003. (e) Padwa, A.; Bur, S. Chem. Rev. 2004, 104, 2401.

- 14. J. Elguero, In Comprehensive Heterocyclic Chemistry; A. R. Katritzky, Ed.; Pergamon Press: New York, 1984, 5, 291-297.
- 15. F. R. de Souza; M. R Fighera; Lima, T. T. F.; de Bastiani, J.; Almeida, C. E.; Oliveira, M. R.; Bonacorso, *BehaVior*, 2001, 68, 525:
- 16. GMenozz.; P. Schenone; L. Mosti; F. Mattioli J. Heterocycl. Chem. 1993, 30, 997.
- 17. S. P. Singh; R. Naithani; O. Prakesh Ind. J. Heterocycl. Chem. 1992, 11, 7.
- 18. L. V. G. Nargund; V. Hariprasad; G. R. N. Reddy J. Pharm. Sci. 1992, 81, 892.
- W. T. Ashton; S. M. Hutchins; W. J. Greenlee; G. J. Doss; S. D Kivlighn.; P. K. Siegl J. Med. Chem. 1993, 36, 3595.
- V. J. Bauer;, H. P. Fanshawe; S. RSafir.; E. C. Tocus; C. R. Boshart J. Med. Chem. 1968, 11, 981 Dalalian.
- 21. N. R. Sperandeo; R. Brun. ChemBioChem; 2003, 4, 69.
- 22. R. P. Nargund; L. H. T. Vander Ploeg; T. M. Fong; D. J. MacNeil; H. Y. Chen; D. J. Marsh; J. U.S. Warmke Pat. Appl. Publ., 2004, 43, 225-227.
- 23. (a) J. Elguero In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V., Eds.; Pergamon-Elsevier Science: Oxford, UK, 1996, 6, 1-75.
 (b) Sutharchanadevi, M.; Murugan, R. In Comprehensive Heterocyclic Chemistry II; Katritzky, Eds.; Pergamon-Elsevier Science: Oxford, UK, 1996, 6, 221- 260. (c) M. J. Graneto; R. G. Kurumbail; M. L. Vazquez; H.-S. Shieh; J. L. Pawlitz; J. M. Williams; M. A. Stallings; R. M. Weier; G. J. Hanson; R. J. Moure; R. P. Compton; S. J. Mnich; G. D. Anderson; J. B. Monahan; R. J. Med. Chem. 2007, 50, 5712. (d) P. Diana.; A. Carbone; P. Barraja; A. Martorana; O. Gia; L Dallavia.; G Cirrincione. Bioorg. Med. Chem. Lett. 2007, 17, 6134. (e) N Gokhan-Kelekci.; S. Yabanoglu; E. Kupeli; U. Salgin; O Ozgen.;, G. Ucar; E. Yesilada; E. Kendi; A. Yesilada; A. A. Bilgin Bioorg. Med. Chem. 2007, 15, 5775. (f) M. Patel and R. Desai,; RKIVOC 2004, 123-129.
- 24. M. A. Oyvind and R.M. Kenneth CRC press, Bocaraton, 2006, 1003.

25. (a) J. N. Dominguez; C. Leon; J. Rodrigues; N. G. de Dominguez; J. Gut; P. J. Rosenthal; J. Med. Chem. 2005, 48, 3654-3657; (b) C. X. Xue; S. Y. Cui; C. Liu; Z.

D. Hu; B. T. Fan; Eur. J. Med. Chem. 2004, 39, 745-747; (c) M. Liu; P. Wilarat; M. L. Go.; J. Med. Chem., 2001, 44, 44-46.

- 26. (a) M. Liu; P. Wilairat; S. L. Croft; A. L.Tan; M. Go; *Bioorg. Med. Chem.*, 2003, 11, 27-29; (b) O. Kayser; A. F. Kiderlen; *Phytother. Res.*, 2001, 15, 148-150.
- C. Q. Meng; X. S. Zheng; J. E. Simpson; K. J. Worsencroft; M. R. Hotema; M. D., J. Med. Chem. 2002, 45, 54-57.
- J. W. Skudlarek; J. M. Gilmore; L. K. Hoong; R. R. Hill; E. M. Marino; K. L. Suen; C. Kunsch; *Bioorg. Med. Chem. Lett.* 2004 ,14, 15-13; (b) M. E. Zwaagstra; H. Timmerman; M. Tamura; T. Tohma; Y. Wada; K. Onogi; M. Q. Zhang; *J. Med. Chem.*, 1997, 40, 1075-1077.
- 29. N. Yayli; F. E. Aydin; Y. Gçk; A. C. Baltasi; N. Yildirim; J. Photochem. Photobiol., 2005, 169, 229-231.
- C. A. Williams, and, R. J. Grayer, Anthocyanins and other flavonoids. *Nat. Prod. Rep.* 2004, 21, 539–573.
- 31. Go, M. L., Wu, X., and Liu, X. L. Chalcones: an update on cytotoxic and chemoprotective properties. *Curr. Med. Chem.* 2005, 12, 481-499.
- 32. Nowakowska, Z. A review of anti-infective and anti-inflammatory chalcones. *Eur. J. Med. Chem.* **2007**, *42*, 125–137.
- F. Herencia, M. L. Ferrandiz, A. Ubeda, I. Guillen, J. N. Dominguez, J. E. Charris, G. M. Lobo, and Alcaraz, ; *Free Radical Biol. Med.* 2001, 30, 43-50.
- 34. Y. C. Huang, J. H. Guh, Z. J. Cheng, Y. L. Chang, T. L. Hwang, C. N. Lin, and C. M. Teng, *Life Sci.* 2001, 68, 2435–2447.
- 35. D. N. Dhar; Wiley, New York, 1981.
- 36. J. Elguero, Bulton and McKillop (editors), Comprehensive Heterocyclic Chemistry, Pergamon Press, 1984, 5, 293.
- 37. D. B. Dambal, P. P. Pattanashetti, R. K. Tikare, B. V. Badami, and G. S. Puranik, *Indian J. Chem.*, 1984, 23B, 186.
- 38. S. P. Sachchar and A. K. Singh, J. Indian Chem. Soc. 1985, 62, 142.
- 39. S. E. Kulkarni, R. A. Mane, and D. B. Ingle, Indian J. Chem. 1986, 25B, 452.
- 40. R. J. Cremlyn, F. J. Swinbourne, and E. Mookerjee, Indian J. Chem. 1986, 25B, 562.
- 41. N. G. Gawande and M. S. Shingare, Indian J. Chem. 1987, 26B, 351.

- 42. R. Nirar, El-Rayyes, George H. Hovakeemian, and Hayat S. Hmoud; J. Chem. Eng. Data, 1984, 29, 225-229.
- 43. R.H., Kobus Wellinga, and Arnold C. Grosscurt; J. Agric. Food Chem. 1978, 26, 4.
- 44. Karolin Alex, Annegret Tillack, and Matthias Beller; Zinc-Catalyzed Synthesis of Pyrazolines and Pyrazoles via Hydrohydrazination; *organic letters*; 2008, 10, 2377-2339.
- 45. A. L. Johnson; P. B. Sweetser; J. Org. Chem. 1976, 41, 110-114.
- 46. M. R. Islami; J. Abedini-Torghabeh; S. J. Fatemi; Z. Hassani; Amiry, Synlett, 2004, 10, 1707–1710.
- 47. M. A. Berghot and E. B. Moawad. Convergent synthesis and antibacterial activity of pyrazoles and pyrazolines of diazepam. *Eur. J. Pharm. Sci.* 2003, 20, 173-179.

Chapter - 2 Synthesis and Characterization ,

2A: SYNTHESIS OF CHALCONE DERIVATIVES

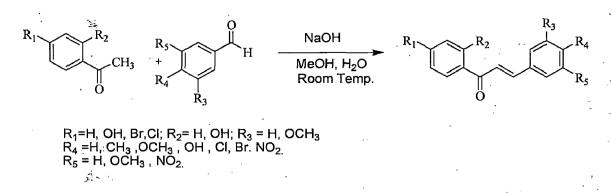
2A.1 Instruments and Chemicals used

Flash chromatography was performed on silica gel (Rankem laboratory, 60-120 mesh). TLC was performed on aluminum-backed silica plates (contain -13% CaSO₄ ½ H₂O, SILICA Gel /UV₂₅₄), which were developed by using UV fluorescence. Melting points were determined on kofler apparatus melting point apparatus and are uncorrected. Elemental analysis was performed on a Vario EL CHNS analyzer. Infrared spectra were recorded on a Nexus Thermo Nicolet FT-IR spectrometer using KBr pellets .¹H NMR spectra were recorded at 500 MHz on Brucker ultra shield AC 300 and DPX 300 instruments, respectively; ¹³C NMR spectra at 125.5 MHz's chemical shifts are given in parts per million (ppm) referenced to TMS. High resolution mass spectra (m/z) were recorded on a Perkin Elamer GC-MS spectrometer. All commercially available chemicals were used with further purification. Anhydrous solvents were distilled from appropriate drying agents prior to use.

2A. 2 General procedure for the synthesis of Chalcones

Different Acetophenone (1 mmol) and different aromatic aldehyde (1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow/white cloudy solution. The reaction mixture was stirred for 2-5 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product.

÷. ;



Scheme 2

2A.3 Synthesis of chalcone(1a)

71

Acetophenone (0.116 mL, 1 mmol) and benzaldehyde (0.101 ml, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white solution. The reaction mixture was stirred for 2 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield=92%, m.p. = 71-73°C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.05-8.03 (m, Ar-H, 2H), 7.83 (d, *J*=11.5Hz, =CH-Ar, 1H), 7.69-7.63 (m, Ar-H, 2H), 7.62-7.57 (m, Ar-H, 1H), 7.55-.51 (m, Ar-H, 3H), 7.47-7.43 (m, Ar-H, 3H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1640, N-H_{bending} 1534, 1203, 1157.47 1085.32, 997.17, 820.12, 760.04, 679.28.

2A.4 Synthesis of (E)-3-(4-chlorophenyl)-1-phenylprop-2-en-1-one(2a)

Acetophenone (0.116 mL, 1 mmol) and p-chlorobenzaldehyde (140.5 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white solution. The reaction mixture was stirred for 4 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2×20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography

is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield=93%, m.p. =111-112^oC. ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.06-8.04 (m, Ar-H, 2H), 7.96 (d, *J*=15.5Hz, =CH-Ar, 1H), 7.83-7.64 (m, Ar-H, 2H), 7.63 (d, *J*=12.0Hz, O=C-CH=, 1H), 7.61-7.54 (m, Ar-H, 2H), 7.46-7.40 (m, Ar-H, 3H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1644, N-H_{bending} 1523, 1206, 1157, 1075, 997, 820, 760, 679.

2A. 5 Synthesis of (E)-1, 3-bis(4-chlorophenyl)prop-2-en-1-one(3a)

4-chloroacetophenone (0.1295 mL, 1 mmol) and 4-chlorobenzaldehyde (140.5 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellowish solution. The reaction mixture was stirred for 6 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield= 91%, m.p. = 78-81°C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.03-7.93 (m, , Ar-H, 2H), 7.80 (d, *J*=12.5Hz,=CH-Ar, 1H), 7.77-7.58 (m, Ar-H, 2H), 7.52-7.50 (m, Ar-H, 2H), 7.46 (d,*J* = 13.0Hz, 1H), 7.44-7.41 (m, Ar-H, 2H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1647, N-H_{bending} 1533, 1206, 1157, 1076, 987, 824, 764, 679.

3A. 6 Synthesis of (E)-3-(4-bromophenyl)-1-phenylprop-2-en-1-one(4a)

Acetophenone (0.1205 mL, 1 mmol) and 4-bromobenzaldehyde (185.03 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow solution. The reaction mixture was stirred for 3 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield = 93%, m.p. = 74-79°C ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.02-8.00 (m, Ar-H, 2H), 7.68 (d, *J*=6.5Hz, =CH-Ar, 1H), 7.62-7.55 (m, 3H), 7.53 (d, *J*=12.0Hz, CO-CH=, 1H), 7.51-7.41 (m, Ar-

14 C 10 C

H, 2H), 7.40-7.28 (m, Ar-H, 2H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1640, N-H_{bending} 1534, 1200, 1158, 1076, 987, 824, 764, 679, 643, 512.

3A.7 Synthesis of (E)-1-(4-bromophenyl)-3-phenylprop-2-en-1-one(5a)

4-Bromoacetophenone (199.00 mg, 1 mmol) and benzaldehyde (0.101 mL, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow solution. The reaction mixture was stirred for 5 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield = 92%, m.p. = 111-113°C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 7.63 (d, J=8.5Hz, Ar-H, 1H), 7.72 (d, J=11.0Hz, =CH-Ar, 1H), 7.35-7.31(m, Ar-H, 1H), 7.28 (d, J=11.5Hz, -CO-CH=, 1H), 7.18-7.16 (m, Ar-H, 2H), 7.12-7.09 (m, Ar-H, 1H), 7.04-7.01 (m, Ar-H, 1H), 6.93 (d, J= 11.0Hz, CO-CH=, ,1H), 6.89-6.88 (m, Ar-H, 2H), IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1647, N-H_{bending} 1533, 1206, 1157, 1076, 987, 824, 764, 679, 511, 502.

3A.8 Synthesis of (E)-1,3-bis(4-bromophenyl)prop-2-en-1-one(6a)

4-Bromoacetophenone (199.00 mg, 1 mmol) and 4-bromobenzaldehyde (0.196 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow solution. The reaction mixture was stirred for 6 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product.(Yellow solid). Yield = 94%, m.p. = 125-127 °C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.07-8.06 (m, Ar-H, 2H), 7.75 (d, *J*=15.5Hz, =CH-Ar, 1H), 7.59-7.57 (m, Ar-H, 2H), 7.53-7.49 (m, Ar-H, 3H), 7.22-7.20 (m, Ar-H, 2H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1643.59, N-H_{bending} 1533.98, 1209.26, 1157.47 1085.32, 997.17, 820.12, 760.04, 679.28.

3A.9 Synthesis of (E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one(7a)

4-Bromoacetophenone (199.00 mg, 1 mmol) and 4-methylbenzaldehyde (0.118 mL, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow solution. The reaction mixture was stirred for 4 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product (yellow solid). Yield = 92%, m.p. = 118-120 °C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 7.89-7.87 (m, Ar-H, 3H), 7.79 (d, *J*=8.0Hz, =CH-Ar, 1H), 7.64-7.63 (m, Ar-H, 3H), 7.45 (d, *J*= 15.5Hz, CO-CH=, 1H), 7.24-7.22 (m, Ar-H, 2H), 2.39(s, 3H). IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1657, N-H_{bending} 1536, 1205, 1156, 1076, 987, 814, 765, 679, 511, 506.

3A.10 Synthesis of (E)-1-(4-bromophenyl)-3-(3, 4-dimethoxy- phenyl)prop-2en-1-one(8a)

4-Bromoacetophenone (199.00 mg, 1 mmol) and 3,4-methoxybenzaldehyde (166.15 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow cloudy solution. The reaction mixture was stirred for 2 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product (white Solid). Yield= 94%. m.p. = 97-100^oC ⁻¹HNMR (CDCl₃, 500MHz) δ ppm: 7.89-7.87 (m, Ar-H, 2H), 7.81-7.75 (m, Ar-H, 1H), 7.65-7.59 (m, Ar-H, 2H), 7.34 (d, *J*= 15.5Hz, =CH-Ar, 1H), 7.27-7.22 (m, Ar-H, 2H), 6.92 (d, *J*= 8.5Hz, CO-CH=, 1H), 3.92 (s, 6H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1650, N-H_{bending} 1535, 1206, 1156, 1076, 987, 824, 764, 679, 514, 502.

17

1. Wang to sares

3A.11 Synthesis of (E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl) prop-2en-1-one(9a)

4-Bromoacetophenone (199.00 mg, 1 mmol) and 3,4,5-methoxybenzaidehyde (196.20 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white cloudy solution. The reaction mixture was stirred for 3 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. (White crystalline solid) Yield= 93%. m.p. = 119-121 ^oC. ¹HNMR (CDCl₃, 500MHz) δ ppm: 7.89-7.86 (m, Ar-H, 2H), 7.74 (d, *J*=15.5Hz, =CH-Ar, 1H), 7.73-7.63 (m, Ar-H, 2H), 7.36 (d, *J*=12.5Hz, OC-CH=, 1H), 6.86-6.85 (m, Ar-H, 2H), 3.99 (s, 9H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1650, N-H_{bending} 1535, 1206, 1156, 1076, 987, 824, 764, 679, 514, 502.

3A.12 Synthesis of (E)-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one (10a)

Acetophenone (0.116 mL, 1 mmol) and 3,4,5-trimethoxybenzaldehyde (196.20 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white cloudy solution. The reaction mixture was stirred for 3 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield= 95%. m.p. = 90-95°C. ¹HNMR (CDCl₃, 500MHZ) δ ppm: 8.04 (dd, *J*=3Hz, 1Hz, Ar-H, 2H), 7.82 (d, *J*=15.5, =CH-Ar, 1H), 7.59-7.55 (m, Ar-H, 2H), 7.53 (d, *J*=15Hz, -CO-CH= 1H), 7.51-7.48 (m, Ar-H, 3H), 7.22 (d, *J*=8Hz, Ar-H, 2H), 3.83 (s, 9H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1647, N-H_{bending} 1537, 1200, 1156, 1076, 987, 824, 764, 679, 514, 502.

3A.13 Synthesis of (E)-3-(4-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1one(11a)

2-hydroxyacetophenone (0.120 mL, 1 mmol) and 4-Chlorobenzaldehyde (140.5 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white cloudy solution. The reaction mixture was stirred for 3 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. (White crystalline solid) Yield= 96%. m.p. = 98-100 $^{\circ}$ C ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.01-7.99 (m, Ar-H, 2H), 7.73 (d, *J*=16Hz, =CH-Ar, 1H), 7.59-7.55 (m, Ar-H, 1H), 7.55 (d, *J*=11.0Hz, -CO-CH=, 1H), 7.52-7.45 (m, Ar-H, 5H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1656, N-H_{bending} 1535, 1200, 1146, 1075, 987, 827, 765 , 679, 514, 508.

3A.14 Synthesis of (E)-1-(4-hydroxyphenyl)-3-phenylprop-2-en-1-one(12a)

4-hydroxyacetophenone (0.120 mL, 1 mmol) and benzaldehyde (0.101 mL, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white cloudy solution. The reaction mixture was stirred for 5 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. (white solid) Yield 94%, m.p. = 131-132 °C. ¹HNMR (CDCl₃, 500MHz δ ppm: 8.46 (dd, *J*=2Hz, 1.5Hz, Ar-H, 1H), 8.22 (dd, *J*=1.0Hz, 1.0Hz, Ar-H, 1H), 8.03 (d, *J*=7Hz, Ar-H, 2H), 7.92 (d, *J*= 8Hz, 1H), 7.80 (d, *J*= 16Hz, =CH-Ar, 1H), 7.65 (d, *J*=15.5Hz, -CO-CH=, 1H), 7.60-7.57 (m, Ar-H, 2H), 7.51 (dd, *J*=7.5Hz, 1.5Hz, Ar-H, 2H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1640.59, N-H_{bending} 1533.98, 1209.26, 1157.47 1085.32, 997.17, 820.12, 760.04, 679.28.

4.52 1. 200

3A.15 Synthesis of (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one(13a)

Acetophenone (0.116 mL, 1 mmol) and 4-hydroxybenzaldehyde (122.0 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white solution. The reaction mixture was stirred for 5 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. (white solid) Yield 94%, m.p. = 128-130 °C. ¹HNMR (CDCl₃, 500MHz δ ppm: 8.03-8.01 (m, Ar-H, 2H), 7.78 (d, *J*=15.0Hz, O=C-CH=, 1H), 7.60-7.58 (m, Ar-H, 1H), 7.53-7.47 (m, 3H), 7.31-7.28 (m, ArH, 1H), 7.23 (d, *J*=15.5Hz, Ar-H, 1H), 7.51-7.14 (m, Ar-H, 1H), 6.92 (m, Ar-H, 1H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1640.59, N-H_{bending} 1533.98, 1209.26, 1157.47 1085.32, 997.17, 820.12, 760.04, 679.28.

3A.16 Synthesis of (E)-1-(2-hydroxyphenyl)-3-(4-nitrophenyl)prop-2-en-1-one (14a)

2-hydroxyacetophenone (0.12 mL, 1 mmol) and 4-nitrobenzaldehyde (151.0 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a white solution. The reaction mixture was stirred for 5 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. (white solid) Yield 94%, m.p. = 120-122 °C. ¹HNMR (CDCl₃, 500MHz δ ppm: 8.00-7.98 (m, Ar-H, 2H), 7.86 (d, *J*=15.5Hz, O=C-CH=, 1H), 7.68-7.66 (m, Ar-H, 2H), 7.51-7.48 (m, 2H), 7.38-7.35 (m, ArH, 1H), 7.23 (d, *J*=15.5Hz, =C-H, 1H), 7.51-7.14 (m, Ar-H, 1H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1640.59, N-H_{bending} 1533.98, 1209.26, 1157.47 1085.32, 997.17, 820.12, 760.04, 679.28.

3A.17 Synthesis of (E)-1-(4-chlorophenyl)-3-(3-nitrophenyl)prop-2-en-1-one (15a)

4-chloroacetophenone (0.1295 mL, 1 mmol) and 3-nitrobenzaldehyde (151.5 mg, 1 mmol) were dissolving in methanol (10 mL) with stirring at room temperature. Aqueous NaOH (0.080 g, 2 mmol) was added in portion, which gave a yellow solution. The reaction mixture was stirred for 4 hours at the same temperature. The solution was directly poured into crushed ice and 1N HCl (10 mL) was added to neutralize the solution. The resulting solution was precipitated on neutralization, filtered, and washed with water (2 × 20 mL). The precipitate was crystallized from solvent (EtOH or MeOH/CH₂Cl₂) to give the product or silica gel column chromatography is used in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent to purify the product. Yield= 91%, m.p. = 78-81°C. ¹HNMR (CDCl₃, 500MHz) δ ppm: 8.49-8.43 (m, Ar-H, 1H), 8.25 (d, *J*=12.5Hz,=CH-Ar, 1H), 8.05-8.03 (m, Ar-H, 2H), 7.91 (d, *J*=15.5Hz, O=C-CH=,1H), 7.87 (d,*J* = 13.0Hz, Ar-H, 1H), 7.67-7.63 (m, Ar-H, 2H), 7.56 (m, Ar-H, 2H); IR v_{max} (KBr, cm⁻¹): CO v_{stretch} 1650, N-H_{bending} 1535, 1206, 1156, 1076, 987, 824, 764, 679, 514, 502.

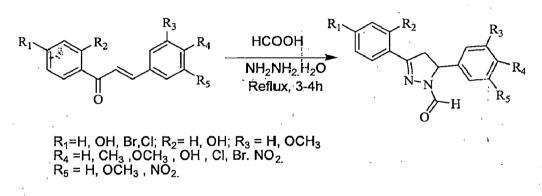
21

Sec. Sec. Capit

2B. SYNTHESIS OF N-FORMYLPYRAZOLINE

2B.1 General procedure for the synthesis of N-formylpyrazoline

Chalcones (1.00 mmol), hydrazine hydrate (5.00 mmol) and formic acid (5 mL) was heated at reflux for 3-4h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to obtain the N-formyl-pyrazolines in 81-96% yields.



Scheme 3

2B.2 Synthesis of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde(1b)

Chalcones (104 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 96% Yield; White powder m.p: 140-142⁰C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.97 (s, 1H), 7.75-7.73 (m, 2H), 7.45-7.42 (m, 3H), 7.36-7.28 (m, 5H), 5.54(dd, *J*= 4.5, 12.0Hz, 1H), 3.82 (dd, *J*= 12.0, 16.0Hz, 1H), 3.22 (dd, *J*= 4.5, 16.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ (ppm): 159.92, 155.59, 140.40, 130.70, 130.49, 128.86(2C), 128.65(2C), 127.82, 126.51(2C), 125.47(2C), 58.84, 42.45; IR v_{max} (KBr, cm⁻¹):1656, 1605, 1424, 1383, 1329, 1262, 1142, 904, 764, 696, 536; GCMS (m/z) = 250 [M⁺, C₁₆H₁₄N₂O],222, 221, 145(100%), 119, 114, 77, 65, 51.

2B.3 Synthesis of 5-(4-chlorophenyl)-3-phenyl-4,5-dihydropyrazole-1-carb - aldehyde(2b)

(E)-3-(4-chlorophenyl)-1-phenylprop-2-en-1-one (121.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3.30h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 95% yields; White solid m.p: 102-104 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.98 (s, 1H), 7.74 (d, *J* = 6.0Hz, 2H), 7.46-7.43 (m, 3H), 7.32-7.30 (m, 2H), 7.21 (d, *J* = 7.0Hz, 2H), 5.50 (dd, *J* = 4.5, 11.5Hz, 1H), 3.82 (dd, *J* = 11.5, 17.5Hz, 1H), 3.21 (dd, *J* = 4.5, 16.9Hz, 1H); ¹³C NMR (CDCl₃, 125MHz) δ ppm:160.22, 152.59, 135.46, 131.74; 130.00, 128.82 (2C),128.47 (2C), 127.37, 125.68 (2C), 125.60 (2C), 59.28, 40.09; IR v_{max} (KBr, cm⁻¹):1664, 1604, 1494, 1421, 1326, 1241, 1089, 1026, 823, 755; GCMS (m/z) = 284 [M⁺, C₁₆H₁₃ClN₂O], 286, 285, 255, 249, 207, 178, 97, 65, 51.

2B.4 Synthesis of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-carbaldehy - de(3b)

(E)-1,3-bis(4-chlorophenyl)prop-2-en-1-one (138.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 95% yields; White solid; m.p: 145-146^oC; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.93 (s, 1H), 7.68-7.65(m, 2H), 7.42-7.40 (m, 2H), 7.33-7.30(m, 2H), 7.19-7.17 (m, 2H), 5.50(dd, *J* = 5.0, 12.0, 1H), 3.80 (dd, *J* = 11.5, 17.5Hz, 1H), 3.16 (dd, *J* = 5.5, 16.0 Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.19, 153.89, 132.94, 131.04, 131.02, 130.12, 128.17 (2C), 125.97 (2C), 124.36 (2C), 123.95 (2C), 59.00, 40.80; IR v_{max} (KBr, cm⁻¹): 1676, 1616, 1480, 1426, 1314, 1251, 1229, 1086, 1013, 980, 826, 765, 645; GCMS (m/z) = 318 [M⁺, C₁₉H₁₈Cl₂N₂O], 320, 318, 291,191,179,153,138(100%), 222, 209, 206, 194, 153, 145, 77, 69, 51.

1.

1. 1. 1. 50 J. 18 .

2B. 5 Synthesis of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1-carb - aldehyde(4b)

(E)-3-(4-bromophenyl)-1-phenylprop-2-en-1-one (142.92 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 94% yields; White solid; m.p: 104-106 0 C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.95 (s, 1H), 7.77-7.71 (m, 2H), 7.54-7.29 (m, 5H), 7.18-7.12 (m, 2H), 5.50 (dd, J = 5.0, 12.5Hz, 1H), 3.82 (dd, J = 11.9, 18.0Hz, 1H), 3.20 (dd, J = 5.0, 16Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.09, 155.66, 139.63, 132.21, 130.84, 130.70 (2C), 128.91 (2C), 127.54 (2C), 126.72 (2C), 124.98, 60.53, 42.45. IR v_{max} (KBr, cm⁻¹): 1669, 1598, 1423, 1322, 1134, 1071, 1017, 831, 756, 690, 536, 442. GCMS (m/z) = 328 [M⁺, C₁₆H₁₃BrN₂O], 330, 329, 327, 299, 250, 222, 153, 145, 145, 77, 51.

2B.6 Synthesis of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-carb - aldehyde(5b)

(E)-1-(4-bromophenyl)-3-phenylprop-2-en-1-one (142.92 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 93% yields; White solid; m.p: 145-147 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.95 (s, 1H), 7.59-7.50 (m, 4H), 7.45-7.35 (m, 1H), 7.29-7.23(m, 4H), 5.54 (dd, *J* = 5.0, 17.0Hz, 1H), 3.78 (dd, *J* = 12.0, 13.0Hz, 1H), 3.20 (dd, *J* = 5.0, 17.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ (ppm): 159.93, 155.71, 139.59, 132.15, 130.61, 130.42 (2C), 128.88 (2C), 127.45 (2C), 126. 27 (2C), 124.98, 60.48, 42.30; IR v_{max} (KBr, cm⁻¹): 1668, 1598, 1413, 1322, 1134,1071, 1017, 831, 756, 690, 536, 442; GCMS (m/z) = 328 [M⁺, C₁₆H₁₃BrN₂O], 330, 329, 327, 299, 250, 222, 153, 145, 145, 77, 51.

2B.7 Synthesis of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1-carbaldehyde (6b)

(E)-1,3-bis(4-bromophenyl)prop-2-en-1-one (142.92 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 92% yields; White solid; m.p: 138-140 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.93 (s, 1H), 7.86-7.84(m, 2H), 7.58-7.54 (m, 2H), 7.45-7.36 (m, 2H), 7.28-7.19 (m, 2H), 5.53 (dd, *J* = 5.0, 12.0Hz, 1H), 3.79 (dd, *J* = 12.5, 17.0Hz, 1H), 3.19 (dd, *J* = 5.0, 16.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.21, 151.89, 131.94, 131.54, 131.22, 130.16, 128.22 (2C), 127.67 (2C), 125.97 (2C), 124.36 (2C), 60.00, 40.80; IR v_{max} (KBr, cm⁻¹): 1649, 1611, 1511, 1430, 1362, 1324, 1225, 1129, 1076, 1017, 823, 735, 536; GCMS (m/z) = 405 [M⁺, C₁₆H₁₂Br₂N₂O],408, 407, 376, 328, 248, 221, 207, 154, 91, 77, 51.

2B.8 Synthesis of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-carb aldehyde(7b)

(E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one (150.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3.10h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 91% yields; White solid; m.p: 180-182 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.93 (s, 1H), 7.71-7.65 (m, 4H), 7.22-7.13 (m, 4H), 5.49 (dd, J = 5.0, 11.0Hz, 1H), 3.76 (dd, J = 12.0, 17.5Hz, 1H), 3.18 (dd, J = 5.0, 16.5Hz, 1H), 2.18 (s, 3H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.08, 154.67, 137.89, 137.54, 132.08, 130.75, 129.95 (2C), 128.12 (2C), 125.61(2C), 125.0 (2C), 59.06, 42.51, 21.14; IR v_{max} (KBr, cm⁻¹): 1650, 1596, 1507, 1423, 1317, 1246, 1122, 1059, 1005, 819, 748, 553, 408; GCMS (m/z) = 342-10.14; M⁺, C₁₇H₁₅BrN₂O], 344, 343, 342, 327, 313, 249, 243, 221, 145, 104, 77, 51, 15.

25

· مطاقيقة بأجاز ماتار بالمحشد

S ACC No Date

T. ROORY

2B.9 Synthesis of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde(8b)

(E)-1-(4-broinophenyl)-3-(3,4-dimethoxyphenyl)prop-2-en-1-one (173.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 93% yields; White solid; m.p: 118-120 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.95 (s, 1H), 7.70-7.65 (m, 4H), 6.68-6.62 (m, 3H), 5.49 (dd, J = 4.5, 11.9Hz, 1H), 3.82 (s, 6H), 3.74 (dd, J = 11.0,18.0Hz, 1H), 3.18 (dd, J = 5.0, 17.0Hz, 1H); ¹³C NMR (CDCl₃, 125MHz) δ ppm: 160.26, 154.97, 153.67, 137.55, 136.34, 130.80, 128.90 (2C), 126.73 (2C), 125.34, 124.01, 123.30, 123.10, 112.29, 60.90, 59.37, 40.90; IR v_{max} (KBr, cm⁻¹): 2942, 1659, 1590, 1514, 1421, 1311, 1248, 1165, 1069, 1020, 822, 755, 632, 586, 528; GCMS (m/z) = 388 [M⁺, C₁₈H₁₇BrN₂O₃], 390, 389, 388, 359, 357, 327, 309, 249, 221, 154, 119, 91, 77, 65, 51, 31.

2B. 10 Synthesis of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-dihydro -pyrazole-1-carbaldehyde(9b)

(E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one (188.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3.15h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 93% yields; White solid; m.p: 175-177 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.98 (s, 1H), 7.81-7.65 (m, 4H), 6.46 (s, 2H), 5.47 (dd, *J* = 5.0, 13.0Hz, 1H), 3.80 (s, 9H), 3.60 (dd, *J* = 12.0, 17.0Hz, 1H), 3.16 (dd, *J* = 5.0, 16.5Hz, 1H); ¹³C NMR (CDCl₃, 125MHz) δ ppm: 160.09, 155.79 153.52, 137.42, 136.28, 130.74 (2C), 128.88 (2C), 126.65 (2C), 125.04, 103.19 (2C), 60.81, 59.31, 56.21 (2C), 42.82; IR v_{max} (KBr, cm⁻¹): 2929, 1665, 1595, 1503, 1422, 1327, 1241, 1124, 1022, 835, 752, 648, 542; GCMS (m/z) = 418 [M⁺, C₁₉H₁₉BrN₂O₄], 420, 419, 418, 390, 389, 388, 359, 357, 327, 309, 249, 221, 154, 119, 91, 77, 65, 51, 31.

2B.11 Synthesis of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1carbaldehyde(10b)

(E)-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one (149.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 94% yields; White solid; m.p: 145-146 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 9.00 (s, 1H), 7.73 (dd, *J* = 1.5, 6.5Hz, 2H), 7.48-7.42 (m, 3H), 6.46 (s, 2H), 5.47 (dd, *J* = 5.0, 12.0Hz, 1H), 3.83 (s, 9H), 3.80 (dd, *J* = 5.0, 16.0Hz, 1H), 3.23 (dd, *J* = 5.0, 16.5Hz, 1H). ¹³C-NMR (CDCl₃, 125MHz) δ (ppm): 160.26, 155.97, 153.77, 137.55, 136.34, 130.80 (2C), 128.90 (2C), 126.73 (2C), 125.12, 102.29 (2C), 60.9, 59.27, 56.15 (2C), 42.90; IR v_{max} (KBr, cm⁻¹): 1667, 1593, 1508, 1459, 1419, 1325, 1243, 1226, 1011, 827, 759, 693,644. GCMS (m/z) = 340 [M⁺, C₁₉H₂₀N₂O₄], 340, 339, 311, 237, 222, 209, 206, 194, 19, 153, 145, 77(100%), 76,65, 61.

2B.12 Synthesis of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-dihydro pyrazole-1-carbaldehyde(11b)

(E)-3-(4-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1-one (129.0 mg, 0.50 mmol), hyd razine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 4h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 87% yields; Yellow solid; m.p: 185-187 °C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.90 (s, 1H), 7.44-7.32 (m, 4H), 7.22-7.20 (m, 3H), 6.92 (m, 1H), 5.48 (dd, J = 5.0, 12.0Hz, 1H), 3.90 (dd, J = 11.5, 17.0Hz, 1H), 3.33 (dd, J = 5.0, 17.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm:. 160.45, 153.05, 151.56, 140.85, 132.55, 132.34, 131.44, 128.95 (2C), 127.29 (2C), 127.19, 125.23, 112.26, 59.14, 41.96; IRv_{max} (KBr, cm⁻¹): 3423, 1670, 1610, 1490, 1428, 1324, 1251, 1219, 1086, 1013, 990, 825, 765, 640, 557; GCMS (m/z) = 300 [M⁺, C₁₆H₁₃ClN₂O₂], 302, 301, 300, 283, 265, 271, 249,221, 173, 97,77,51.

27

A Salar and

2B.13 Synthesis of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydro- pyrazole-1carbaldehyde(12b)

(E)-1-(4-hydroxyphenyl)-3-phenylprop-2-en-1-one (112.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 3.50h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 85% yields; White solid; m.p: 128-130 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 9.00 (s, 1H), 7.73-7.71 (m, 2H), 7.46-7.42 (m, 3H), 7.29-7.23 (m, 1H), 7.17-7.00 (m, 1H), 6.70-6.68(m, 2H), 5.49 (dd, *J* = 4.5, 12.0Hz, 1H), 3.79 (dd, *J* = 12.0, 17.0Hz, 1H), 3.23 (dd, *J* = 4.0, 16.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.45, 154.21, 151.12, 131.54, 131.56, 130.15, 128.12 (2C), 127.32 (2C), 125.40 (2C), 123.50 (2C), 59.56, 40.14; IR v_{max} (KBr, cm⁻¹): 3443, 1660, 1596, 1490, 1428, 1324, 1251, 1219, 1086, 1013, 990, 825, 765, 640, 557; GCMS(m/z) = 266 [M⁺, C₁₆H₁₄N₂O₂], 268, 267, 266, 249, 221, 173, 97, 51.

2B.14 Synthesis of 5-(3-hydroxyphenyl)-3-phenyl-4,5-dihydro- pyrazole-1carbaldehyde(13b)

(E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one (112.0 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 4h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 83% yields; White solid; m.p: 176-178 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 9.00 (s, 1H), 7.77-7.72 (m, 2H), 7.48-7.41 (m, 3H), 7.29-7.23 (m, 1H), 7.17(m, 1H), 6.70-6.68 (m, 2H), 5.49 (dd, *J* = 4.5, 11.5Hz, 1H), 3.79 (dd, *J* = 12.0, 17.2Hz,1H), 3.23 (dd, *J* = 4.5, 16.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ (ppm): 159.45, 152.65, 151.10, 132.34, 130.56, 130.15, 127.82 (2C), 127.32 (2C), 124.40 (2C), 122.50 (2C), 59.36, 40.84; IR v_{max} (KBr, cm⁻¹): 3443, 1660, 1596; 1490, 1428, 1324, 1251, 1219, 1086, 1013, 990, 825, 765, 640, 557; GCMS(m/z) = 266 [M⁺, C₁₆H₁₄N₂O₂]; 268, 267, 266, 249, 221, 173, 97, 51.

2B.15 Synthesis of 3-(2-hydroxyphenyl)-5-(4-nitrophenyl)-4,5-dihydro pyrazole-1-carb -aldehyde(14b)

(E)-1-(2-hydroxyphenyl)-3-(4-nitrophenyl)prop-2-en-1-one (134.5 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 4h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 85% yields; White solid; m.p: 178-180 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 8.95 (s, 1H), 7.76-7.67 (m, 2H),7.41-7.35 (m, 3H), 7.34-7.25 (m, 3H), 5.53 (dd, *J* = 5.0, 11.0Hz, 1H), 3.75 (dd, J = 12.0, 17.0Hz, 1H), 3.20 (dd, *J* = 4.5, 16.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.32, 153.23, 151.23, 140.79, 133.51, 132.29, 131.16, 128.58 (2C), 128.10 (2C), 127.15, 125.52, 113.46, 59.14, 41.96; IR v_{max} (KBr, cm⁻¹): 3440, 1667, 1588, 1490, 1428, 1324, 1251, 1219, 1086, 1013, 990, 835, 746, 640, 557; GCMS (m/z) = 311 [M⁺, C₁₆H₁₃N₃O₄]; 313, 312, 311, 265, 249, 221, 219, 144, 97,77, 51,45.

2B.16 Synthesis of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydro pyrazole-1-carbaldehyde(15b)

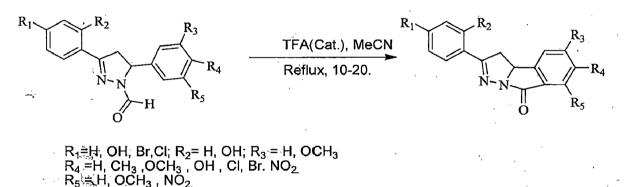
(E)-1-(4-chlorophenyl)-3-(3-nitrophenyl)prop-2-en-1-one (143.5 mg, 0.50 mmol), hydrazine hydrate (0.121 mL, 2.50 mmol) and formic acid (2.50 mL, 64.50 mmol) was heated at reflux for 4h. After completion of the reaction as indicated by TLC, the reaction mixture was poured into crushed ice. The precipitate was separated by filtration, washed with water under reduced pressure, and recrystallized from methanol to gave 81% yields; White solid; m.p: 139-141 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.96 (s, 1H), 8.21-8.18 (m, 2H), 7.67-7.58 (m, 3H), 7.35-7.22 (m, 3H), 5.65 (dd, *J* = 5.0, 11.5Hz,1H), 3.89 (dd, *J* = 12.0, 16.0Hz, 1H), 3.21 (dd, *J* = 5.0, 16Hz,1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 160.41, 140.10, 139.35, 132.34, 131.46, 130.14, 128.92 (2C), 127.32 (2C), 125.17, 124.19, 123.82, 116.40, 59.36, 41.14; IR v_{max}(KBr, cm⁻¹): 1636, 1594, 1528, 1420, 1351, 1092, 823, 760, 686, 477; GCMS(m/z) = 329 [M⁺, C₁₆H₁₂ClN₃O₃], 331, 330, 329, 294, 249, 221, 144, 97, 77, 51, 29.

1 to det

2C SYNTHESIS OF PYRAZOLOISOINDOLE

2C.1 General procedure for the synthesis of pyrazoloisoindole

A well stirred and refluxed solution of N-formyl-pyrazoline (1.0 mmol) in acetonitrile (10 mL) was added 6-8 drops of TFA and the reaction was continued for 10-20 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent.



Scheme 4

2C.2 Synthesis of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(1c)

A well stirred and refluxed solution of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde (100 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 10 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 94% yields; White crystaline solid; m.p: $108-110^{0}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 7.72-7.70 (m, 2H), 7.41-7.36 (m, 3H), 7.29-

7.22 (m, 2H), 7.18-7.16 (m, 2H), 5.54 (dd, J = 4.5, 11.5Hz, 1H), 3.73 (dd, J = 11.5, 17.5Hz, 1H), 3.19 (dd, J = 4.5, 17.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.87, 138.47, 131.02 130.32, 129.03, 128.15 (2C), 127.85 (2C), 127.37 , 126.19 (2C), 124.55 (2C), 60.33, 40.76; IR v_{max} (KBr, cm⁻¹): 1692, 1471, 1401, 1233, 1183, 1028, 943, 823, 750. GCMS(m/z) = 248 [M⁺, C₁₆H₁₂N₂O], 241, 221, 215, 206, 205, 204, 178, 165, 146, 132, 115, 104, 103, 102; HR-MS (m/z) for C₁₆H₁₂N₂O calcd. 248.0950; found: 248.0946.

2C.3 Synthesis of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a] isoindol-8-one(2c)

A well stirred and refluxed solution of 5-(4-chlorophenyl)-3-phenyl-4,5-dihydropyrazole-1carbaldehyde (113.6 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 12 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 90% yields; White solid; m.p: 145-147 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 7.71 (d, *J* = 8.0Hz, 2H), 7.42-7.38 (m, 3H), 7.24 (d, *J* = 6.5Hz, 1H), 7.12 (d, *J* = 6.5Hz, 2H), 5.50 (dd, *J* = 4.5, 11.5Hz, 1H), 3.73 (dd, *J* = 11.5, 17.5, 1H), 3.16 (dd, *J* = 4.5, 17.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm:156.21, 136.47, 132.74, 129.90, 128.47, 127.82 (2C), 127.37 (2C), 125.68 (2C), 125.60 (2C), 123.54, 59.28, 40.09; IR v_{max} (KBr, cm⁻¹): 2924, 1703, 1560, 1469, 1403, 1275, 1221, 1196, 1170, 1144, 1084, 1009, 947, 843, 760, 69⁵8; GCMS(m/z) = 282 [M⁺, C₁₆H₁₁ClN₂O], 284, 283, 282, 271, 206, 191,104, 102; HR-MS (m/z) for C₁₆H₁₁ClN₂O calcd. 282.0560; found: 282.0546.

2C.4 Synthesis of 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo [5,1-a]isoindol-8-one(3c)

A well stirred and refluxed solution of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1carbaldehyde (127.2 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 13 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane

31

, Pit

the start

and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 93% yields; White solid; m.p. 146-148^oC; ¹HNMR (CDCl₃, 500Mz) δ (ppm):7.72-7.69 (m, 2H), 7.43-7.41 (m, 2H), 7.33-7.30 (m, 2H), 7.18-7.16 (m, 1H), 5.52 (dd, J = 5.0, 11.5Hz, 1H), 3.78 (dd, J = 11.5, 17.5Hz, 1H), 3.20 (dd, J = 5.0, 17.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.45, 135.42, 132.74, 129.00, 127.82 (2C), 127.47 (2C), 125.68, 125.60, 124.63, 123.93, 122.14, 118.02, 60.18, 41.09; IR v_{max} (KBr, cm⁻¹): 1706, 1567, 1464, 1410, 1270, 1221, 1196, 1160, 1154, 1084, 1014, 943, 863, 763, 686; GCMS (m/z) = 316[M⁺, C₁₆H₁₀Cl₂N₂O]; 318, 317, 316, 281, 247, 171, 104, 102; HR-MS (m/z) for C₁₆H₁₀Cl₂N₂O calcd. 316.0170; found: 316.0181.

2C.5 Synthesis of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8one(4c)

A well stirred and refluxed solution of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1carbaldehyde (131.2 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 14 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 89% yields; Light pink solid; m.p: 162-164 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm):7.76-7.69 (m, 2H), 7.55-7.45 (m, 2H), 7.45-7.28 (m, 2H), 7.10-7.04 (m, 2H), 5.49 (dd, *J* = 5.0, 11.5Hz, 1H), 3.72 (dd, *J* = 5.0, 17.5Hz, 1H), 3.15 (dd, *J* = 4.5, 16.5Hz; 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.32, 137.53, 130.60, ⁻¹130.33, 128.31, 127.64 (2C), 127.04 (2C), 125.17, 124.28, 123.95, 122.54, 116.20, 59.89, 40.09, IR v_{max} (KBr, cm⁻¹): 1691, 1463, 1401, 1282, 1199,1159, 1070, 1010, 943, 826, 732, 657, 543. GCMS (m/z) =326 [M⁺, C₁₆H₁₁BrN₂O]; 328, 327, 326, 247, 249, 171, 102; HR-MS (m/z) for C₁₆H₁₁BrN₂O calcd. 326.0055; found: 326.0145.

2C.6 Synthesis of 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8one(5c)

A well stirred and refluxed solution of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1carbaldehyde (131.2 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 12 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 90% yields; White solid; m.p: 146-148 ⁰C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.76-7.69 (m, 2H), 7.56-7.49 (m, 3H), 7.43-7.40 (m, 3H), 5.65 (dd, *J* = 4.5, 11.5Hz, 1H), 3.76 (dd, *J* = 12.0, 17.5Hz, 1H), 3.23 (dd *J* = 4.5, 17.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.89, 137.28, 130.10, 130.01, 128.47, 127.45 (2C), 127.34 (2C), 125.32, 124.16, 123.41, 122.51, 117.18, 60.52, 41.05; IR v_{max} (KBr, cm⁻¹):1697, 1586, 1468, 1391, 1215, 1178, 1071, 1013, 943, 832, 752, 698, 658, 538. GCMS (m/z) =326 [M⁺ C₁₆H₁₁BrN₂O]; 328, 327, 326, 247, 249, 171, 102; HR-MS (m/z) for C₁₆H₁₁BrN₂O calcd. 326.0055; found: 326.0154.

2C.7 Synthesis of 6-bromo-2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a] isoindol-8-one(6c)

A well stirred and refluxed solution of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1carbaldehyde (161.97 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 11 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃. After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 89% yields; White solid; m.p: 138-140 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.79 (dd, J = 5.5, 8.0Hz, 2H), 7.48 (d, J = 8.0Hz, 2H), 7.15-7.11 (m, 3H), 5.55(dd, J = 4.5, 11.5Hz, 1H), 3.80 (dd, J = 11.5, 17.0Hz, 1H), 3.20 (dd J = 4.5, 16Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.20, 136.98, 133.25, 129.15, 128.51 (2C), 126.92 (2C), 125.59, 125.40, 124.57, 124.00, 122.03, 119.10,

33

59.23, 40.10; IR v_{max} (KBr, cm⁻¹): 1702, 1598, 1452, 1403, 1341, 1219, 1161, 1017, 951, 839, 735, 594, 545; GCMS (m/z) = 404 [M⁺, C₁₆H₁₀Br₂N₂O]; 407, 405, 403, 324, 247, 171, 104, 102. HR-MS(m/z) for C₁₆H₁₁BrN₂O calcd. 403.9160; found: 403.9154.

2C.8 Synthesis of 2-(4-bromophenyl)-6-methyl-3,3a-dihydro pyrazolo[5,1a] isoindol-8-one(7c)

A well stirred and refluxed solution of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-carbaldehyde (136.81 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 14 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 89% yields; Yellow solid; m.p: 165-167 0 C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.72-7.65 (m, 2H), 7.63-7.58 (m, 2H), 7.19-7410 (m, 3H), 5.58 (dd, *J* = 4.5, 11.0Hz, 1H), 3.72 (dd, J = 11.5, 16.5Hz, 1H), 3.18 (dd =4.5, 16.0Hz, 1H), 2.34 (s, 3H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.30, 136.21, 130.60, 130.33, 128.22, 127.59 (2C), 127.04 (2C), 126.23, 125.35, 124.28, 124.20, 123.77, 59.89, 40.09, 28.12; IR v_{max} (KBr, cm⁻¹): 1698, 1596, 1507, 1413, 1318, 1256, 1112, 1049, 1005, 819, 748,-543, 40; GCMS (m/z) =340 [M⁺, C₁₇H₁₃BrN₂O]; 342, 341, 340, 324, 247, 171, 102; HR-MS (m/z) for C₁₇H₁₃BrN₂O calcd. 340.2110; found: 340.2117.

2C. 9 Synthesis of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydro pyrazolo - [5,1-a]isoindol-8-one(8c)

A well stirred and refluxed solution of 3-(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde (155.21 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 13 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 88% yields; Yellow solid; m.p: 142-144 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 7.57 (d, *J* = 8.5Hz, 2H), 7.51 (d, J = 8.5Hz, 2H), 6.75 (m, 2H), 5.49 (dd, J = 4.5, 11.0Hz, 1H), 3.76 (m, 6H), 3.67 (dd, J = 4.5, 16.5Hz, 1H), 3.16 (dd, J = 4.5, 16Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm:156.32, 147.87, 147.53, 130.60 (2C), 130.33, 127.64, 127.06 (2C), 124.28, 116.20, 115.77, 113.59, 110.02, 59.69, 54.36 (2C), 40.09; IR v_{max} (KBr, cm⁻¹): 1695, 1598, 1519, 1461, 1409, 1253, 1221,1193, 1157, 1024, 810, 727. GCMS (m/z) = 386 [M⁺, C₁₈H₁₅BrN₂O₃]; 386, 387, 386, 307, 277, 247, 171, 102; HR-MS (m/z) for C₁₈H₁₅BrN₂O₃ calcd. 386.0266; found: 386.0255.

2C.10 Synthesis of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydropyrazolo[5,1-a] isoindol-8-one(9c)

A well stirred and refluxed solution of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde (167.22 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 14 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃. After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 87% yields; Yellowish solid; m.p: 188-190 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.65 (dd, *J* = 1.5, 6.5Hz, 2H), 7.59-7.57 (m, 2H), 6.40 (s, 1H), 5.56 (dd, *J* = 4.5, 11.5Hz, 1H), 3.80 (s, 9H), 3.79 (dd, *J* = 11.5, 18.0Hz, 1H), 3.23 (dd, *J* = 4.5, 18.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 158.00, 147.79, 137.94, 135.04, 132.22, 129.12, 128.67 (2C), 127.52 (2C), 125.97, 117.36, 116.17, 115.23, 60.79, 57.80, 56.14 (2C), 41.80, ³TR v_{max} (KBr, cm⁻¹): 1695, 1592, 1503, 1464, 1418, 1337, 1224, 1164,1126, 1067, 1001, 837, 701,636, 536; GCMS (m/z) = 416 [M⁺, C₁₉H₁₇BrN₂O₄]; 418, 417, 416, 385, 355, 337, 324, 247, 171; HR-MS (m/z) for C₁₉H₁₇BrN₂O₄ ealed. 416.0372; found: 416.0364.

2C.11 Synthesis of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydro pyrazolo[5,1-a] isoindol-8-one(10c)

A well stirred and refluxed solution of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde (136.14 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 13 h. After TLC monitoring, the reaction

mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃. After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 90% yields; White solid; m.p: 175-178 0 C; ¹H NMR (CDCl₃, 500Mz) δ (ppm):7.80 (d, *J* = 6.5Hz, 2H), 7.47 (dd, *J* = 7.0, 12.5Hz, 3H), 6.42 (d, *J* = 3.0, 1H), 5.57 (dd, *J* = 4.5, 11.5Hz, 1H), 3.82-3.79 (3s, 9H), 3.49 (dd, *J* = 11.5, 16.5Hz, 1H), 3.27 (dd, *J* = 4.5, 16.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm:158.45, 146.85, 132.55, 131.54, 131.44, 128.95 (2C), 127.29 (2C), 126.52, 125.42, 118.64, 117.48, 116.71, 61.64, 60.59, 56.45 (2C), 41.96; IR v_{max} (KBr, cm⁻¹): 1702, 1594, 1510, 1462, 1421, 1342, 1163, 1130, 1002, 893, 831, 771, 631, 537 631, 537. GCMS (m/z) = 338 [M⁺; C₁₉H₁₈N₂O₄],340, 339, 311, 236(100%), 222, 209, 206, 194, 153, 145, 77, 69, 51; HR-MS (m/z) for C₁₉H₁₈N₂O₄ calcd. 338.1267; found 338.1271.

2C.12 Synthesis of 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a] isoindol-8-one(11c)

A well stirred and refluxed solution of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5dihydropyrazole-1-carbaldehyde (120.02 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 16 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 78% yields; Yellow solid; m.p: 157-159 0 C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.34-7.33 (m, 2H), 7.24-7.18 (m, 3H), 7.10-6.97 (m, 1H), 6.97-6.94 (m, 1H), 5.57 (dd, *J* = 4.5, 11.5Hz, 1H), 3.94 (dd, J = 12.0, 17.5Hz, 1H), 3.38 (dd *J* = 4.0, 17.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.85, 141.12, 132.34, 130.46, 130.25, 127.92 (2C), 126.89 (2C), 125.42, 124.37, 123.19, 120.54, 119.48, 60.36, 40.80; IR v_{max} (KBr, cm⁻¹): 3431, 1694, 1606, 1470, 1407, 1304, 1225, 1073, 1013, 910, 814, 764, 503; GCMS(m/z) = 298 [M⁺, C₁₆H₁₁ClN₂O₂]; 300, 299, 298, 281, 263, 247, 171, 77; HR-MS (m/z) for C₁₆H₁₁ClN₂O₂ calcd 298.0509; found: 298,0510.

2C.13 Synthesis of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(12c)

A well stirred and refluxed solution of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydropyrazole-1carbaldehyde (106.44 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 17 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 76% yields; White solid; m.p: 126-128 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm):7.60-7.59 (m, 2H), 7.29-7.27 (m, 2H), 7.22 (d, J = 7.5Hz, 1H), 7.18-7.16 (m, 2H), 6.80-6.78 (m, 1H), 5.52 (br, D₂-Exchangeble, 1H), 5.52 (dd, *J* = 4.5, 11.5Hz, 1H), 3.26 (dd, *J* = 11.5, 17.5Hz, 1H), 3.15(dd, *J* = 4.5, 17.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm:156.38, 141.13, 131.39, 130.49, 130.22, 128.88 (2C), 127.27 (2C), 125.31, 123.09, 117.32, 115.49, 112.90, 60.63, 40.12; IR v_{max} (KBr, cm⁻¹): 3349, 1669, 1600, 1490, 1451, 1358, 1271, 1222, 1169, 1038, 939, 841, 773, 615, 541, 470. GCMS (m/z) = 264 [M⁺, C₁₆H₁₂N₂O₂]; 266, 265, 264, 247, 171, 102, 77, 55; HR-MS (m/z) for C₁₆H₁₂N₂O₂ calcd. 264.0899; found: 264.0889.

2C.14 Synthesis of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a] isoindol-8-one(14c)

A well stirred and refluxed solution of 5-(4-hydroxyphenyl)-3-phenyl-4,5-dihydropyrazole-1carbaldehyde (106.44 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 18 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 75% yields; White solid; m.p: 148-150 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 7.78-7.72 (m, 2H), 7.50-7.42 (m, 3H), 6.77-6.67 (m, 3H), 5.55 (br, D₂-Exchangeble, 1H), 5.53 (dd, *J* = 4.5, 12.0Hz, 1H), 3.78 (dd, *J* = 11.5, 17.5Hz, 1H), 3.26 (dd, *J* = 4.5, 17.0Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.40, 141.10, 131.44, 130.56, 130.15, 128.92 (2C), 127.32 (2C), 125.34, 123.14, 117.40, 115.54, 112.70, 61.36, 41.84; IR v_{max} (KBr, cm⁻¹): 3457, 1679, 1604, 1588, 1402, 1353, 1217, 1158, 760, 695. GCMS (m/z)= 264 [M⁺, C₁₆H₁₂N₂O₂]; 266, 265, 264, 247, 171, 102, 77, 55; HR-MS (m/z) for C₁₆H₁₂N₂O₂ calcd. 264.0899; found: 264.0898.

2C.15 Synthesis of 2-(2-hydroxyphenyl)-6-nitro-3,3a-dihydro- pyrazolo[5,1a]isoindol-8-one(14c)

A well stirred and refluxed solution of 3-(2-hydroxyphenyl)-5-(4-nitrophenyl)-4,5dihydropyrazole-1-carbaldehyde (124.43 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 18 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 74% yields; White solid; m.p: 156-158 $^{\circ}$ C; ¹H NMR (CDCl₃, 500Mz) δ (ppm): 7.64 (dd, *J* = 1.5, 7.0Hz, 2H), 7.62-7.33 (m, 2H), 7.29-7.27 (m, 1H), 7.22 (m, 2H), 5.55 (dd, *J* = 4.5, 11.5Hz, 1H), 3.70 (dd, *J* = 11.5, 17.5Hz, 1H), 3.15 (dd *J* = 4.5, 17.5Hz, 1H); ¹³CNMR (CDCl₃, 125MHz) δ ppm:156.45, 143.85, 137.55, 132.34, 131.44, 128.95, 127.29 (2C), 126.82, 125.23, 123.85, 122.17, 121.35, 119.06, 60.82, 41.96; IR v_{max} (KBr, cm⁻¹): 3443, 1696, 1584, 1480, 1418, 1334, 1246, 1215, 1086, 1014, 986, 845, 746, 641, 558; GCMS(m/z) = 309 [M⁺, C₁₆H₁₁N₃O₂]; 311, 310, 309, 292, 247, 171, 102, 77, 55; HR-MS (m/z) for C₁₆H₁₁N₃O₂ calcd. 309.0750; found: 309.0744.

2C.16 Synthesis of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo [5,1-a] isoindol-8-one(15c)

A well stirred and refluxed solution of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5dihydropyrazole-1-carbaldehyde (131.62 mg, 0.40 mmol) in acetonitrile (4 mL) was added 3-5 drops of TFA and the reaction was continued for 20 h. After TLC monitoring, the reaction mixture was cooled, evaporated the solvent under reduced pressure. The residue was dissolved in dichloromethane and washed with 10% aq.Na₂CO₃ After extraction, the organic layer was dried on anhyd. Na₂SO₄ and evaporated under reduced pressure. The product was purified by silica gel column chromatography in a mixture of hexane and ethyl acetate (9:1 v/v) as eluent we got 75% yields; White solid; m.p: 180-182 $^{\circ}$ C; ¹HNMR (CDCl₃, 500Mz) δ (ppm): 8.21-8.19 (m, 1H), 7.73 (d, J = 8.5Hz, 2H), 7.58-7.55 (m, 2H), 7.46 (d, J = 9.0Hz, 2H), 5.58 (dd, J = 5.0, 12.0Hz, 1H), 3.92 (dd, J = 11.5, 17.5, 1H), 3.23 (dd, J = 4.5, 17.5Hz, 1H); ¹³C-NMR (CDCl₃, 125MHz) δ ppm: 156.40, 141.10, 142.35, 132.44, 130.56, 130.15, 128.92, 127.32 (2C), 126.52 (2C), 125.11, 124.30, 117.40, 60.36, 40.14; IR v_{max} (KBr,cm⁻¹): 1701, 1594, 1538, 1425, 1355, 1087, 833, 766, 714, 676, 524 467, GCMS(m/z) =327 [M⁺, C₁₆H₁₀ClN₃O₃]; 329, 328, 327, 292, 291, 171, 77, 55; HR-MS (m/z) for C₁₆H₁₀ClN₃O₃ calcd. 327.0411; found: 327.0412.

Reference

- Pramod Singh, Jagmohan S. Negi, Geeta Joshi nee Pant and Mohan S.M. Rawat Molbank.
 2010, 1, M650.
- 2. S. Wattanasin and W. S. Murphy, Synthesis, 1980, 8, 647.
- 3. J. Safaei-Ghomi, A. H. Bamoniri, and M. Soltanian-Telkabadi; Chemistry of Heterocyclic Compounds, 2006, 42, 7.
- 4. R.Nirar, El-Rayyes, 'H. George and S Hayat; J. Chem. Eng. Data 1984, 29, 225-229.
- (a) N. Ahmed; J.E. van Lier; *TetrahedronLett.* 2007, 48, 5407–5409; (b) N. Ahmed; J.E. van Lier; *TetrahedronLett.* 2007, 48, 13–15; (c) N. Ahmed; J.E. van Lier; *Tetrahedron Lett.* 2006, 47, 5345–5349; (d) N. Ahmed; J.E. van Lier; *TetrahedronLett.* 2006, 47, 2725–2729; (e) N. Ahmed; H. Ali; J.E. van Lier; *TetrahedronLett.* 2005, 46, 253–256; (f) N. Ahmed; H. Ali; J.E. van Lier; *TetrahedronLett.* 2006, 10, 1172–1178; (g) N. Ahmed; W.H.J. Ansari; *J.Chem.Research*(S) 2003, 572–573.

Chapter - 3 **Results and Discussion** 41

CHAPTER 3

RESULTS AND DISCUSSION

2.1 Synthesis and Characterization of synthesis and characterization of 2-(substituted aryl)-3,3a-dihydro-8h-pyrazolo [5,1-a]isoindol-8-ones derivatives

Using Claisen-Schmidt reaction condition, acetophenone and benzaldehyde gave α , β unsaturated ketone (chalcone) using 10-60 wt% of alkaline hydroxide as a catalyst over a period of 2-6 hrs at room temperature. The product obtained was in 90% yields. The yield decrease or increase depends upon electron withdraw or electron donating group attached on the different positions of aryl. Followed by cyclization of chalcone derivaties, the reaction was carried out with hydrazine hydrate in formic acid, which gave derivatives of N-formyl-pyrazoline in good yield. Using Lewis acid (trifouroacetic acid) again cyclization of N-formyl-pyrazoline in acetonitrile, which gave pyrazoloisoindole. Along with chalcone, N-formylpyrazoline and final product (pyrazoloisoindole) was characterized on the basis of their spectroscopic data. All product were assigned on the basis of their ¹H NMR (500MHz), ¹³C NMR (125MHz), IR and GC-MS spectra analysis.

3.2 Characterization of Compound 1: {1 2-phenyl-3,3a-dihydropyrazolo[5,1a]isoindol-8-one (3a)}

Characterization of (E)-chalcone(1a)

In ¹HNMR 7.82 (d, J=15.5Hz, =CH-Ar, 1H), 7.59 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J_{Ha-Hb} = 12-16Hz) protons and ten aromatic protons were present from 8.08 – 7.30 ppm; IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1640-1660 cm⁻¹ and, peak at 1500-1400 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-chalcone.

Characterization of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde(1b)

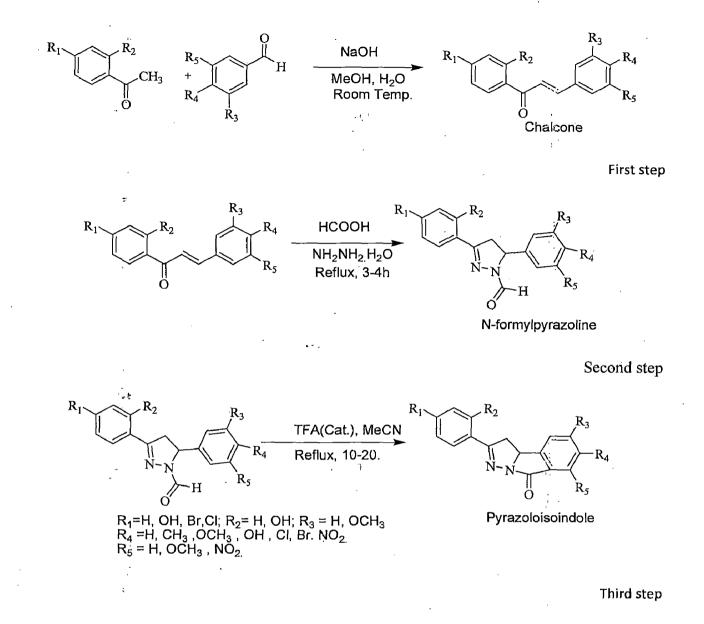
The product was characterization for the molecular formula $C_{16}H_{14}N_2O_{1}$ melting point 140-142⁰C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.21–3.25, 3.79–3.85 and 5.53–5.50 with J_{AB} = 17.5 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 12.0 Hz, respectively, formyl proton as a singlet at 8.90 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 42.37, a methine carbon at 58.36, C=N carbon at 148.6, and aldehyde carbon at 160.14 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1656, C=N at 1602 and C-N at 1142. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (1b). The molecular ion peak at 250 [M⁺, C₁₆H₁₄N₂O], 222, 221, 145(100%), 119, 114, 77, 65, 51. On the basis of these spectral data of product was characterized as 3,5-diphenyl-4,5dihydropyrazole-1-carbaldehyde.

Characterization of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(1c)

The product was characterization for the molecular formula $C_{16}H_{12}N_2O_5$ melting point 108-110°C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.19, 3.37 and 5.54 with J_{AB} = 17.5 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.76, a methine carbon at 60.33, C=N carbon at 138.47, and ketonic carbon at 156.87 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1692, C=N at 1401 and C-N at 1183. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (1c). The molecular ion peak at 248 [M⁺, C₁₆H₁₂N₂O], 241, 221, 215, 206, 205, 204, 178, 165, 146, 132, 115, 104, 103, 102. On the basis of these spectral data of product was characterized as 2-phenyl-3,3a-dihydropyrazolo[5,1a]isoindol-8-one.

43

1.





3.3 Characterization of Compound 2: {5-(4-chlorophenyl)-3-phenyl-4,5-di - hydropyrazole -1-carbaldehyde (2b)}

Characterization of (E)-3-(4-chlorophenyl)-1-phenylprop-2-en-1-one(2a)

In ¹HNMR 7.96 (d, J=15.5Hz, =CH-Ar, 1H), 7.63 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and nine aromatic protons were present from 8.06 – 7.40 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1642 cm⁻¹ and, peak at 1500-

1400 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-3-(4-chlorophenyl)-1-phenylprop-2-en-1-one.

Characterization of 5-(4-chlorophenyl)-3-phenyl-4,5-dihydropyrazole-1-carbaldehyde(2b)

The product was characterization for the molecular formula $C_{16}H_{13}CIN_2O$; melting point 102-104 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.21, 3.82 and 5.50 with J_{AB} = 16.9 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 11.5 Hz, respectively, formyl proton as a singlet at 8.98 ppm; ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.09, a methine carbon at 59.28, C=N carbon at 152.59, and aldehyde carbon at 160.22 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1664, C=N at 1604 and C-N at 1241. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (2b). The molecular ion peak at 284 [M⁺, C₁₆H₁₃ClN₂O], 286, 285, 255, 249, 207, 178, 97, 65, 51. On the basis of these spectral data of product was characterized as 5-(4-chlorophenyl)-3-phenyl-4,5dihydropyrazole-1-carbaldehyde.

Characterization of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(2c)

The product was characterization for the molecular formula $C_{16}H_{11}ClN_2O$ melting point 145-147 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.16, 3.73 and 5.50 with J_{AB} = 17.0 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.09, a methine carbon at 59.28, C=N carbon at 136.47, and ketonic carbon at 156.21 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1703, C=N at 1403 and C-N at 1196. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (2c). The molecular ion peak at 282 [M⁺, C₁₆H₁₁ClN₂O], 284, 283, 282, 271, 206, 191,104, 102. On the basis of these spectral data of product was characterized as 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

1.1.

3.4 Characterization of Compound 3: {6-chloro-2-(4-chlorophenyl)-3,3adihydro pyrazolo [5,1-a] isoindol-8-one (3c)}

Characterization of (E)-1, 3-bis(4-chlorophenyl)prop-2-en-1-one (3a)

In ¹HNMR 7.80 (d, J=15.5Hz, =CH-Ar, 1H), 7.46 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and eight aromatic protons were present from 8.03 – 7.41 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1655 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1, 3-bis(4-chlorophenyl)prop-2-en-1-one.

Characterization of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-carbaldehyde(3b)

The product was characterization for the molecular formula $C_{19}H_{18}Cl_2N_2O_2$ melting point 145-146^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.16, 3.80 and 5.50 with J_{AB} = 16.9 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 11.5 Hz, respectively, formyl proton as a singlet at 8.93 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.80, a methine carbon at 59.00, C=N carbon at 153.89, and aldehyde carbon at 160.19 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1676, C=N at 1480 and C-N at 1229. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (3c). The molecular ion peak at 318 [M⁺, C₁₉H₁₈Cl₂N₂O], 320, 318, 291,191,179,153,138(100%), 222, 209, 206, 194, 153, 145, 77, 69, 51. On the basis of these spectral data of product was characterized as 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (3c)

The product was characterization for the molecular formula $C_{16}H_{10}Cl_2N_2O_2$ melting point 146-148⁰C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.20, 3.78 and 5.52 with J_{AB} = 17.0 Hz, J_{AX} = 5.0Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.09, a methine carbon at 60.18, C=N carbon at 135.42, and ketonic carbon at 156.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1706,

C=N at 1410 and C-N at 1196. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (3c). The molecular ion peak at $316[M^+, C_{16}H_{10}Cl_2N_2O]$; 318, 317, 316, 281, 247, 171, 104, 102. On the basis of these spectral data of product was characterized as 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.5 Characterization of Compound 4: {6-bromo-2-phenyl-3,3a-dihydro pyrazolo[5,1-a]isoindol-8-one (4c)}

Characterization of (E)-3-(4-bromophenyl)-1-phenylprop-2-en-1-one(4a)

In ¹HNMR 7.68 (d, J=15.5Hz, =CH-Ar, 1H), 7.53 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and nine aromatic protons were present from 8.02 - 7.28 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1640 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-3-(4-bromophenyl)-1-phenylprop-2-en-1-one.

Characterization of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1-carbaldehyde(4b)

The product was characterization for the molecular formula $C_{16}H_{13}BrN_2O$ melting point 104-106 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.20, 3.82 and 5.50 with J_{AB} = 18.0 Hz, J_{AX} = 5.0 Hz, and J_{BX} = 12.5 Hz, respectively, formyl proton as a singlet at 8.95 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 42.45, a methine carbon at 60.53, C=N carbon at 155.66, and aldehyde carbon at 160.09 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1669, C=N at 1423 and C-N at 1134. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (4b). The molecular ion peak at 328 [M⁺, C₁₆H₁₃BrN₂O], 330, 329, 327, 299, 250, 222, 153, 145, 145, 77, 51. On the basis of these spectral data of product was characterized as 5-(4-bromophenyl)-3phenyl-4,5-dihydropyrazole-1-carbaldehyde .

47

S. D.

Characterization of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(4c)

The product was characterization for the molecular formula $C_{16}H_{11}BrN_2O$ melting point 162-164 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.15, 3.72 and 5.49 with J_{AB} = 17.5 Hz, J_{AX} = 5.0Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.09, a methine carbon at 59.89, C=N carbon at 137.53, and ketonic carbon at 156.32 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1691, C=N at 1401 and C-N at 1199. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (4c). The molecular ion peak at 326 [M⁺, C₁₆H₁₁BrN₂O]; 328, 327, 326, 247, 249, 171, 102. On the basis of these spectral data of product was characterized as 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.6 Characterization of Compound 5: {2-(4-bromophenyl)-3,3a-dihydro pyrazolo[5,1-a]isoindol-8-one (5c)}

Characterization of (E)-1-(4-bromophenyl)-3-phenylprop-2-en-1-one(5a)

In ¹HNMR 7.72 (d, J=15.5Hz, =CH-Ar, 1H), 7.28 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and nine aromatic protons were present from 8.02 – 7.28 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1651 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(4-bromophenyl)-3-phenylprop-2-en-1-one.

Characterization of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-carbaldehyde (5b)

The product was characterization for the molecular formula $C_{16}H_{13}BrN_2O$ melting point 145-147 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.20, 3.82 and 5.50 with J_{AB} = 17.5 Hz, J_{AX} = 5.0 Hz, and J_{BX} = 12.0 Hz, respectively, formyl proton as a singlet at 8.95 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 42.30, a methine carbon at 60.48, C=N carbon at 155.71, and aldehyde carbon at 159.93 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1668, C=N at 1413 and C-N at 1134. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (5b). The molecular ion peak at 328 [M⁺, $C_{16}H_{13}BrN_2O$], 330, 329, 327, 299, 250, 222, 153, 145, 145, 77, 51. On the basis of these spectral data of product was characterized as 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(5c)

The product was characterization for the molecular formula $C_{16}H_{11}BrN_2O$ melting point 146-148 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.23, 3.76 and 5.65 with J_{AB} = 17.5 Hz, J_{AX} = 5.5Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.05, a methine carbon at 60.52, C=N carbon at 137.28, and ketonic carbon at 156.89 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1697, C=N at 1391 and C-N at 1178. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (5c). The molecular ion peak at 326 [M⁺ C₁₆H₁₁BrN₂O]; 328, 327, 326, 247, 249, 171, 102. On the basis of these spectral data of product was characterized 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.7 Characterization of Compound 6: {6-bromo-2-(4-bromophenyl)-3,3adihydropyrazolo[5,1-a]isoindol-8-one (6c)}

Characterization of (E)-1,3-bis(4-bromophenyl)prop-2-en-1-one (6a)

In ¹HNMR 7.75 (d, J=15.5Hz, =CH-Ar, 1H), 7.22 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J_{Ha-Hb} = 12-16Hz) protons and eight aromatic protons were present from 8.07 – 7.20 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1660 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1,3-bis(4-bromophenyl)prop-2-en-1-one.

Characterization of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1-carbaldehyde(6b)

The product was characterization for the molecular formula $C_{16}H_{12}Br_2N_2O_1$, melting point 138-140 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.19, 3.79 and 5.53 with $J_{AB} = 17.5$ Hz, $J_{AX} = 5.0$ Hz, and $J_{BX} = 12.0$ Hz, respectively, formyl proton as a singlet at 8.93 ppm, ¹³CNMR

49

1. 4- 1.19

spectrum δ (ppm), revealed the presence of a methylene carbon at 40.80, a methine carbon at 60.00, C=N carbon at 151.89, and aldehyde carbon at 160.21 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1649, C=N at 1430 and C-N at 1129. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (6b). The molecular ion peak at 405 [M⁺, C₁₆H₁₂Br₂N₂O],408, 407, 376, 328, 248, 221, 207, 154, 91, 77, 51. On the basis of these spectral data of product was characterized as 3,5-bis (4-bromophenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 6-bromo-2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (6c)

The product was characterization for the molecular formula $C_{16}H_{11}BrN_2O$, melting point 138-140 ⁰C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_x appear as double doublets at 3.20, 3.80 and 5.55 with J_{AB} = 17.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.10, a methine carbon at 59.23, C=N carbon at 137.28, and ketonic carbon at 156.20 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1702, C=N at 1403 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (6c). The molecular ion peak at 404 [M⁺, C₁₆H₁₀ Br₂N₂O]; 407, 405, 403, 324, 247, 171, 104, 102. On the basis of these spectral data of product was characterized 6-bromo-2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.8 Characterization of Compound 7 :{2-(4-bromophenyl)-6-methyl-3,3adihydro –pyrazolo[5,1a]isoindol-8-one (7c)}

Characterization of (E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one(7a)

In ¹HNMR 7.89 (d, J=15.5Hz, =CH-Ar, 1H), 7.45 (d, J=10.5Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons, eight aromatic protons were present from 7.89 – 7.22 ppm and three methyl protons were present on 2.39 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1656 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one.

Characterization of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-carbaldehyde(7b)

The product was characterization for the molecular formula $C_{17}H_{15}BrN_2O$, melting point 180-182 ⁰C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.18, 3.76 and 5.49 with J_{AB} = 17.5 Hz, J_{AX} = 5.0 Hz, and J_{BX} = 12.0 Hz, respectively, formyl proton as a singlet at 8.93 ppm, methyl proton as a singlet at 2.18 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methyl carbon 21.14, methylene carbon at 42.51, a methine carbon at 59.06, C=N carbon at 154.67, and aldehyde carbon at 160.08 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1650, C=N at 1423 and C-N at 1122. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (7b). The molecular ion peak at 342 [M⁺, C₁₇H₁₅BrN₂O], 344, 343, 342, 327, 313, 249, 243, 221, 145, 104, 77, 51, 15. On the basis of these spectral data of product was characterized as 3-(4-bromophenyl)-5-p-tolyl-4,5dihydropyrazole-1-carbaldehyde.

Characterization of 2-(4-bromophenyl)-6-methyl-3,3a-dihydropyrazolo[5,1a]isoindol-8-one (7c)

The product was characterization for the molecular formula $C_{17}H_{13}BrN_2O$ melting point 165-167 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.20, 3.80 and 5.55 with J_{AB} = 17.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.10, a methine carbon at 59.23, C=N carbon at 137.28, and ketonic carbon at 156.20 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1702, C=N at 1413 and C-N at 1112. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (7c). The molecular ion peak at 340 [M⁺, C₁₇H₁₃BrN₂O]; 342, 341, 340, 324, 247, 171, 102. On the basis of these spectral data of product was characterized 2-(4-bromophenyl)-6-methyl-3,3a-dihydropyrazolo[5,1a]isoindol-8-one.

3.9 Characterization of Compound 8 :{ 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (8c)}

Characterization of (E)-1-(4-bromophenyl)-3-(3,4-dimethoxy- phenyl)prop-2-en-1-one (8a)

In ¹HNMR 7.34 (d, J=15.5Hz, =CH-Ar, 1H), 6.92 (d, J=8.5Hz, CO-CH=, 1H) indicated trans $(J_{Ha-Hb} = 12-16Hz)$ protons, eight aromatic protons were present from 7.89 – 7.22 ppm and six methoxy protons were present 3.92 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1659 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(4-bromophenyl)-3-(3,4-dimethoxy- phenyl)prop-2-en-1-one.

Characterization of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-dihydropyrazole-1carbaldehyde(8b)

The product was characterization for the molecular formula $C_{18}H_{17}BrN_2O_3$; melting point 118-120 °C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.18, 3.74 and 5.49 with $J_{AB} = 17.0$ Hz, $J_{AX} = 5.0$ Hz, and $J_{BX} = 11.0$ Hz, respectively, formyl proton as a singlet at 8.95 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.90, a methine carbon at 59.37, C=N carbon at 153.67, and aldehyde carbon at 160.26 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1659, C=N at 1421 and C-N at 1165. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (8c). The molecular ion peak at 388 [M⁺, $C_{18}H_{17}BrN_2O_3$], 390, 389, 388, 359, 357, 327, 309, 249, 221, 154, 119, 91, 77, 65, 51, 31. On the basis of these spectral data of product was characterized as 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydropyrazolo[5,1-a]isoindol -8-one (8c)

The product was characterization for the molecular formula $C_{18}H_{15}BrN_2O_3$; melting point 142-144 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.67, 3.76 and 5.75 with $J_{AB} = 16.5$ Hz, $J_{AX} = 4.5$ Hz, and $J_{BX} = 11.0$ Hz, respectively, methoxy hydrogen at 3.76 ppm, ¹³CNMR spectrum δ

(ppm), revealed the presence of a methoxy carbon at 54.36, methylene carbon at 40.10, a methine carbon at 59.23, C=N carbon at 147.53, and ketonic carbon at 156.32 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1695, C=N at 1409 and C-N at 1193. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (8c). The molecular ion peak at 386 [M⁺, C₁₈H₁₅BrN₂O₃]; 386, 387, 386, 307, 277, 247, 171, 102. On the basis of these spectral data of product was characterized 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.10 Characterization of Compound 9 :{ 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydropyrazolo[5,1-a]iso indol-8-one (9c)}

Characterization of (E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl) prop-2-en-1-one(9a)

In ¹HNMR 7.74 (d, J=15.5Hz, =CH-Ar, 1H), 6.92 (d, J=8.5Hz, CO-CH=, 1H) indicated trans $(J_{Ha-Hb} = 12-16Hz)$ protons, seven aromatic protons were present from 7.89 – 6.85 ppm and nine methoxy protons were present 3.99 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1661 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl) prop-2-en-1-one.

Characterization of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-dihydropyrazole-1carbaldehyde(9b)

The product was characterization for the molecular formula $C_{19}H_{19}BrN_2O_4$; melting point 175-177 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.16, 3.60 and 5.47 with $J_{AB} = 17.0$ Hz, $J_{AX} = 5.0$ Hz, and $J_{BX} = 12.0$ Hz, respectively, formyl proton as a singlet at 8.95 ppm, methoxy carbon signal at 3.80, ¹³CNMR spectrum δ (ppm), revealed the presence of a methoxy carbon at 56.21, methylene carbon at 40.90, a methine carbon at 60.81, C=N carbon at 153.52, and aldehyde carbon at 160.26 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1659, C=N at 1421 and C-N at 1165. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (9b). The molecular ion peak at 418 [M⁺, $C_{19}H_{19}BrN_2O_4$], 420, 419, 418, 390, 389, 388, 359, 357, 327, 309, 249, 221, 154, 119, 91, 77, 65,

53

51, 31. On the basis of these spectral data of product was characterized as 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydropyrazolo[5,1-a]iso indol-8-one(9c)

The product was characterization for the molecular formula $C_{19}H_{17}BrN_2O_4$, melting point 188-190 °C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.79, 3.80 and 5.56 with $J_{AB} = 18.0$ Hz, $J_{AX} = 4.5$ Hz, and $J_{BX} = 11.5$ Hz, respectively, methoxy hydrogen at 3.80 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methoxy carbon at 56.14, methylene carbon at 41.80, a methine carbon at 60.79, C=N carbon at 147.53, and ketonic carbon at 158.0 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1695, C=N at 1418 and C-N at 1164. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (9c). The molecular ion peak at 416 [M⁺, C₁₉H₁₇BrN₂O₄]; 418, 417, 416, 385, 355, 337, 324, 247, 171. On the basis of these spectral data of product was characterized 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.11 Characterization of Compound 10 :{ 5,6,7-trimethoxy-2-phenyl-3,3adihydro -pyrazolo[5,1-a]isoindol-8-one (10c)}

Characterization of (E)-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one(10a)

In ¹HNMR 7.82 (d, J=15.5Hz, =CH-Ar, 1H), 7.53 (d, J=8.5Hz, CO-CH=, 1H) indicated trans $(J_{Ha-Hb} = 12-16Hz)$ protons, seven aromatic protons were present from 8.04 – 7.22 ppm and nine methoxy protons were present 3.83 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1641 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-phenyl-3-(3,4,5-trimethoxyphenyl)prop-2-en-1-one.

Characterization of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carb aldehyde(10b)

The product was characterization for the molecular formula $C_{19}H_{20}N_2O_4$; melting point 145-146 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.23, 3.80 and 5.47 with $J_{AB} = 17.0$ Hz, $J_{AX} =$ 5.0 Hz, and $J_{BX} = 12.0$ Hz, respectively, formyl proton as a singlet at 9.00 ppm, methoxy carbon signal at 3.83, ¹³CNMR spectrum δ (ppm), revealed the presence of a methoxy carbon at 56.15, methylene carbon at 40.90, a methine carbon at 60.90, C=N carbon at 153.77, and aldehyde carbon at 160.26 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1667, C=N at 1419 and C-N at 1226. GC-MS for EIMS fragmentations followed the established pättern, where further confirmed the product (10c). The molecular ion peak at 340 [M⁺, $C_{19}H_{20}N_2O_4$], 340, 339, 311, 237, 222, 209, 206, 194, 19, 153, 145, 77(100%), 76,65, 61. On the basis of these spectral data of product was characterized as 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde.

Characterization of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (10c)

The product was characterization for the molecular formula $C_{19}H_{18}N_2O_4$; melting point 175-178 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.27, 3.49 and 5.57 with J_{AB} = 16.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 12.0 Hz, respectively, methoxy hydrogen at 3.80 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methoxy carbon at 56.14, methylene carbon at 41.96, a methine carbon at 60.59, C=N carbon at 147.53, and ketonic carbon at 158.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1702, C=N at 1421 and C-N at 1130. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (10c). The molecular ion peak at 338 [M⁺, C₁₉H₁₈N₂O₄],340, 339, 311, 236(100%), 222, 209, 206, 194, 153, 145, 77, 69, 51. On the basis of these spectral data of product was characterized 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

5.44.5

3.12 Characterization of Compound 11 :{ 6-chloro-2-(2-hydroxyphenyl)-3,3adi -hydropyrazolo[5,1-a]isoindol-8-one (11c)}

Characterization of (E)-3-(4-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1-one(11a)

In ¹HNMR 7:73 (d, J=16.0Hz, =CH-Ar, 1H), 7.55 (d, J=11.0Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and eight aromatic protons were present from 3.01 - 7.45 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1645 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-3-(4-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1-one.

Characterization of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-dihydropyrazole-1-carb - aldehyde(11b)

The product was characterization for the molecular formula $C_{16}H_{13}CIN_2O_2$; melting point 185-187 °C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.33, 3.90 and 5.45 with J_{AB} = 17.0 Hz, J_{AX} = 5.0 Hz, and J_{BX} = 11.5 Hz, respectively, formyl proton as a singlet at 8.90 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.96, a methine carbon at 60.90, C=N carbon at 153.05, and aldehyde carbon at 160.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1670, C=N at 1428 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (11b). The molecular ion peak at 300 [M⁺, C₁₆H₁₃CIN₂O₂], 302, 301, 300, 283, 265, 271, 249, 221, 173, 97, 77, 51. On the basis of these spectral data of product was characterized as 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(11c)

The product was characterization for the molecular formula $C_{16}H_{11}ClN_2O_2$; melting point 157-159 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A , H_B and H_X appear as double doublets at 3.38, 3.94 and 5.57 with $J_{AB} = 17.0$ Hz, $J_{AX} = 4.0$ Hz, and $J_{BX} = 12.0$ Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.80;, a methine carbon at 60.36, C=N carbon at 141.12, and ketonic

carbon at 156.85 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1694, C=N at 1407 and C-N at 1225. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (11c). The molecular ion peak at 298 [M⁺, C₁₆H₁₁ClN₂O₂]; 300, 299, 298, 281, 263, 247, 171, 77. On the basis of these spectral data of product was characterized 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.13 Characterization of Compound 12 :{ 2-(4-hydroxyphenyl)-3,3a-dihydro pyrazolo[5,1-a]isoindol-8-one(12c)}

Characterization of (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one(12a)

In ¹HNMR 7.80 (d, J=15.5Hz, =CH-Ar, 1H), 7.65 (d, J=16.0Hz, CO-CH=, 1H) indicated trans (J _{Ha-Hb} = 12-16Hz) protons and nine aromatic protons were present from 8.46 - 7.51 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1651 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one.

Characterization of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydropyrazole-1-carbaldehyde (12b)

The product was characterization for the molecular formula $C_{16}H_{14}N_2O_2$; melting point 128-130 °C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.23, 3.79 and 5.49 with J_{AB} = 17.0 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 12.0 Hz, respectively, formyl proton as a singlet at 9.00 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.14, a methine carbon at 59.56, C=N carbon at 151.12, and aldehyde carbon at 160.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1660, C=N at 1428 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (12b). The molecular ion peak at 266 [M⁺, C₁₆H₁₄N₂O₂], 268, 267, 266, 249, 221, 173, 97, 51. On the basis of these spectral data of product was characterized as 3-(4-hydroxyphenyl)-5phenyl-4,5-dihydropyrazole-1-carbaldehyde.

1. 小学校学校 计公式选择的

Characterization of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(12c)

The product was characterization for the molecular formula $C_{16}H_{12}N_2O_2$; melting point 126-128 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.15, 3.26 and 5.52 with J_{AB} = 17.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.12;, a methine carbon at 60.63, C=N carbon at 141.13, and ketonic carbon at 156.38 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1669, C=N at 1451 and C-N at 1222. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (12c). The molecular ion peak at 264 [M⁺, C₁₆H₁₂N₂O₂]; 266, 265, 264, 247, 171, 102, 77, 55. On the basis of these spectral data of product was characterized 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.14 Characterization of Compound 13 :{7-hydroxy-2-phenyl-3,3a-dihydro pyrazolo[5,1-a]isoindol-8-one(13c)}

Characterization of (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one(13a)

In ¹HNMR 7.84 (d, J=15.5Hz, =CH-Ar, 1H), 7.45 (d, J=12.0Hz, CO-CH=, 1H) indicated trans (J_{Ha-Hb} = 12-16Hz) protons and nine aromatic protons were present from 8.03 - 6.92 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1661 cm⁻¹and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-3-(4-hydroxyphenyl)-1-phenylprop-2-en-1-one.

Characterization of 5-(3-hydroxyphenyl)-3-phenyl-4,5-dihydropyrazole-1-carbaldehyde (13b)

The product was characterization for the molecular formula $C_{16}H_{14}N_2O_2$, melting point 176-178 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.23, 3.79 and 5.49 with $J_{AB} = 17.2$ Hz, $J_{AX} =$ 4.5 Hz, and $J_{BX} = 12.0$ Hz, respectively, formyl proton as a singlet at 9.00 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.84, a methine carbon at 59.36, C=N carbon at 152.65, and aldehyde carbon at 159.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1660, C=N at 1428 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (13b). The molecular ion peak at 266 [M^+ , $C_{16}H_{14}N_2O_2$]; 268, 267, 266, 249, 221, 173, 97, 51. On the basis of these spectral data of product was characterized as 5-(3-hydroxyphenyl)-3-phenyl-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one(13c)

The product was characterization for the molecular formula $C_{16}H_{12}N_2O_2$; melting point 148-150 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.26, 3.78 and 5.53 with J_{AB} = 17.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 11.5 Hz, respectively,¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.84, a methine carbon at 61.36, C=N carbon at 141.10, and ketonic carbon at 156.40 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1679, C=N at 1402 and C-N at 1217. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (13c). The molecular ion peak at 264 [M⁺, C₁₆H₁₂N₂O₂]; 266, 265, 264, 247, 171, 102, 77, 55. On the basis of these spectral data of product was characterized 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.15 Characterization of Compound 14 :{ 2-(2-hydroxyphenyl)-6-nitro-3,3adihydropyrazolo[5,1-a]isoindol-8-one(14c)}

Characterization of (E)-1-(2-hydroxyphenyl)-3-(4-nitrophenyl)prop-2-en-1-one(14a)

In ¹HNMR 7.86 (d, J=15.5Hz, =CH-Ar, 1H), 7.23 (d, J=12.0Hz, CO-CH=, 1H) indicated trans (J_{Ha-Hb} = 12-16Hz) protons and eight aromatic protons were present from 8.00 - 7.14 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1651 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(2-hydroxyphenyl)-3-(4-nitrophenyl)prop-2-en-1-one.

59

S. A. Start

Characterization of 3-(2-hydroxyphenyl)-5-(4-nitrophenyl)-4,5-dihydropyrazole-1-carb – aldehyde(14b)

The product was characterization for the molecular formula $C_{16}H_{13}N_3O_4$; melting point 178-180 $^{\circ}$ C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.20, 3.75 and 5.53 with J_{AB} = 17.0 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 12.0 Hz, respectively, formyl proton as a singlet at 9.00 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.96, a methine carbon at 59.14, C=N carbon at 151.23, and aldehyde carbon at 160.32 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1667, C=N at 1428 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (14b). The molecular ion peak at 311 [M⁺, C₁₆H₁₃N₃O₄]; 313, 312, 311, 265, 249, 221, 219, 144, 97,77, 51,45. On the basis of these spectral data of product was characterized as 3-(2hydroxyphenyl)-5-(4-nitrophenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 2-(2-hydroxyphenyl)-6-nitro-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (14c)

The product was characterization for the molecular formula $C_{16}H_{11}N_3O_2$; melting point 156-158 °C. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.15, 3.70 and 5.55 with J_{AE} = 17.5 Hz, J_{AX} = 4.5Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.96, a methine carbon at 60.82, C=N carbon at 143.85, and ketonic carbon at 156.45 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1696, C=N at 1418 and C-N at 1215. GC-MS for EIMS fragmentations followed the established pattern, where for further confirmed the product (14c). The molecular ion peak at 309 [M⁺, C₁₆H₁₁N₃O₂]; 311, 310, 309, 292, 247, 171, 102, 77, 55. On the basis of these spectral data of product was characterized 2-(2-hydroxyphenyl)-6-nitro-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

3.16 Characterization of Compound 15 :{ 2-(4-chlorophenyl)-7-nitro-3,3adihydropyrazolo[5,1-a]isoindol-8-one(15c)}

Characterization of (E)-1-(4-chlorophenyl)-3-(3-nitrophenyl)prop-2-en-1-one(15a)

In ¹HNMR 8.25 (d, J=15.5Hz, =CH-Ar, 1H), 7.91 (d, J=12.0Hz, CO-CH=, 1H) indicated trans (J_{Ha-Hb} = 12-16Hz) protons and eight aromatic protons were present from 8.49 – 7.56 ppm. IR v_{max} (KBr, cm⁻¹): vibration frequency for the carbonyl group at 1641 cm⁻¹ and, peak at 1500-600 cm⁻¹ indicate the presence of aromatic bonds. On the basis of these spectral data of product was characterized as (E)-1-(4-chlorophenyl)-3-(3-nitrophenyl)prop-2-en-1-one.

Characterization of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydropyrazole-1-carb aldehyde(15b)

The product was characterization for the molecular formula $C_{16}H_{12}ClN_3O_3$; melting point 139-141 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.21, 3.89 and 5.65 with J_{AB} = 16.0 Hz, J_{AX} = 5.0 Hz, and J_{BX} = 11.0 Hz, respectively, formyl proton as a singlet at 8.96 ppm, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 41.14, a methine carbon at 59.36, C=N carbon at 140.10, and aldehyde carbon at 160.41 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1636, C=N at 1420 and C-N at 1219. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (15b). The molecular ion peak at 329 [M⁺, C₁₆H₁₂ClN₃O₃], 331, 330, 329, 294, 249, 221, 144, 97, 77, 51, 29. On the basis of these spectral data of product was characterized as 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydropyrazole-1-carbaldehyde.

Characterization of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one (15c)

The product was characterization for the molecular formula $C_{16}H_{10}ClN_3O_3$; melting point 180-182 ^oC. In ¹H NMR gave characteristic peak in δ (ppm), an ABX- type pattern was observed where H_A, H_B and H_X appear as double doublets at 3.23, 3.92 and 5.58 with J_{AB} = 17.5 Hz, J_{AX} = 4.5 Hz, and J_{BX} = 11.5 Hz, respectively, ¹³CNMR spectrum δ (ppm), revealed the presence of a methylene carbon at 40.14, a methine carbon at 60.36, C=N carbon at 142.35, and ketonic carbon

61

Same and the second second

at 156.40 ppm. The FT-IR spectrum in v_{max} (KBr, cm⁻¹) gave a strong band for C=O at 1701, C=N at 1425 and C-N at 1215. GC-MS for EIMS fragmentations followed the established pattern, where further confirmed the product (15c). The molecular ion peak at 327 [M⁺, C₁₆H₁₀ClN₃O₃]; 329, 328, 327, 292, 291, 171, 77, 55. On the basis of these spectral data of product was characterized 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one.

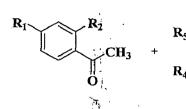
References

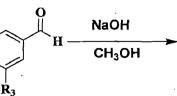
- Srinivasan B, Thomas E. Johnson, Rahul Lad, and Chengguo Xing; J. Med. Chem. 2009, 52, 7228–7235.
- Xiaoliu Li, Hideyo Takahashi, Hiro Ohtake and Shiro Ikegami; *Tetrahedron Letters*, 2004, 45, 4123–4126.
- (a) Weber, F. G.; Brosche, K.; Seedorf, C.; Rinow, A. Monatsh. Chem. 1969, 100, 1924; (b) Joshi, M. G.; Wadodkar, K. N. Indian J. Chem. 1981, 20B, 1090; (c) Sharma, T. C.; Pawar, S. R.; Reddy, N. J. Acta Chim. Hung. 1983, 112, 159; (d) Dhar, D. N.; Raghunathan, R. Indian J. Chem. 1984, 23B, 1187; (e) Orlov, V. D.; Aziz, M. A.; Mchedlov-Petrosyan, N. O.; Asoka, P. K. D. Khim. Geterotsikl, Soedin. 1985, 1511, (f) Sachchar, S. P.; Singh, A. K. J. Indian Chem. Soc. 1985, 62, 142.
- 4. (a) Lévai, A.; Szöllősy, Á; Tóth, G. J. Chem. Research (S) 1985, 392, (b) Tóth, G.; Szöllősy, Á.; Lóránd, T.; Kónya, T.; Szabó, D.; Földesi, A.; Lévai, A. J. Chem. Soc. 1989, 2, 319.

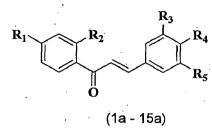
124

਼ੀ ਦ

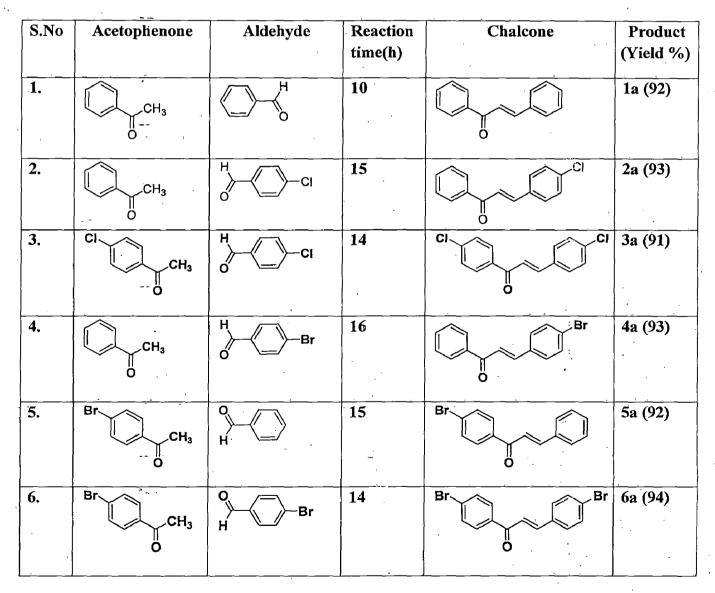
Table 1 Construction of chalcones by a variety of acetophenones and avariety of aromatic aldehydes.

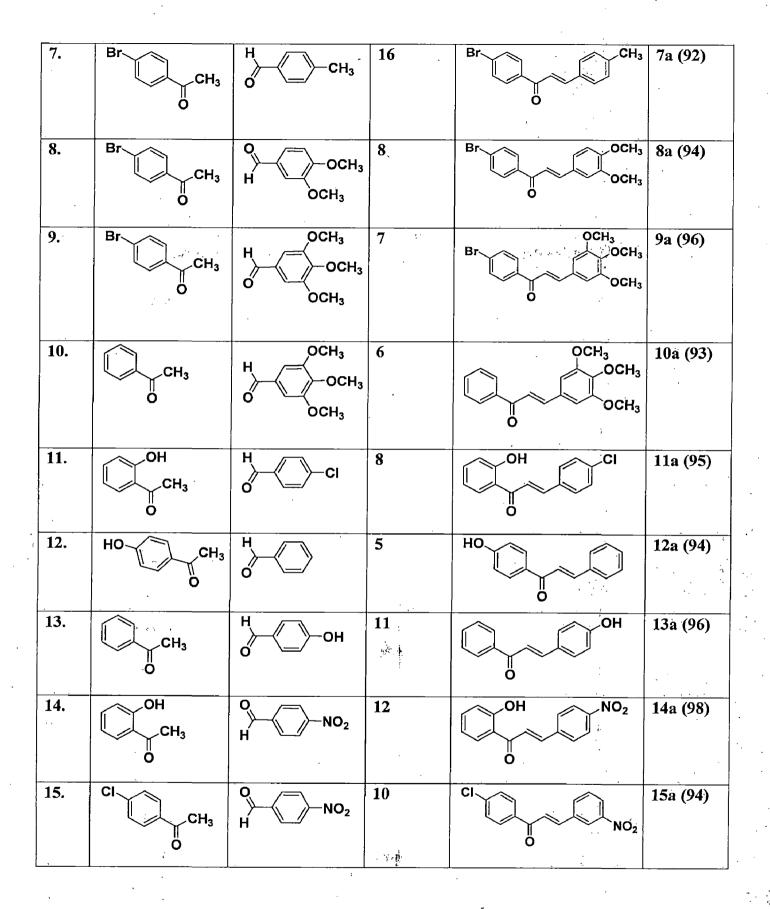






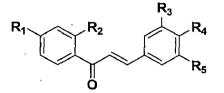
 $\begin{array}{l} {\sf R}_1 = {\sf H}, \, {\sf OH}, \, {\sf Br}, {\sf CI}; \, {\sf R}_2 {=} \, {\sf H}, \, {\sf OH}; \, {\sf R}_3 {=} \, {\sf H}, \, {\sf OCH}_3 \\ {\sf R}_4 = {\sf H}, \, {\sf CH}_3 \, , {\sf OCH}_3 \, , \, {\sf OH} \, , \, {\sf CI}, \, {\sf Br}, \, {\sf NO}_2 \\ {\sf R}_5 {=} \, {\sf H}, \, {\sf OCH}_3 \, , \, {\sf NO}_2 \\ \end{array}$



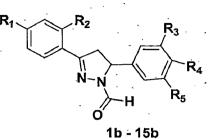


a fair a start of the start of the

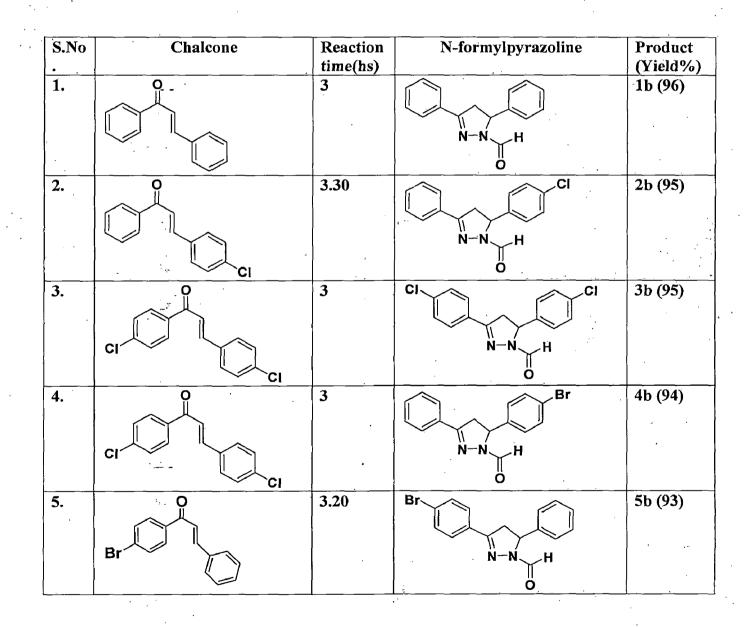
Table 2 Formation of N-formylpyrazoline from chalcone

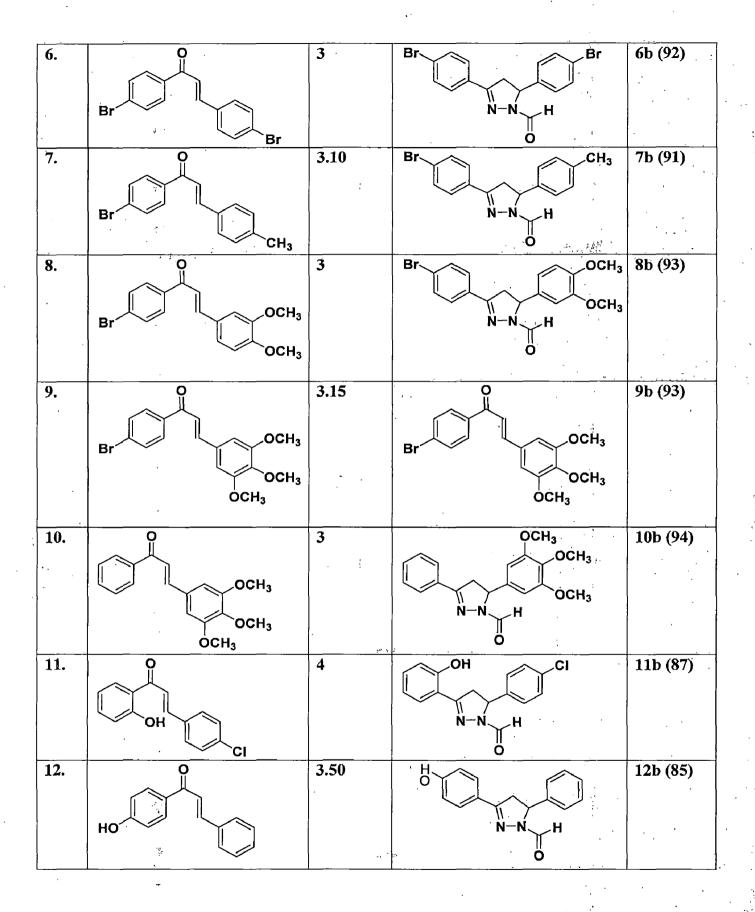


NH2NH2,H2O HCOOH, Reflux, 3-4h



 $\begin{array}{l} \mathsf{R}_1 = \mathsf{H}, \, \mathsf{OH}, \, \mathsf{Br}, \mathsf{CI}; \, \mathsf{R}_2 = \mathsf{H}, \, \mathsf{OH}; \, \mathsf{R}_3 = \mathsf{H}, \, \mathsf{OCH}_3 \\ \mathsf{R}_4 = \mathsf{H}, \, \mathsf{CH}_3 \, , \mathsf{OCH}_3 \, , \, \mathsf{OH} \, , \, \mathsf{CI}, \, \mathsf{Br}. \, \mathsf{NO}_2 \\ \mathsf{R}_5 = \mathsf{H}, \, \mathsf{OCH}_3 \, , \, \mathsf{NO}_2 \\ \end{array}$





and the second second

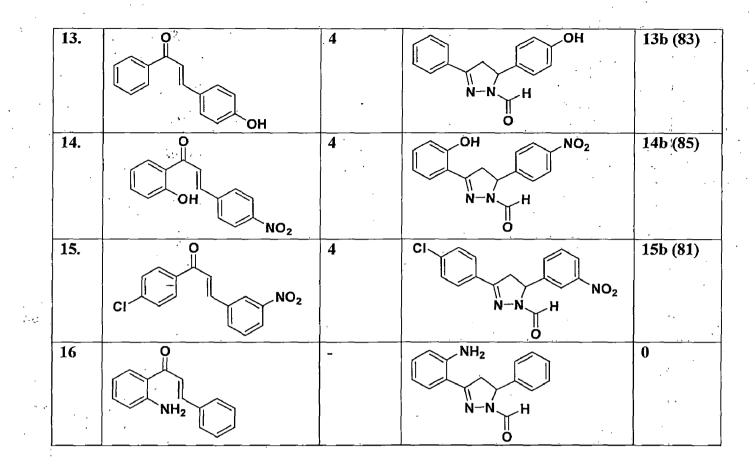
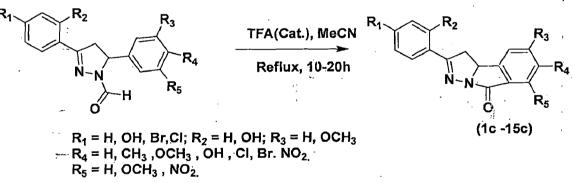
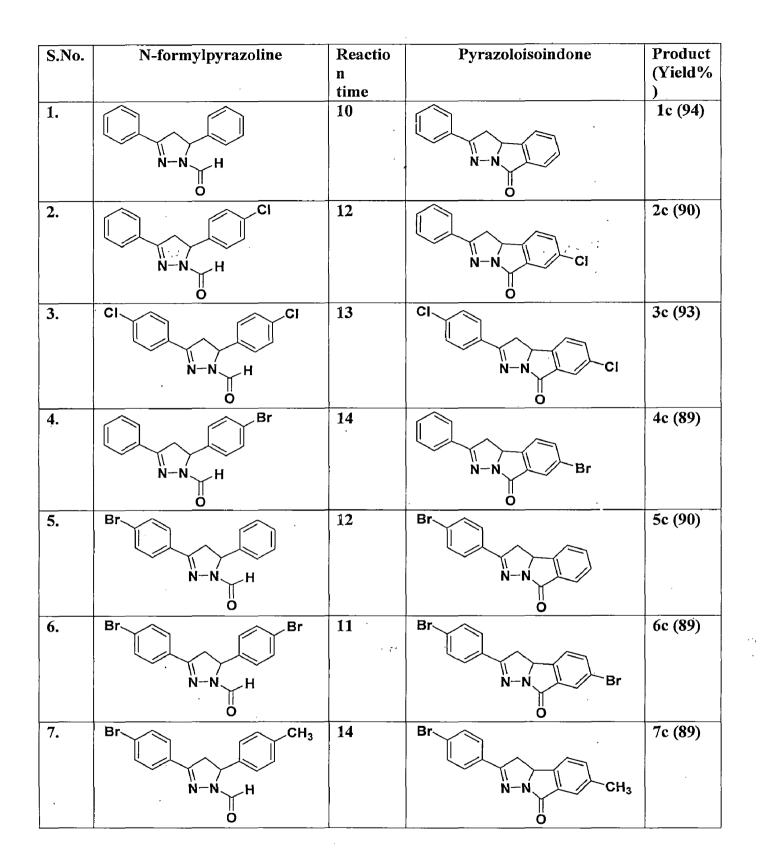
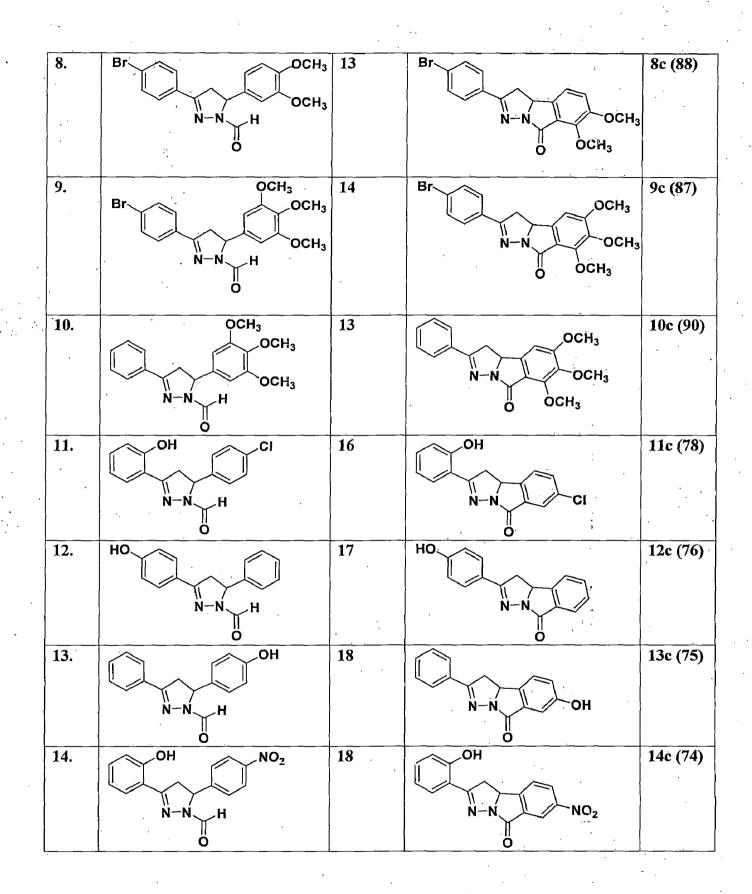
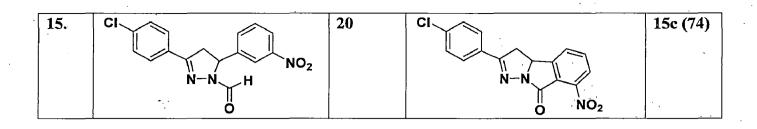


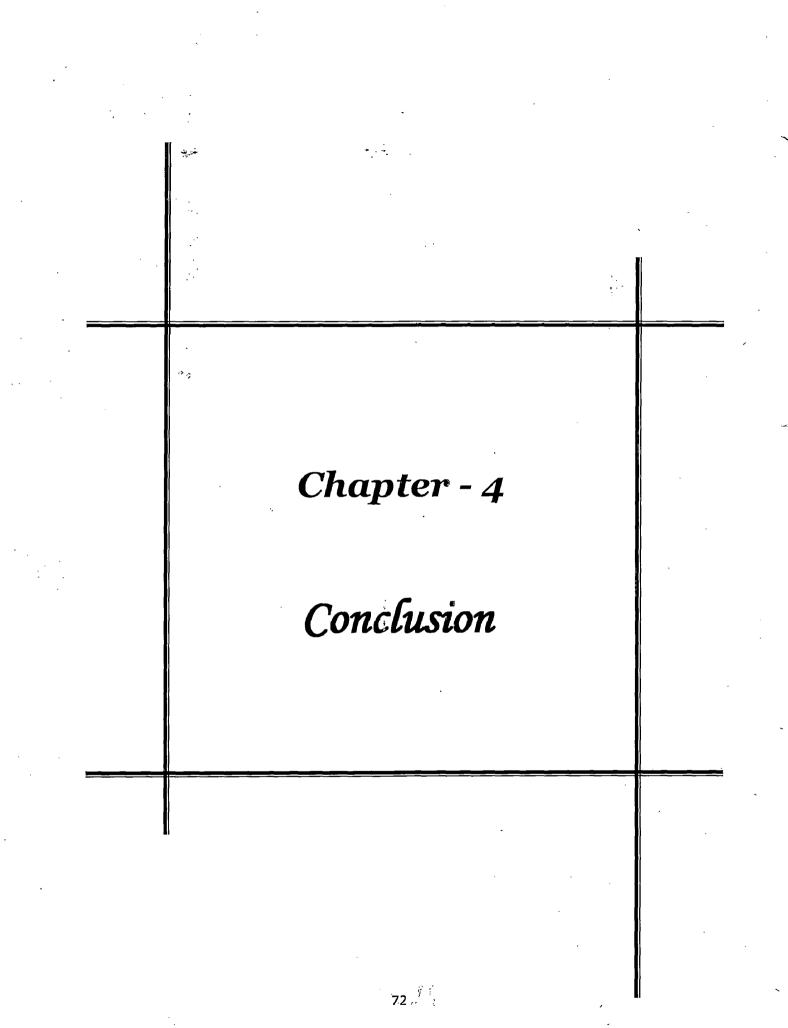
Table 3 Formation of Pyrazoloisoindole from N-formylpyrazoline











Conclusion

An alternate synthetic approach of 2-(substituted aryl)-3,3a-dihydro-8H-pyrazolo [5,1-a] isoindol-8-ones via chalcone based N-formyl-pyrazolines is described. N-formyl-pyrazolines(**1b-15b**) were prepared in excellent yield (81-96%) by the reaction of chalcones (**1a-15a**) with hydrazine hydrate in presence of formic acid, which undergoes intramolecular Friedal-Craft acylation in the presence of trifluoroacetic acid(TFA) as a catalyst to afford functionalized 2-(substituted phenyl)-3,3a-dihydro-8H-pyrazolo[5,1-a]isoindol-8-one (**3c-15c**) in good to excellent yield (74-94%) at refluxed in acetonitrile. Our synthetic route avoids expensive reagents and significantly improved the yield.

SUPPORTING INFORMATION

S.No.	-Title Name	Pase No.
Figure 1a	¹ H NMR spectra of chalcone	79
Figure 1b	¹ H NMR spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde	80
Figure 1b	¹³ C NMR spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde	81
Figure 1b	IR spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde	82
Figure 1b	GC-MS spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde	83
Figure 1c	¹ H NMR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one	84
Figure 1c	¹³ C NMR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-	85
	one	
Figure 1c	IR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one	86
Figure 1c	GC-MS spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one	87
Figure 2b	¹ H NMR spectra of 5-(4-chlorophenyl)-3-phenyl-4,5-dihydropyrazole-1-	88
	carbaldehyde	
Figure 2b	IR spectra of 5-(4-chlorophenyl)-3-phenyl-4,5-dihydropyrazole-1-	89
-	carbaldehyde	
Figure 2c	¹ H NMR spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-	90
	a]isoindol-8-one	
Figure 2c	¹³ C NMR spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-	91
Ū	a]isoindol-8-one	
Figure 2c	GC-MS spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-	92
	a]isoindol-8-one	
Figure 2c	IR spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-	93
1 iguite 20	one	20
Elauro 2h	¹ H NMR spectra of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-	. 94
Figure 3b		
F ' 1	carbaldehyde	95
Figure 3b	GC-MS spectra of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-	73
	carbaldehyde	

Figure 3c	¹ H NMR spectra of 6-chloro-2-(4-chlorophenyl)-3,3aihydropyrazolo	96
	[5,1-a]isoindol-8-one.	
Figure 3c	¹³ C NMR spectra of 6-chloro-2-(4-chlorophenyl)-3,3a-	97
	hydropyrazolo[5,1-a]isoindol-8-one.	
Figure 4b	¹ H NMR spectra of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1-	98
	carbaldehyde	
Figure 4b	¹³ C NMR spectra of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-	99
	1-carbaldehyde	
Figure 4b	IR spectra of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1-	100
	carbaldehyde	
Figure 4c	¹ H NMR spectra of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]	101
	isoindol-8-one.	
Figure 4c	IR spectra of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-	102
	one	
Figure 5b	¹ H NMR spectra of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-	103
	carbaldehyde	
Figure 5b	IR spectra of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-	104
	carbaldehyde	
Figure 5c	¹ H NMR spectra of 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-	105
	a]isoindol-8-one.	
Figure 5c	IR spectra of 2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-	106
	one	
Figure 6b	¹ H NMR spectra of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1-	107
•	carbaldehyde	、 、
Figure 6c	¹ H NMR spectra of 6-bromo-2-(4-bromophenyl)-3,3a-	108
	dihydropyrazolo[5,1-a]isoindol-8-one.	
Figure 6c	IR spectra of 6-bromo-2-(4-bromophenyl)-3,3a-dihydropyrazolo[5,1-	109
	a]isoindol-8-one	
Figure 7a	¹ H NMR spectra of (E)-1-(4-bromophenyl)-3-p-tolylprop-2-en-1-one	110
Figure 7b	¹ H NMR spectra of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-	111
	carbaldehyde	

75

:

÷

Figure 7b	¹³ C NMR spectra of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-	112
· .	1-carbaldehyde	
Figure7c	H NMR spectra of 2-(4-bromophenyl)-6-methyl-3,3a-dihydropyrazolo	113
	[5,1a] iso- indol-8-one	
Figure 7c	¹³ C NMR spectra of 2-(4-bromophenyl)-6-methyl-3,3a-dihydropyrazolo	114
<u>د</u>	[5,1a] isoindol-8-one	
Figure 8a	¹ H NMR spectra of (E)-1-(4-bromophenyl)-3-(3,4-imethoxyphenyl)-	115
•	prop-2-en-1-one	
Figure 8b	¹ H NMR spectra of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-	116
	dihydro pyrazole-1-carbaldehyde	
Figure 8b	¹³ C NMR spectra of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-	117
	dihydro -pyrazole-1-carbaldehyde	· .
Figure 8b	IR spectra of 3(4-bromophenyl)-5-(3,4-dimethoxyphenyl)-4,5-dihydro	118
0	pyrazole-1-carbaldehyde	
Figure 8c	¹ H NMR spectra of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydro	. 119
·	pyrazolo[5,1-a]isoindol-8-one	
		1.00
Figure 8c	¹³ C NMR spectra of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydro-	120
	pyrazolo[5,1-a]isoindol-8-one	· · · ·
Figure 8c	IR spectra of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydropyrazolo	121
	[5,1-a]isoindol-8-one	÷
Figure 9a	¹ H NMR spectra of (E)-1-(4-bromophenyl)-3-(3,4,5-trimethoxyphenyl)	122
	prop-2-en-1-one	
Figure 9b	¹ H NMR spectra of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-	123
	dihydro- pyrazole-1-carbaldehyde	
Figure 9b	IR spectra of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-	124
C C	dihydro pyrazole-1-carbaldehyde	
Figure 9c	¹ H NMR spectra of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydro	125
0	pyrazolo[5,1-a]isoindol-8-one	
Figure 9c	¹³ C NMR spectra of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydro	126
Tigute 20		120
	pyrazolo[5,1-a]isoindol-8-one	

Figure 9c	IR spectra of 2-(4-bromophenyl)-5,6,7-trimethoxy-3,3a-dihydropyrazolo	127
	[5,1-a]iso- indol-8-one	
Figure 10b	¹ H NMR spectra of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydro	128
0	pyrazole-1-carbaldehyde	
Figure 10b	¹³ C NMR spectra of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydro	129
0	pyrazole-1-carbaldehyde	,
Figure 10b	IR spectra of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-	130
	1-carbaldehyde	· · ·
Figure 10c	¹ H NMR spectra of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo	131
	[5,1-a] isoindol-8-one	• •
Figure 10c	¹³ C NMR spectra of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo	.132
	[5,1-a] isoindol-8-one	•
Figure 10c	IR spectra of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]iso	133
	indol-8-one	2
Figure 11 b	¹ H NMR spectra of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-	134
	dihydro pyrazole-1-carbaldehyde	•
Figure 11 b	IR spectra of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-dihydro-	135
	pyrazole-1-carbaldehyde	· .
Figure 11c	¹ H NMR spectra of 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydropyrazolo	136
	[5,1-a] isoindol-8-one	· · ·
Figure 11c	¹³ C NMR spectra of 6-chloro-2-(2-hydroxyphenyl)-3,3a-	137
	dihydropyrazolo [5,1-a]isoindol+8+one	
Figure 11c	IR spectra of 6-chloro-2-(2-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-	138
	a]isoindol-8-one	
Figure 12b	¹ H NMR spectra of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydropyrazole-	139
	1-carbaldehyde	
Figure 12b	¹³ C NMR spectra of 3-(4-hydroxyphenyl)-5-phenyl-4,5-dihydro	140
	pyrazole-1-carbaldehyde	
Figure 12c	¹ H NMR spectra of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]iso	141
	indol-8-one	
Figure 12c	¹³ C NMR spectra of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-	142

_	a]iso indol-8-one	·
Figure 12c	IR spectra of 2-(4-hydroxyphenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-	143
	8-one	
Figure 13b	¹ HNMR spectra of 5-(3-hydroxyphenyl)-3-phenyl-4,5-dihydropyrazole-	144
	1-carbaldehyde	
Figure 13c	¹ H NMR spectra of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]iso	145.
	indol-8-one	
Figure 13c	¹³ C NMR spectra of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]iso	146
	indol-8-one	
Figure 13c	IR spectra of 7-hydroxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-	147
	8-one	
Figure 14b	¹ H NMR spectra of 3-(2-hydroxyphenyl)-5-(4-nitrophenyl)-4,5-dihydro	148
	pyrazole-1-carbaldehyde	
Figure 14c	¹ HNMR spectra of 2-(2-hydroxyphenyl)-6-nitro-3,3a-dihydropyrazolo-	149
	[5,1-a] isoindol-8-one.	
Figure 15b	¹ H NMR spectra of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydro-	150
	pyrazole-1-carbaldehyde	
Figure 15b	IR spectra of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydropyrazole-	151
	1-carbaldehyde	
Figure 15c	¹ H NMR spectra of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo	152
	[5,1-a]iso- indol-8-one	
Figure 15c	¹³ C NMR spectra of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo	153-
	[5,1-a] isoindol-8-one	
Figure 15c	IR spectra of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo[5,1-a]iso	154
	indol-8-one	

7[:]8

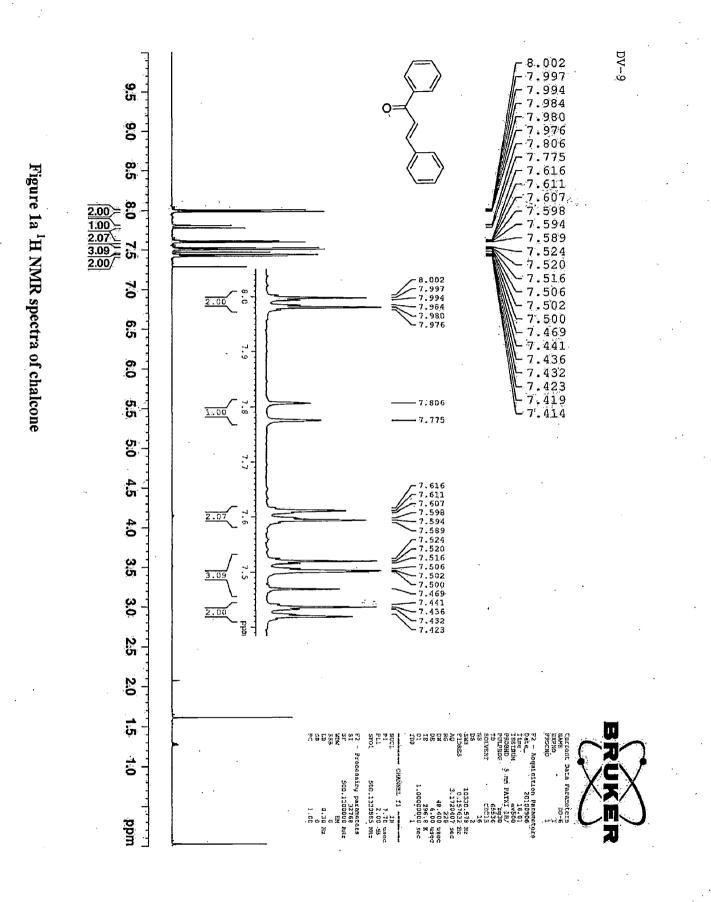
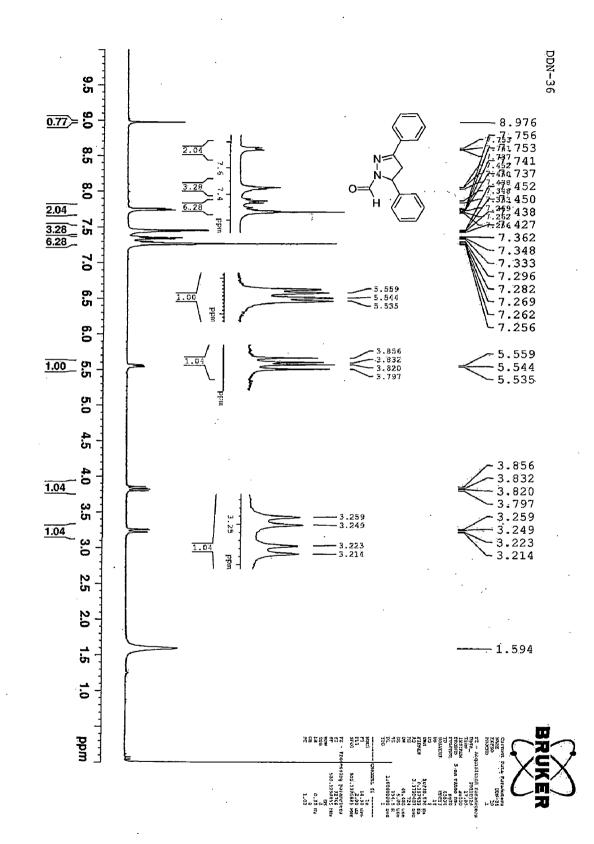


Figure 1b ¹H NMR spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde



80

с^ал

i. Er

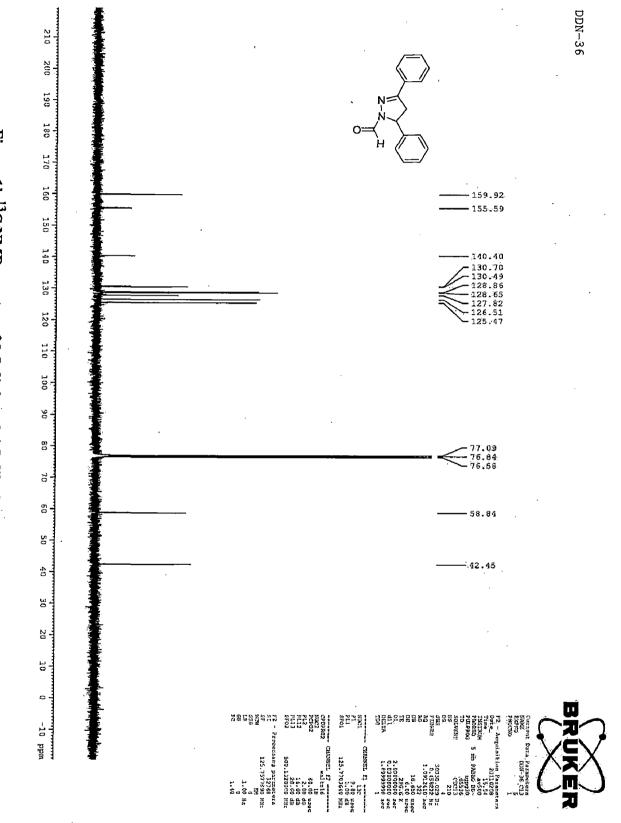
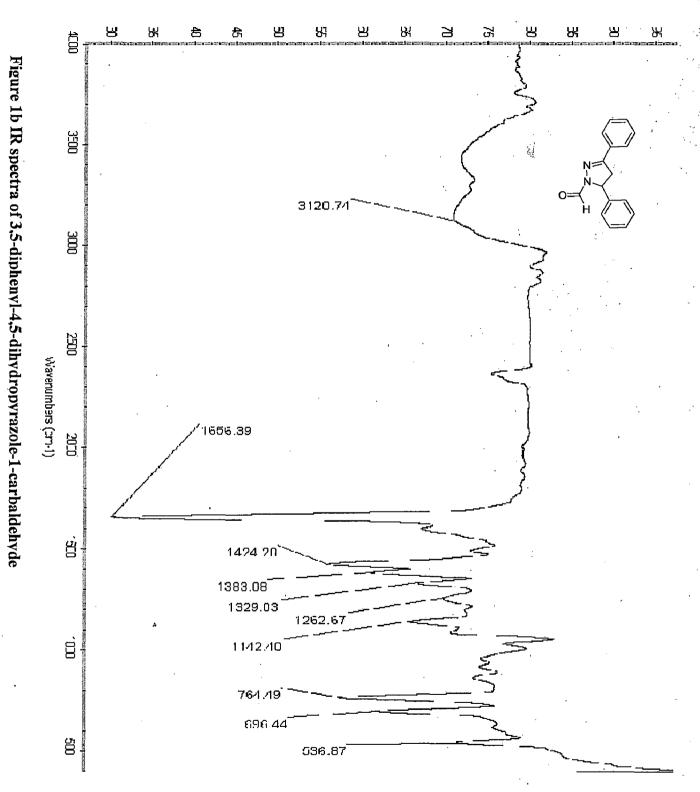


Figure 1b ¹³C NMR spectra of 3,5-diphenyl-4,5-dihydropyrazole-1-carbaldehyde



%Transmittance

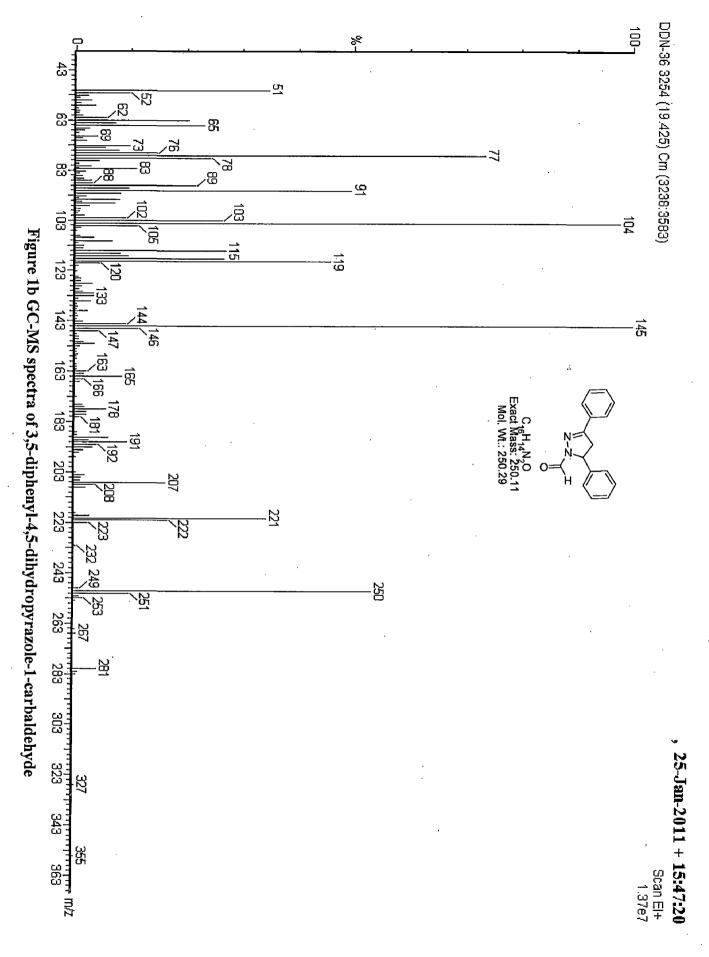
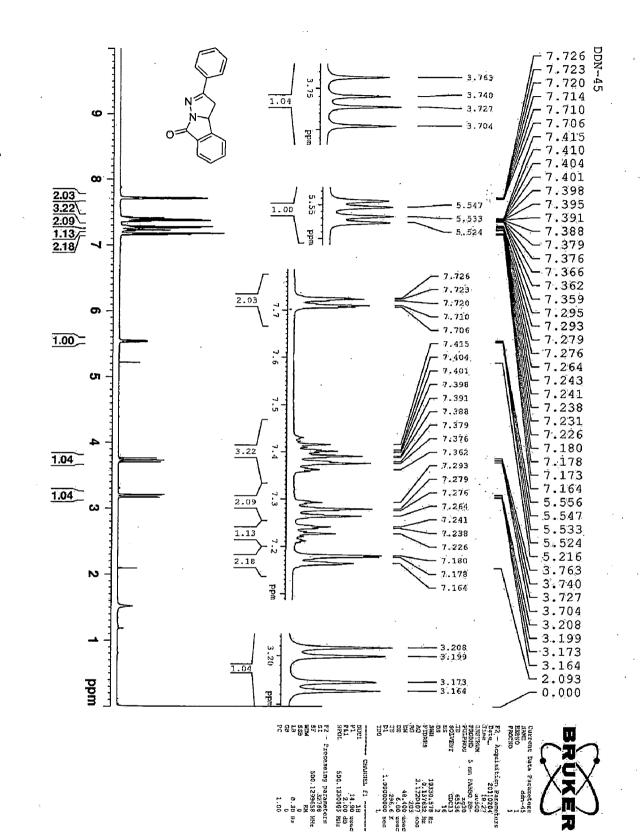


Figure 1c¹H NMR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one



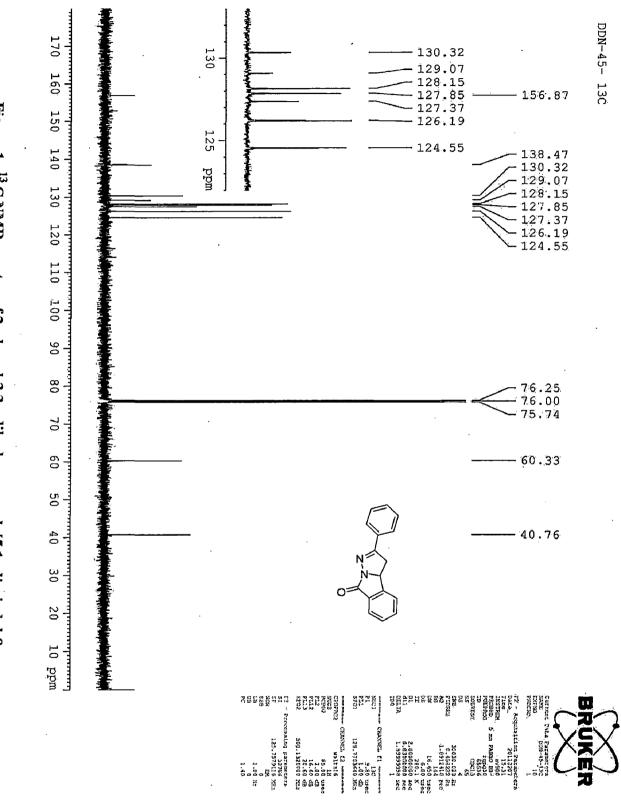
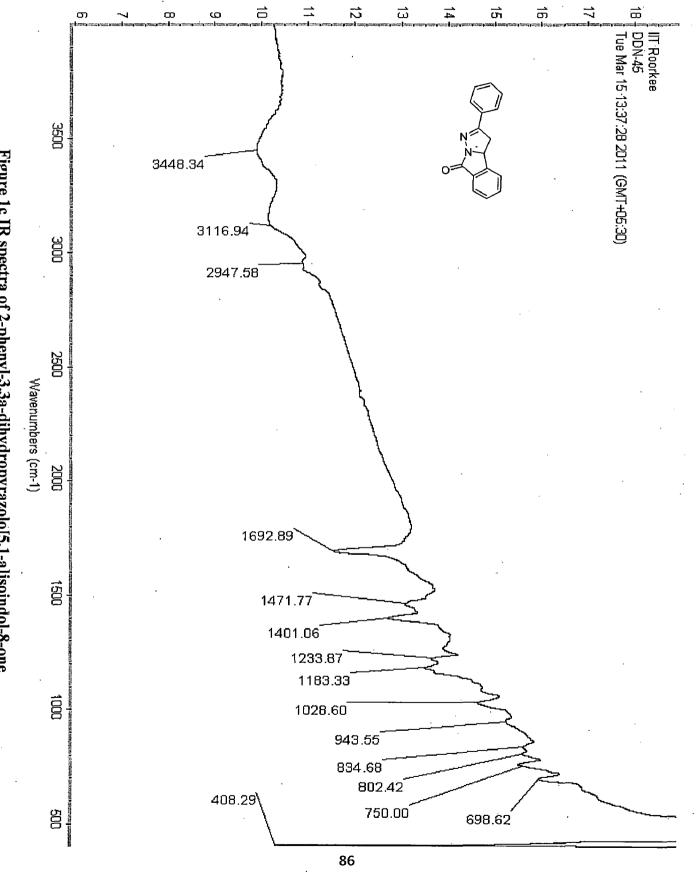


Figure 1c¹³ C NMR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one

85

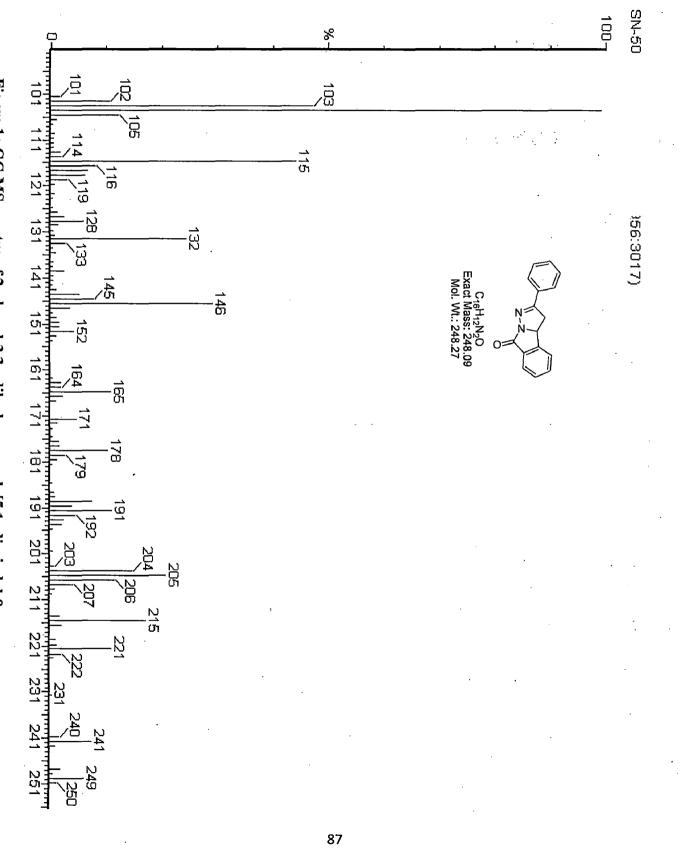
Ц 70

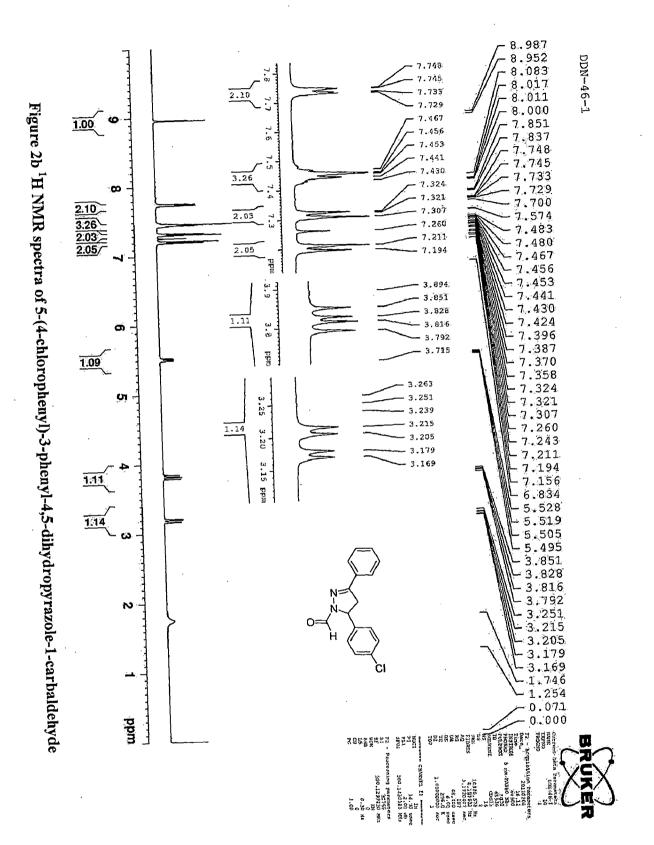


%Transmittance

Figure 1c IR spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one

Figure 1c GC-MS spectra of 2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one





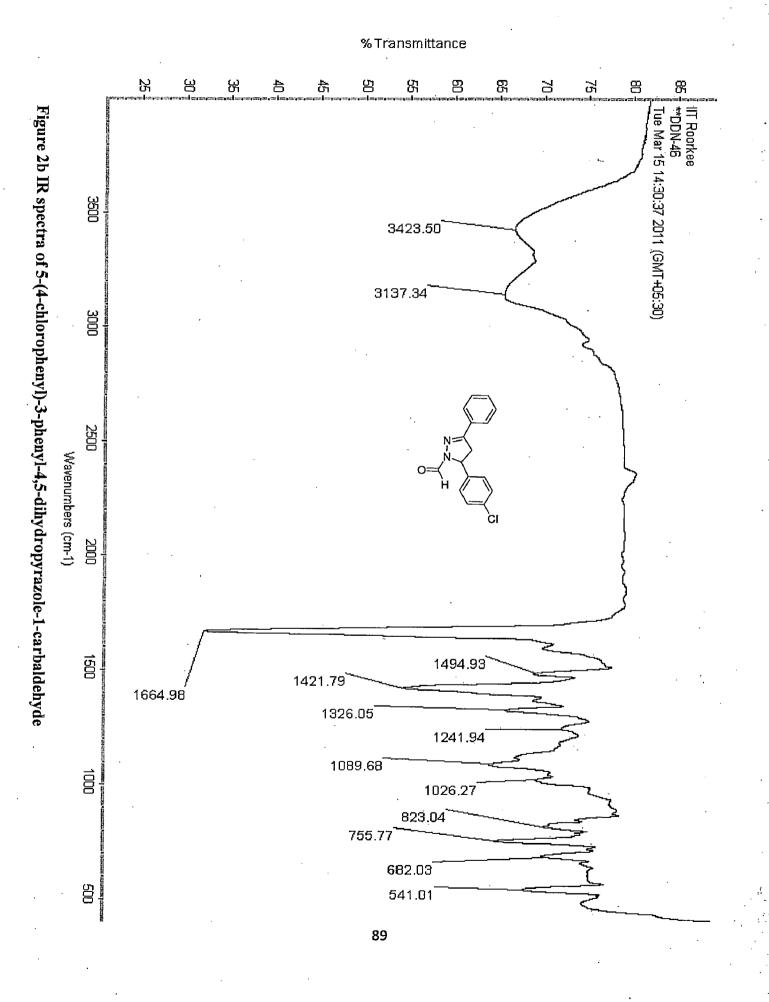
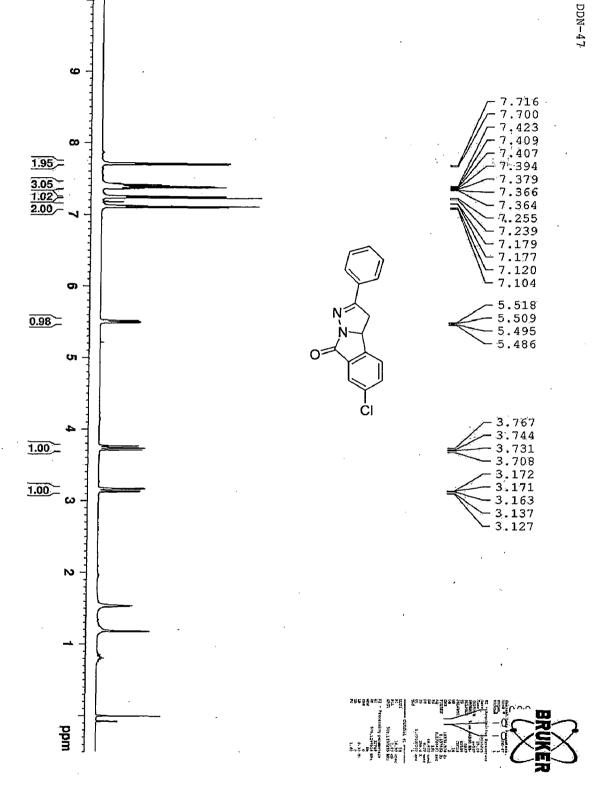
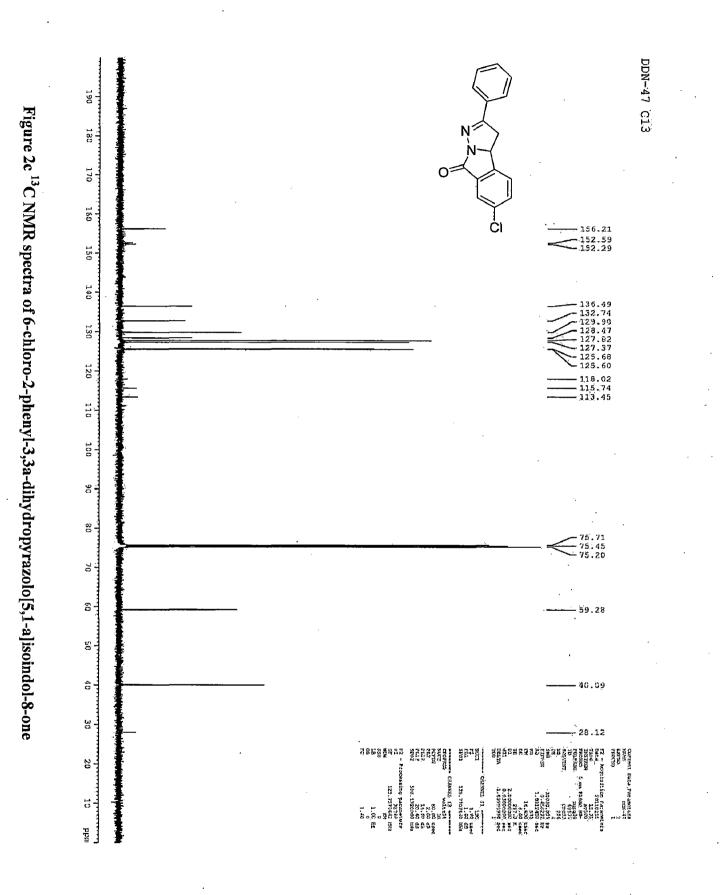


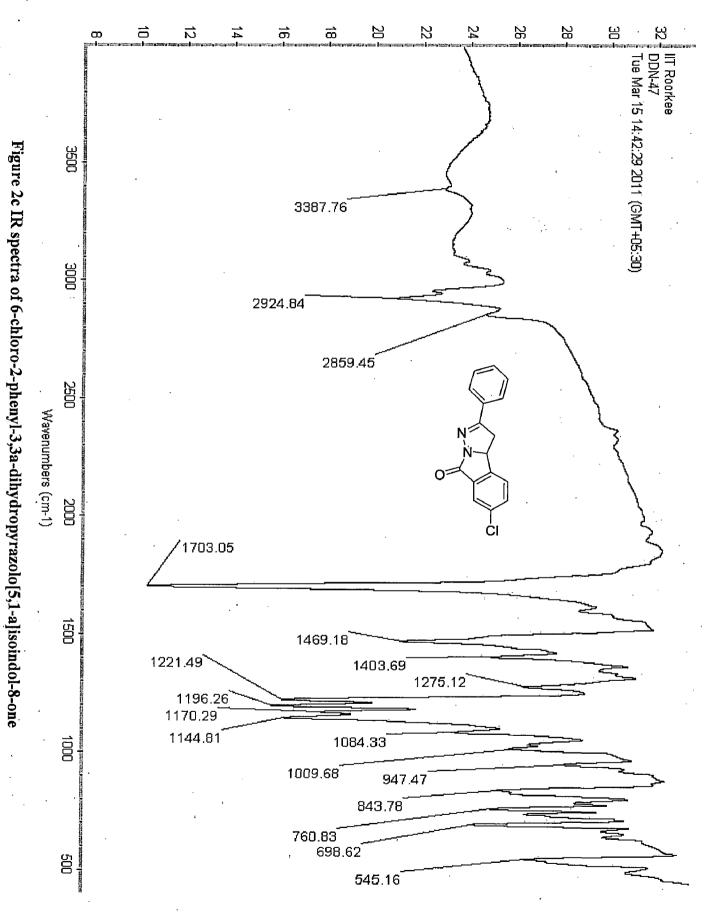
Figure 2c¹H NMR spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one





[.] 91

:



% Transmittance



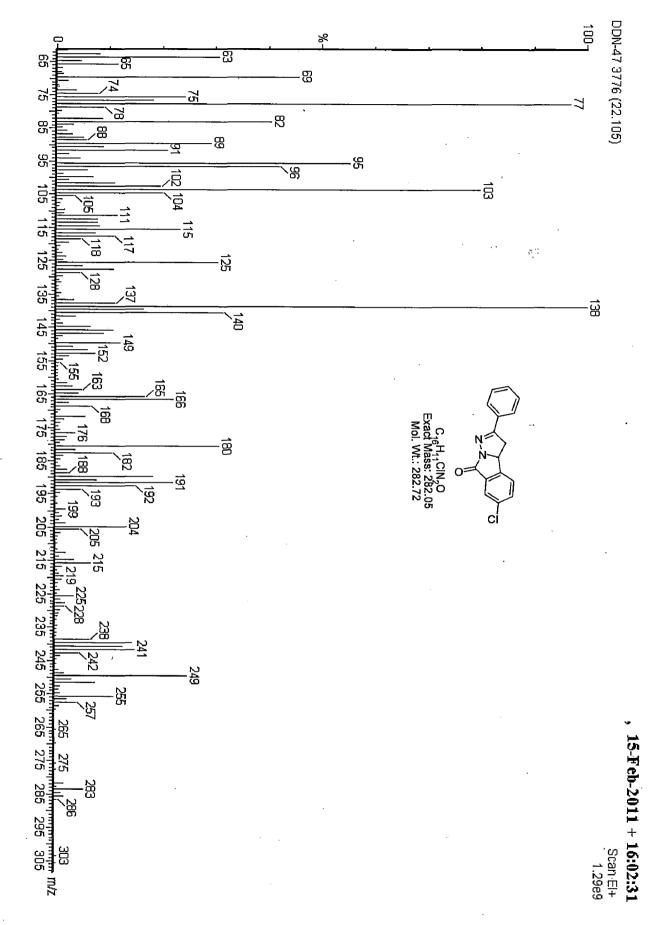


Figure 2c GC-MS spectra of 6-chloro-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one

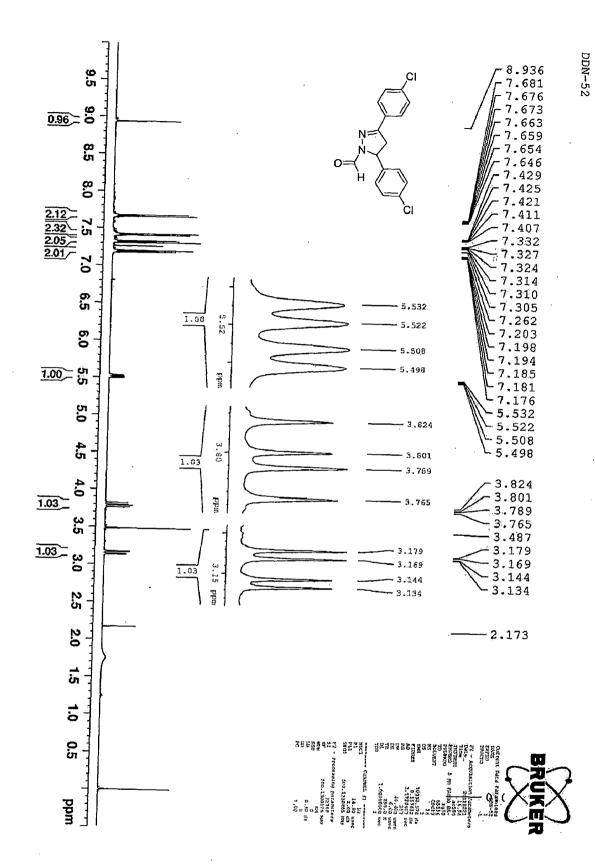


Figure 3b¹H NMR spectra of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-carbaldehyde

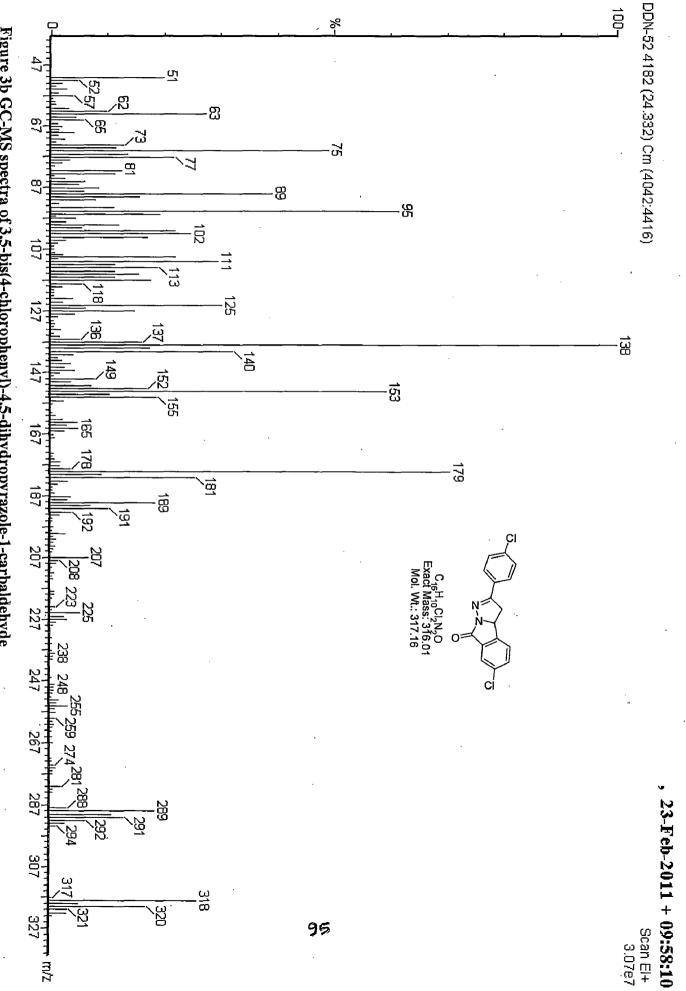
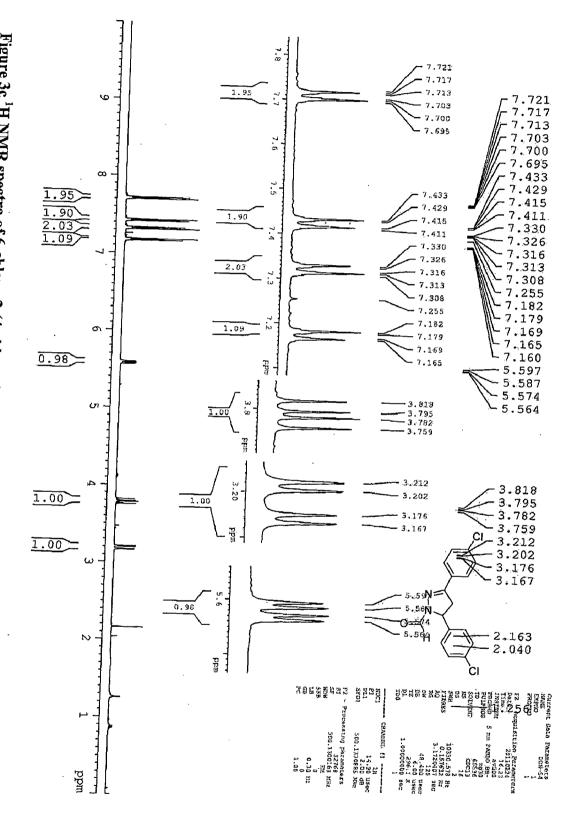


Figure 3b GC-MS spectra of 3,5-bis(4-chlorophenyl)-4,5-dihydropyrazole-1-carbaldehyde

Figure 3c¹H NMR spectra of 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one



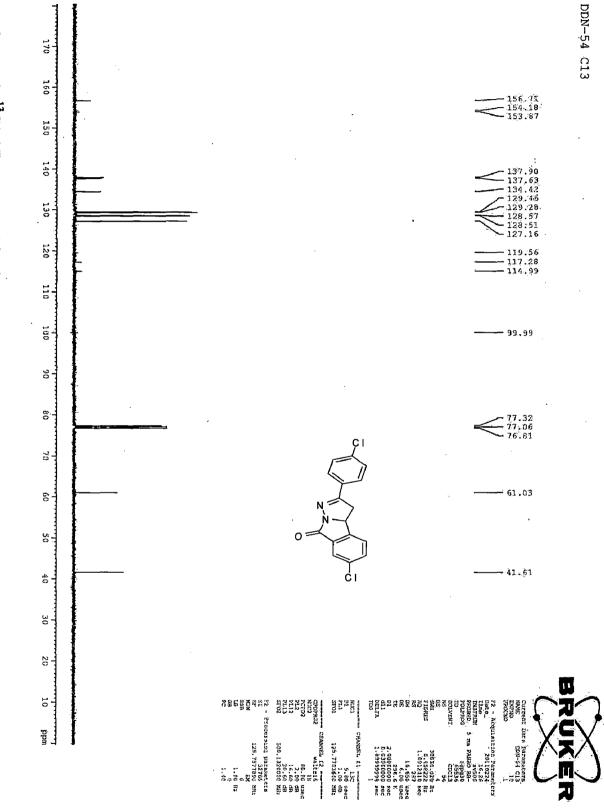
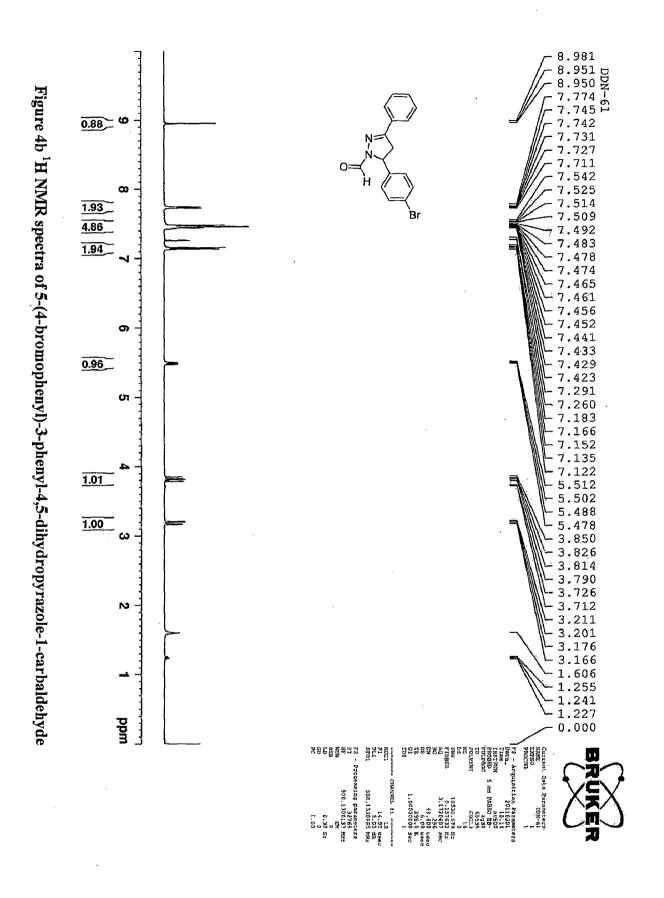


Figure 3c ¹³C NMR spectra of 6-chloro-2-(4-chlorophenyl)-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one



>

۰,

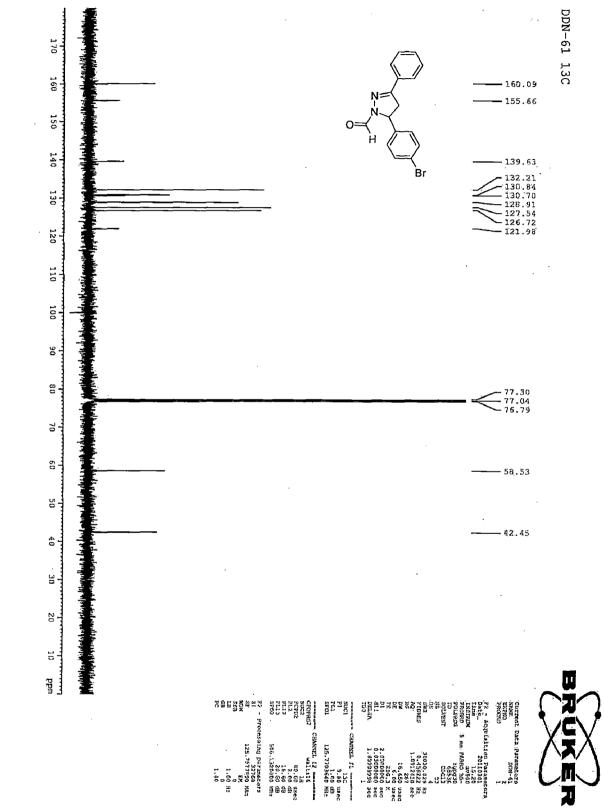
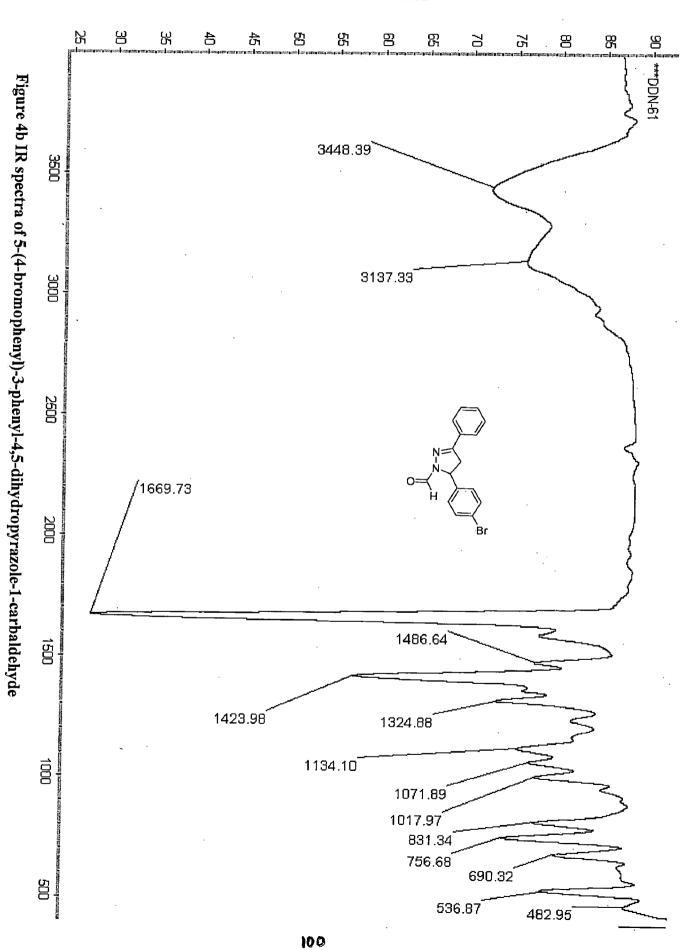


Figure 4b ¹³C NMR spectra of 5-(4-bromophenyl)-3-phenyl-4,5-dihydropyrazole-1-carbaldehyde

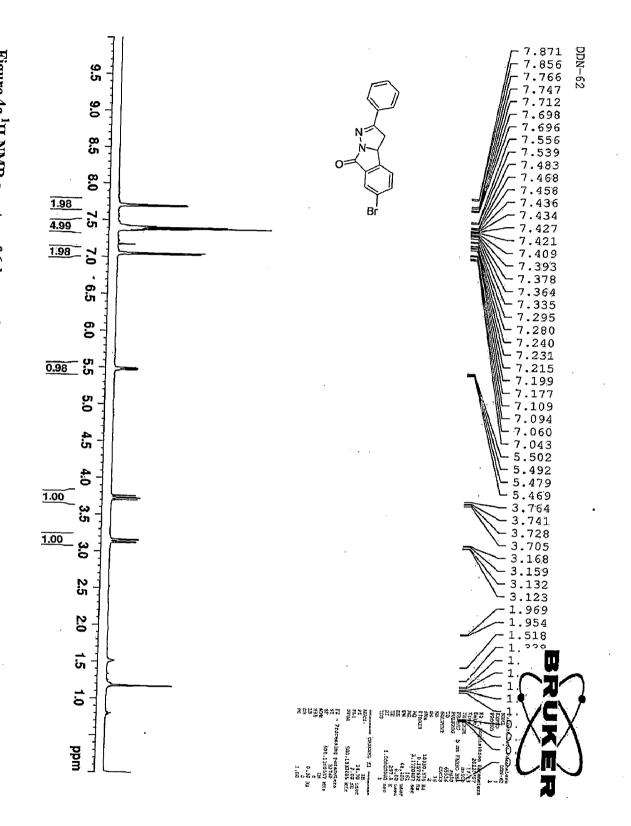
`. X



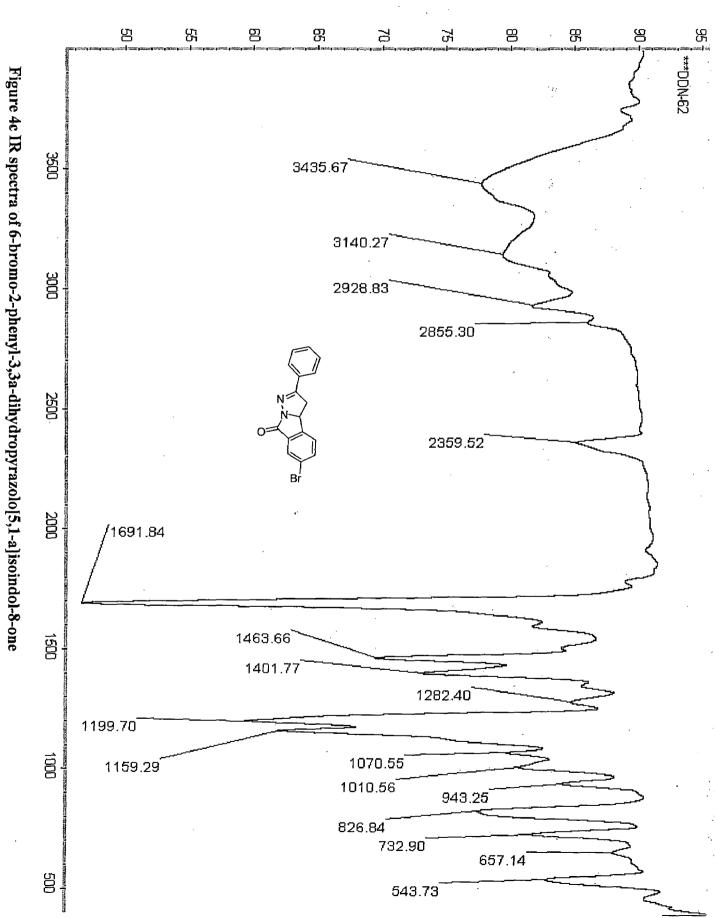
%Transmittance

Ę

Figure 4c ¹H NMR spectra of 6-bromo-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one



•



% Transmittance

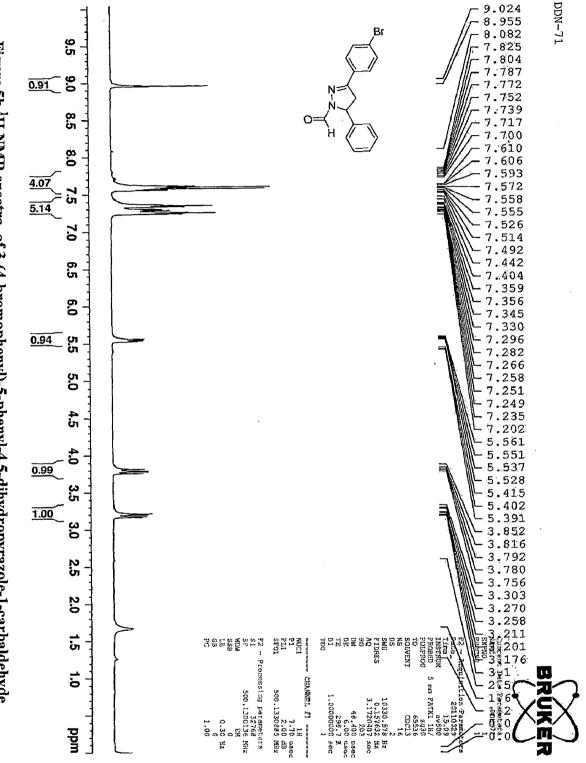
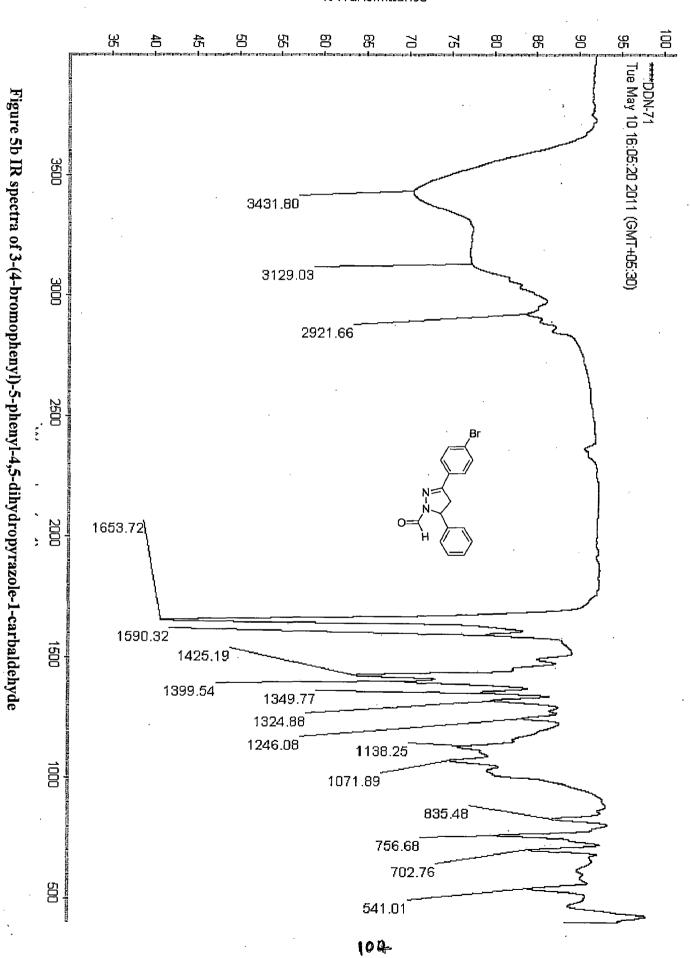
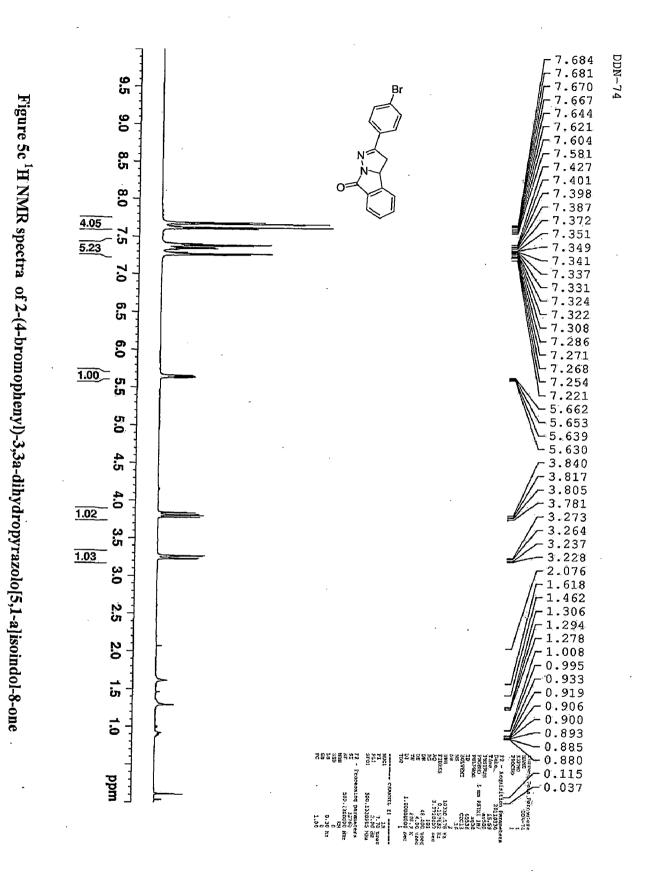
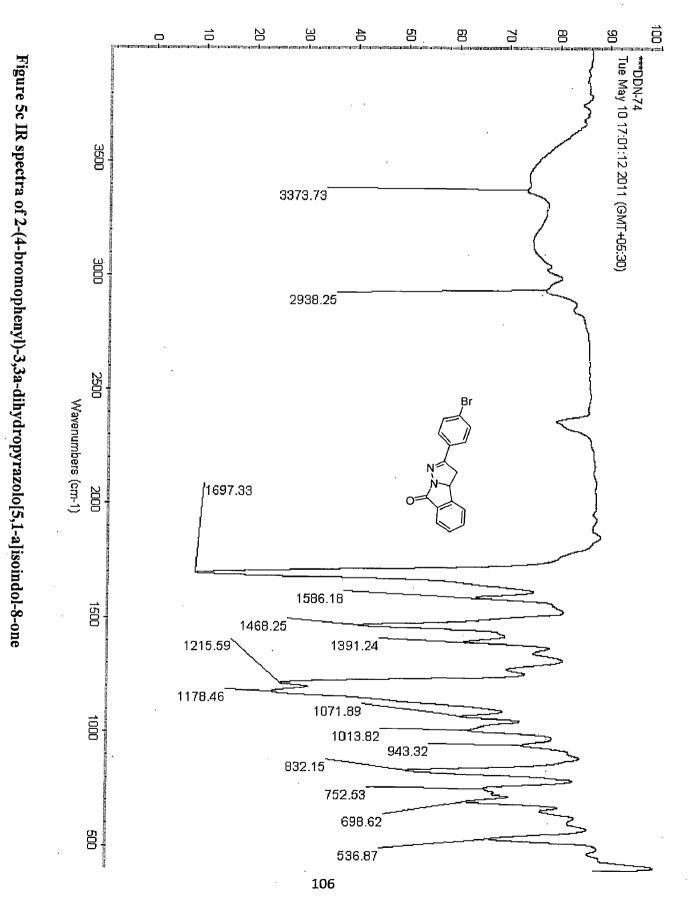


Figure 5b ¹H NMR spectra of 3-(4-bromophenyl)-5-phenyl-4,5-dihydropyrazole-1-carbaldehyde



%Transmittance





% Transmittance

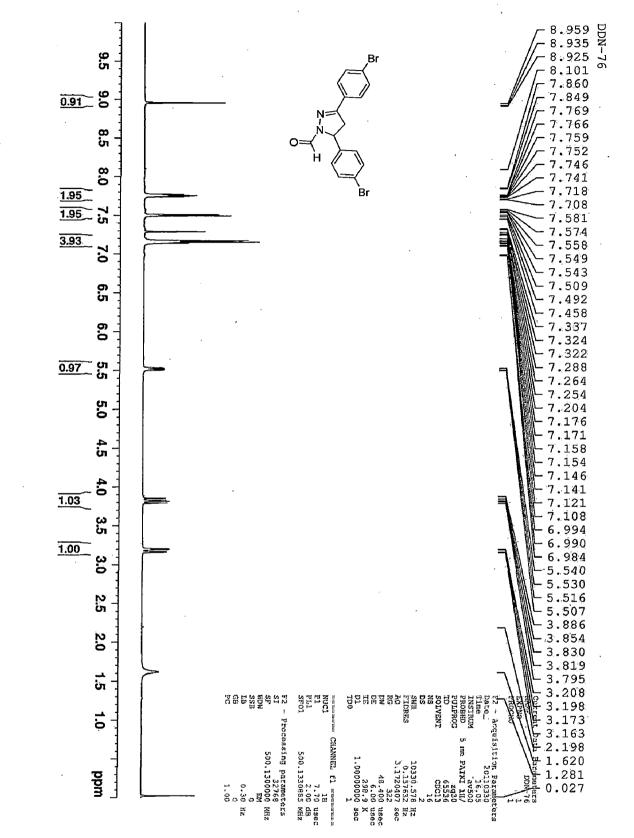
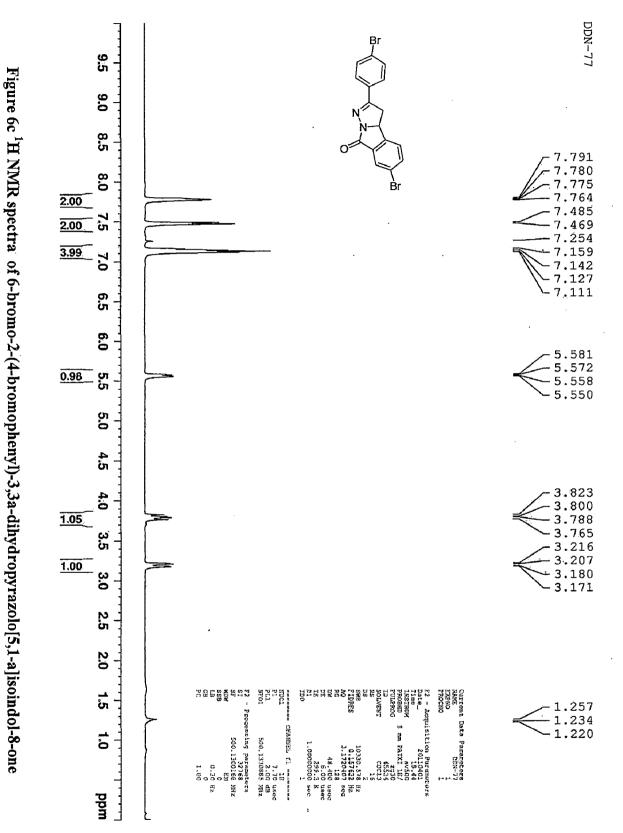
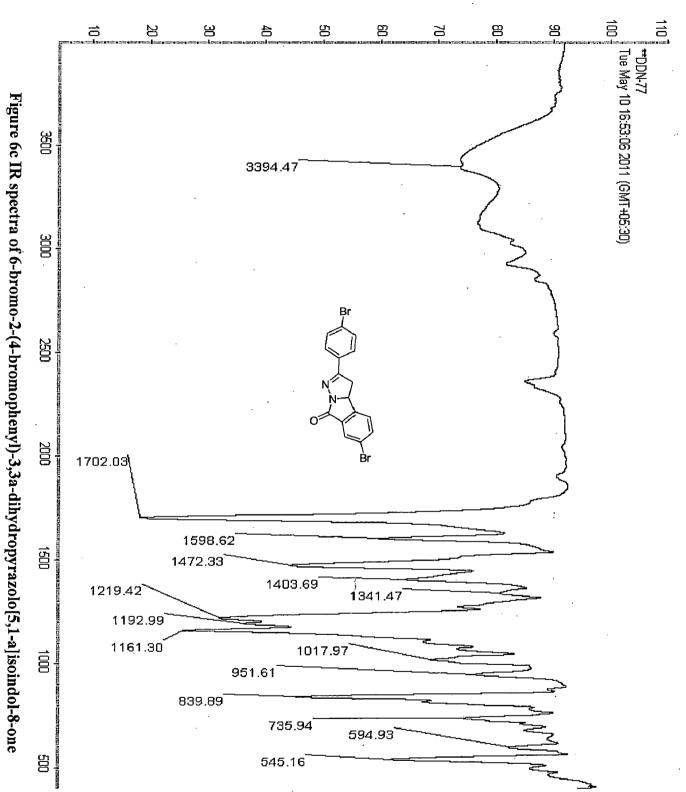
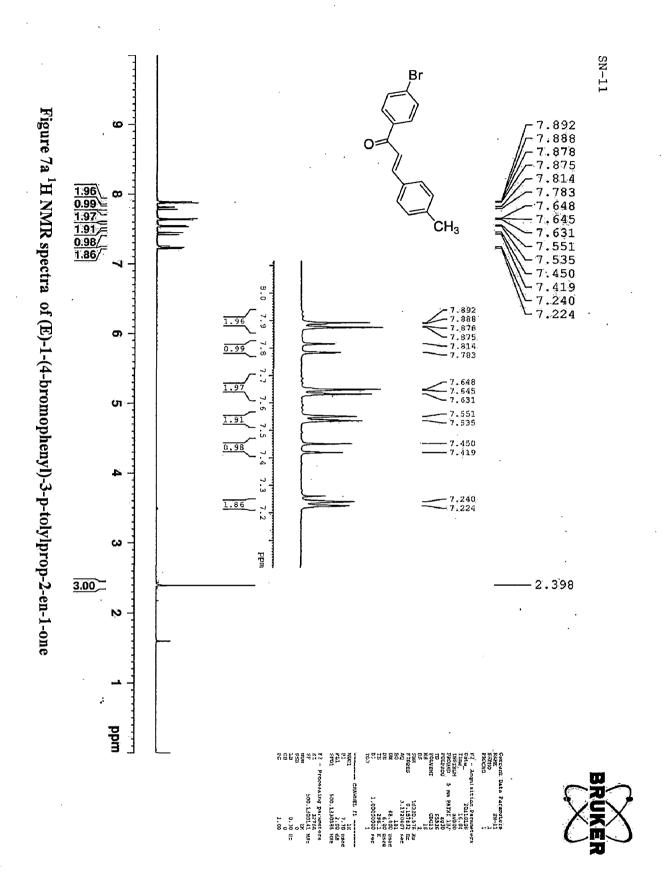


Figure 6b¹H NMR spectra of 3,5-bis(4-bromophenyl)-4,5-dihydropyrazole-1-carbaldehyde





% Transmittance





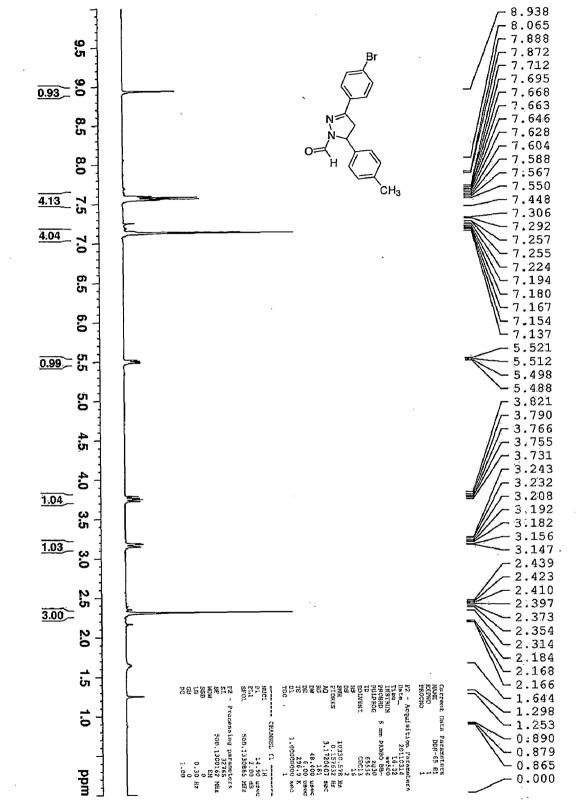


Figure 7b ¹H NMR spectra of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-carbaldehyde



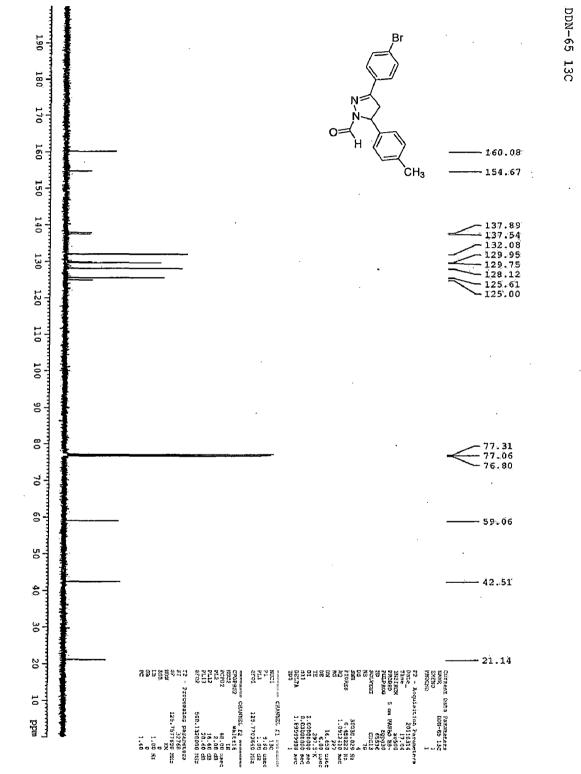


Figure 7b¹³C NMR spectra of 3-(4-bromophenyl)-5-p-tolyl-4,5-dihydropyrazole-1-carbaldehyde



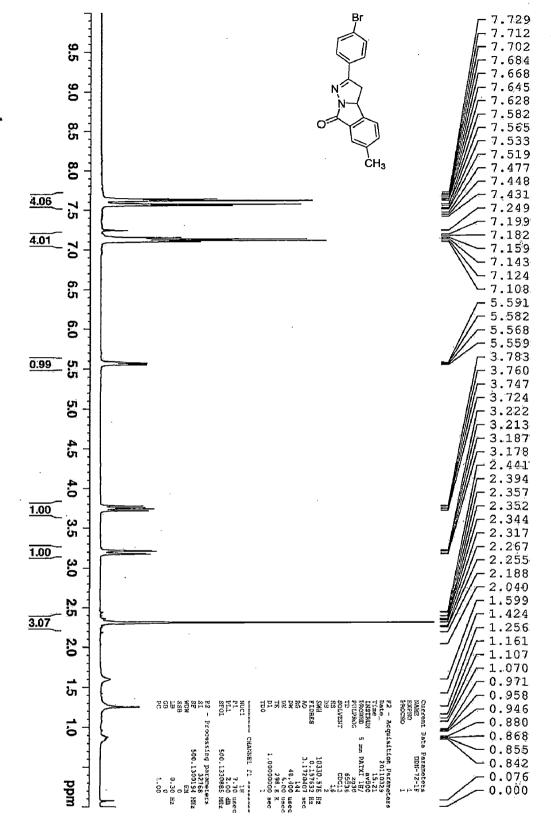
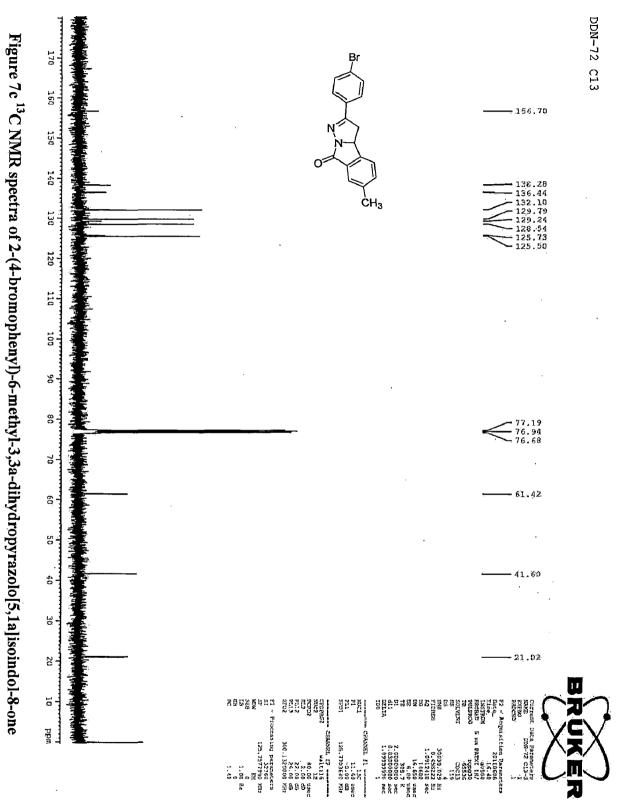


Figure 7c¹H NMR spectra of 2-(4-bromophenyl)-6-methyl-3,3a-dihydropyrazolo[5,1a]isoindol-8-one

113

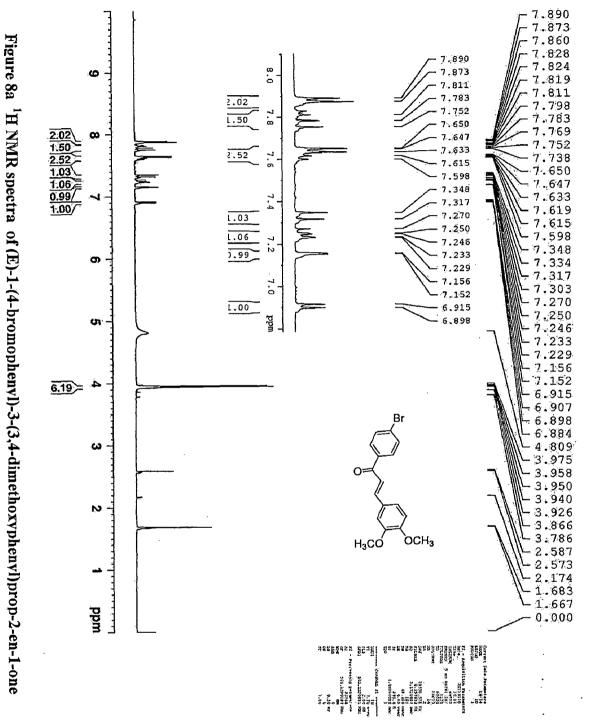
DDN-72-1F



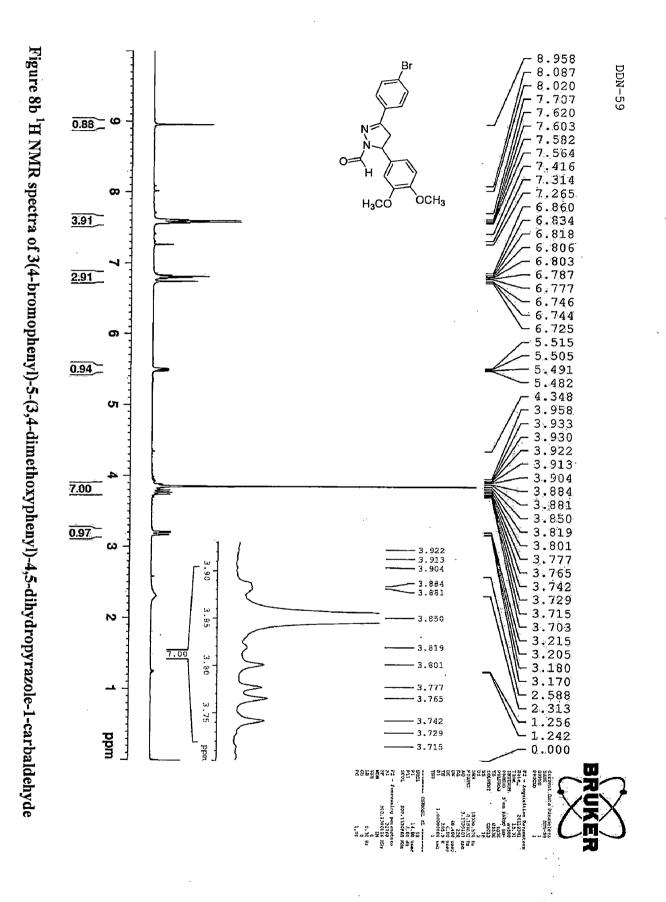




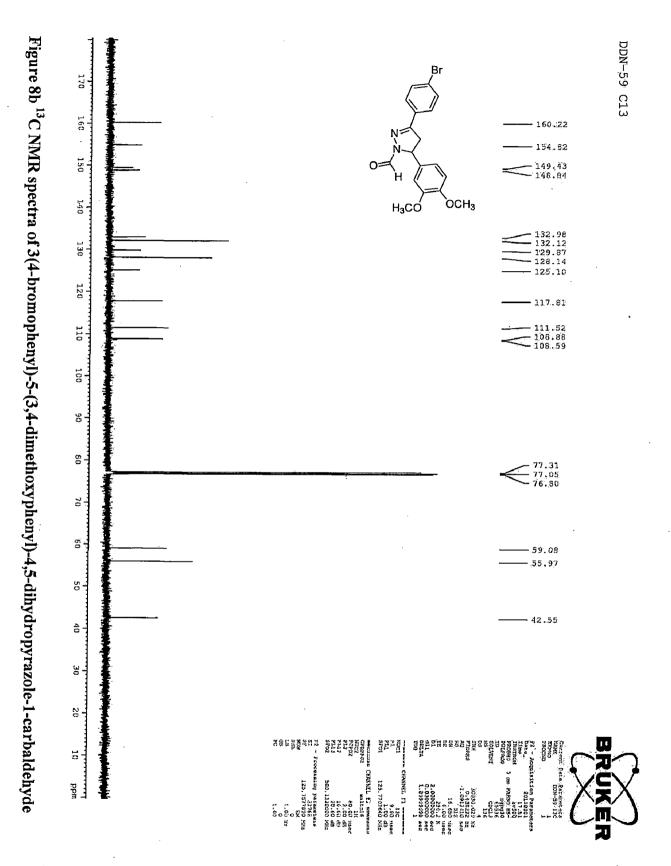
y tja



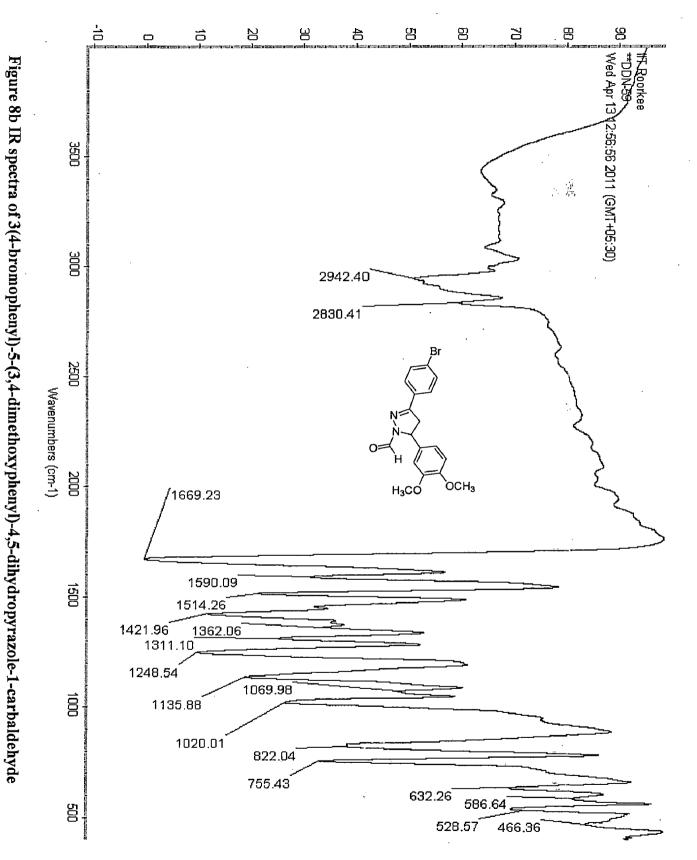
SN-14



ť,

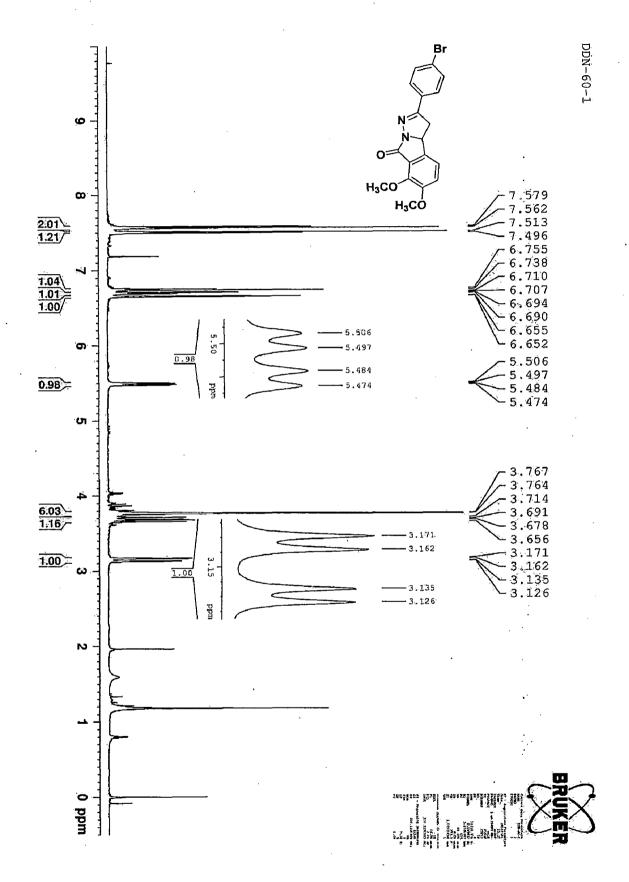


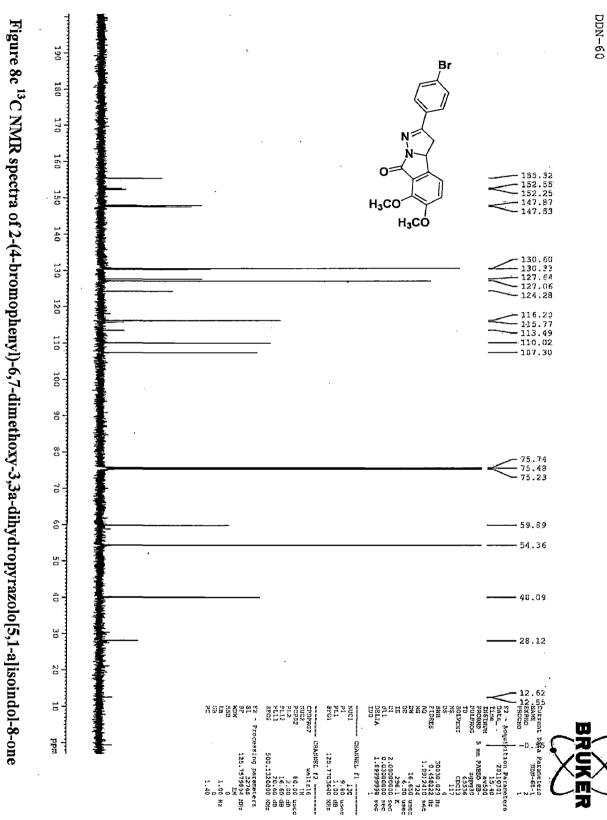
11/



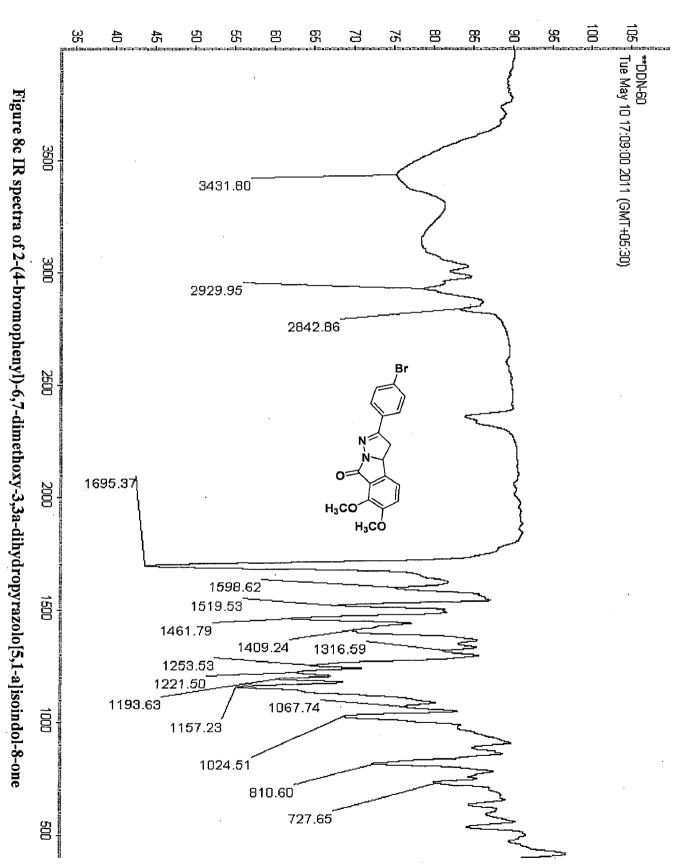
%Transmittance

Figure 8c¹H NMR spectra of 2-(4-bromophenyl)-6,7-dimethoxy-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one





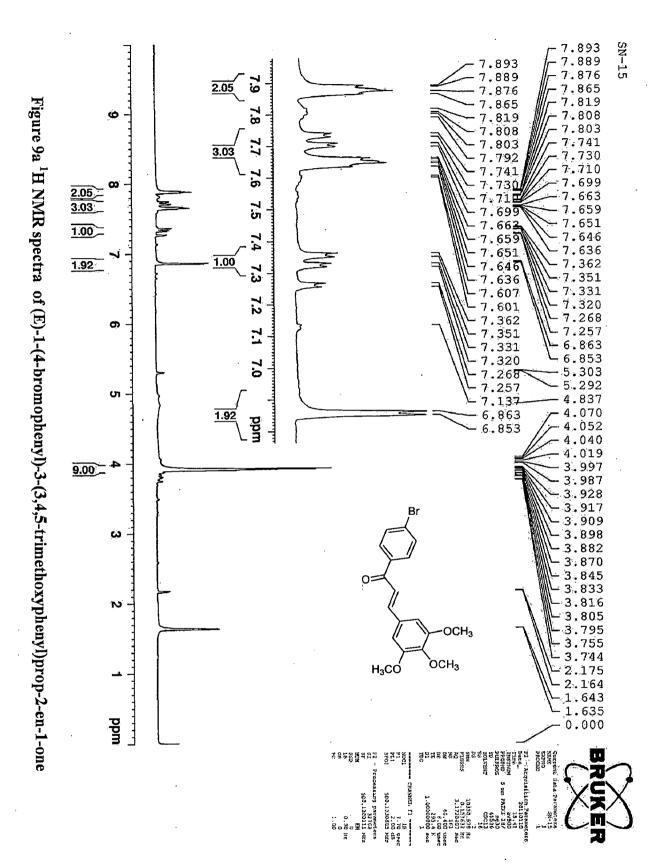
. . '.



%Transmittance

121

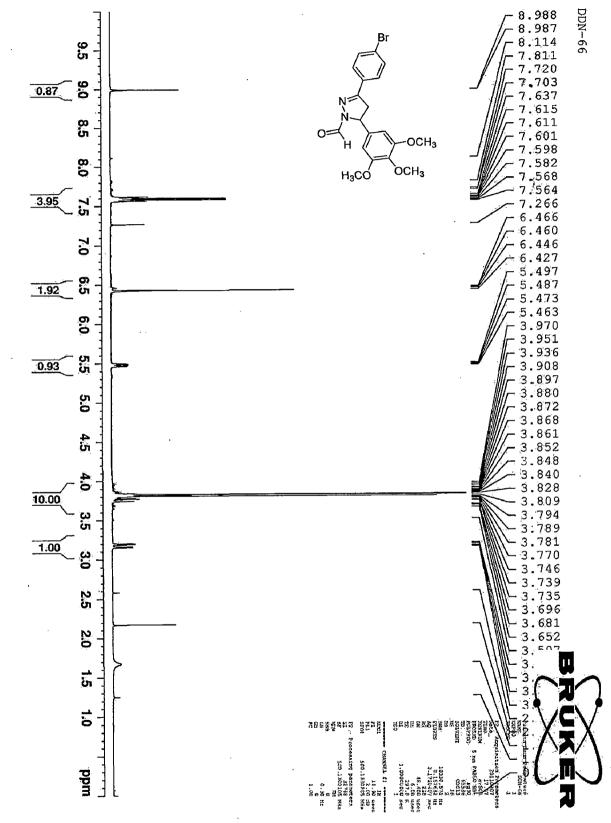
ŝ



•••

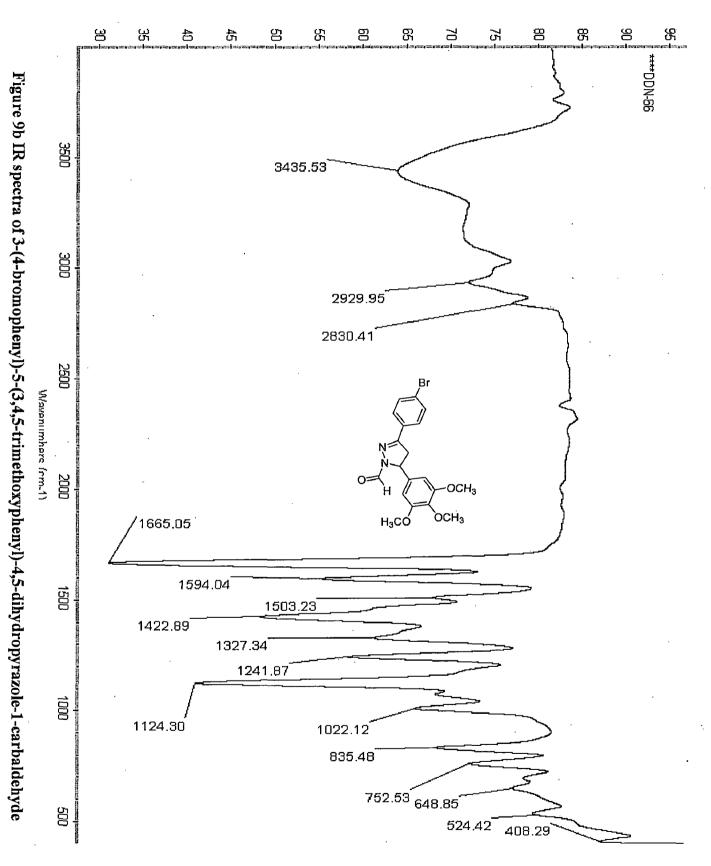
122

÷., j

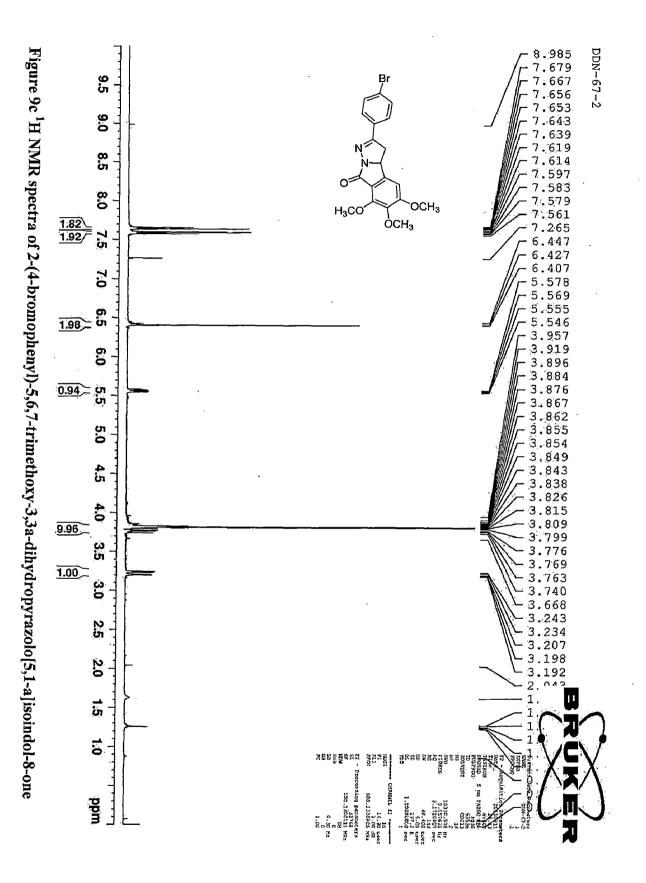


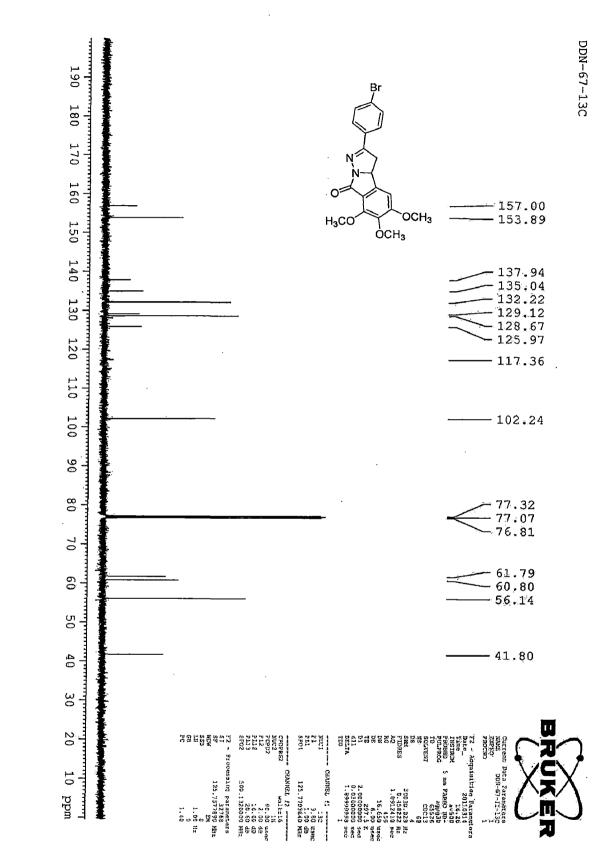
123

Figure 9b¹H NMR spectra of 3-(4-bromophenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde



%Transmittance



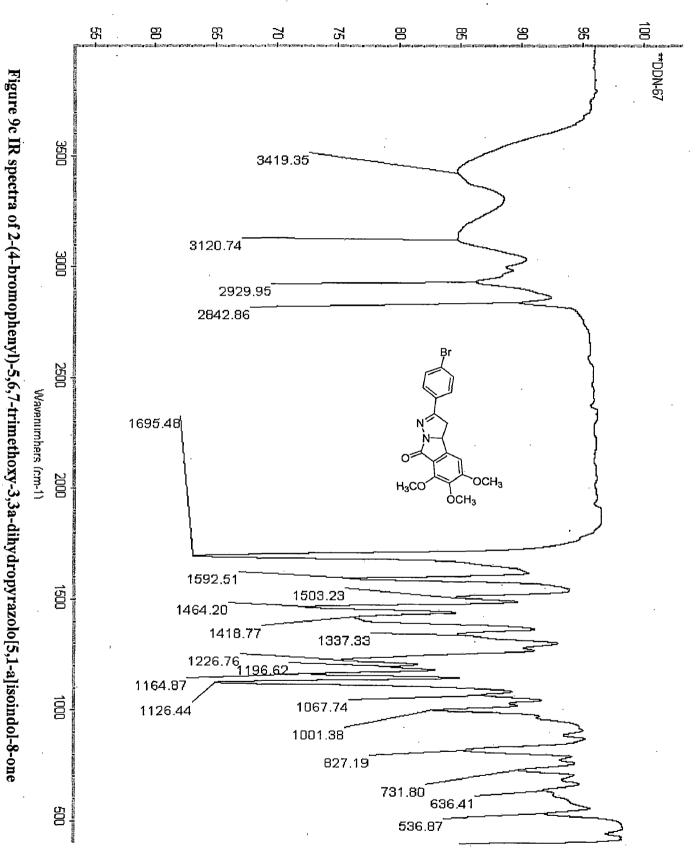


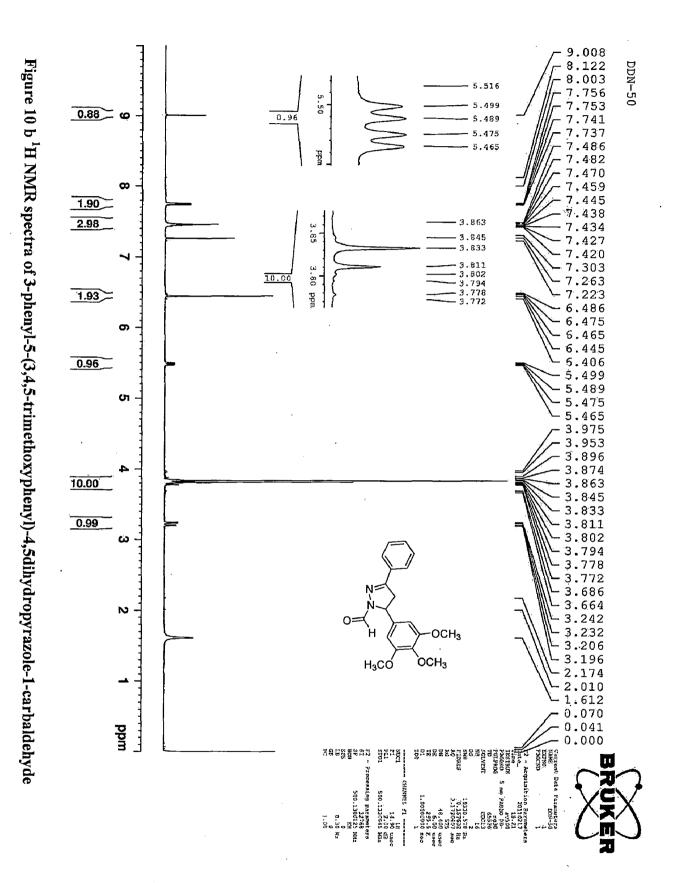


.

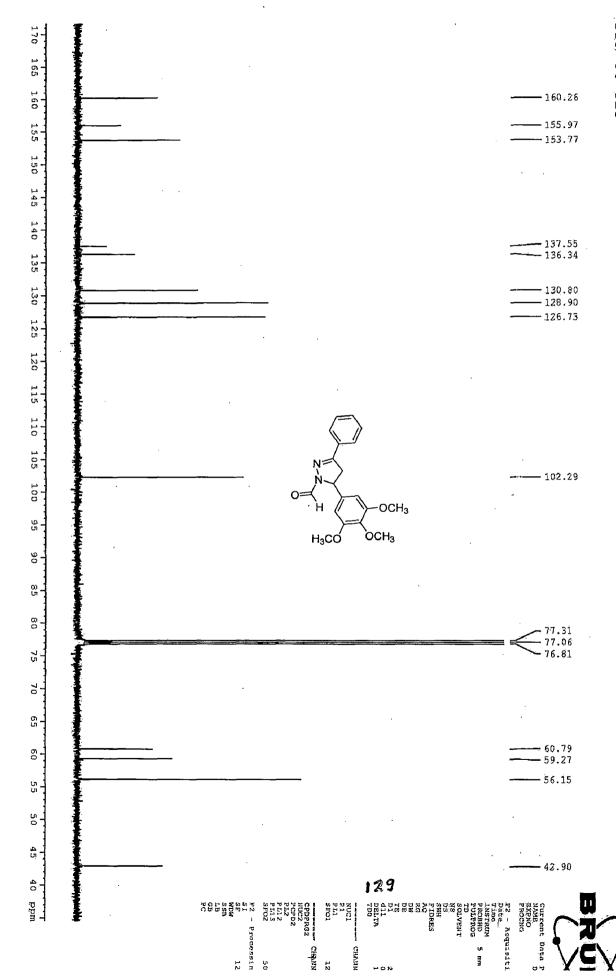








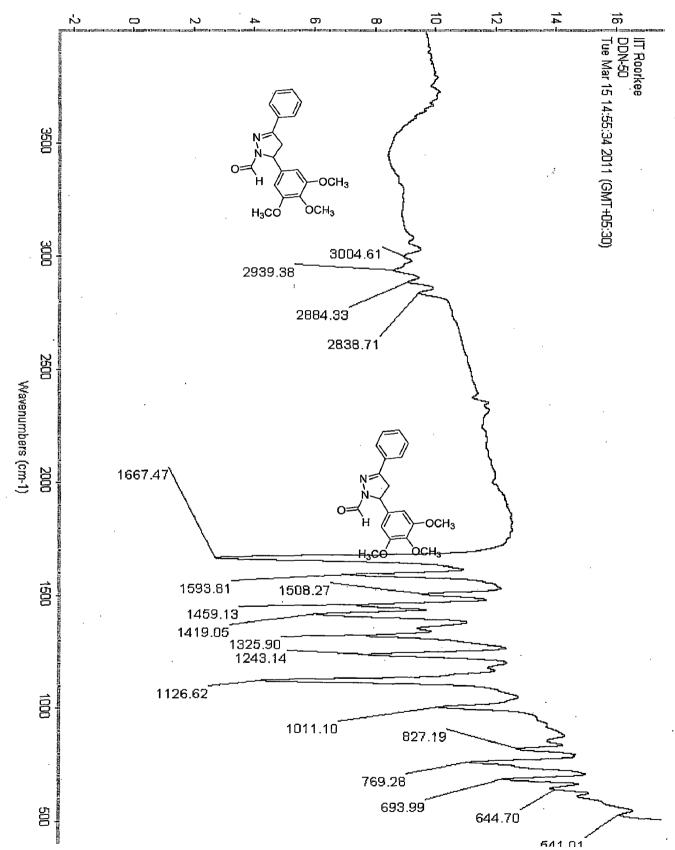
ан



12

Figure 10b ¹³C NMR spectra of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde

DDN-50 C13



%Transmittance

Figure 10b IR spectra of 3-phenyl-5-(3,4,5-trimethoxyphenyl)-4,5dihydropyrazole-1-carbaldehyde

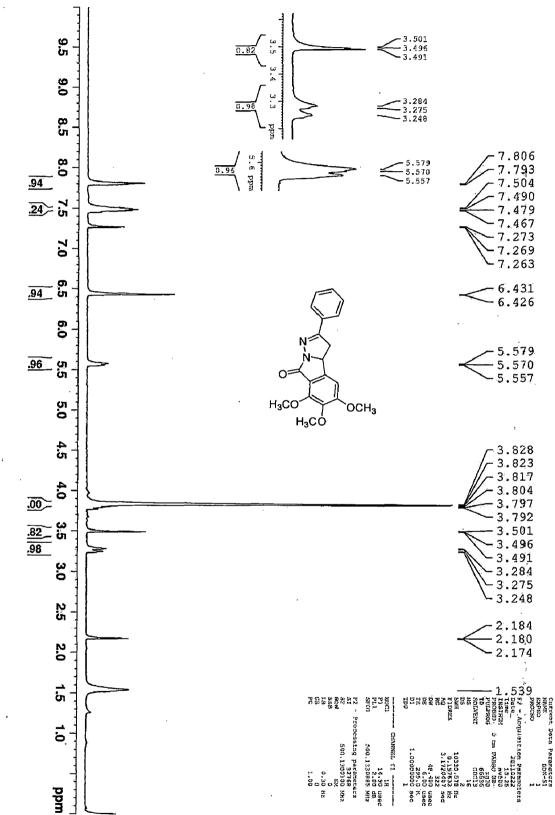
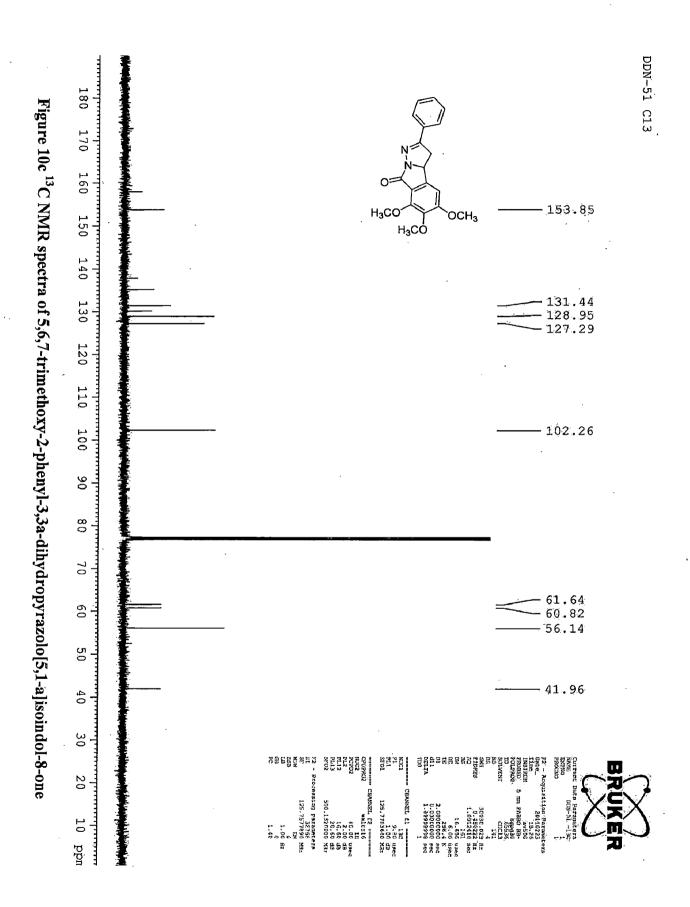


Figure 10c¹H NMR spectra of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one







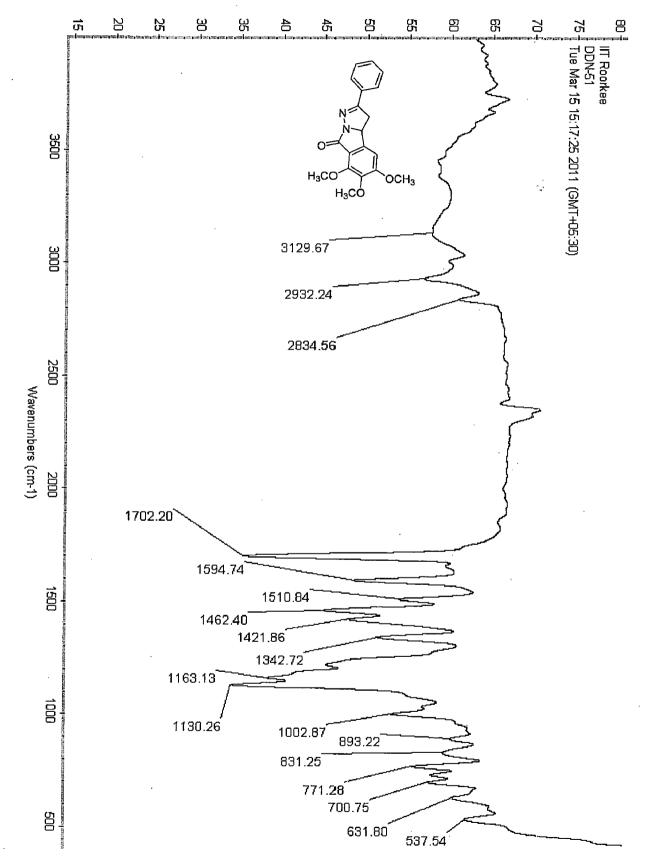


Figure 10c IR spectra of 5,6,7-trimethoxy-2-phenyl-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one

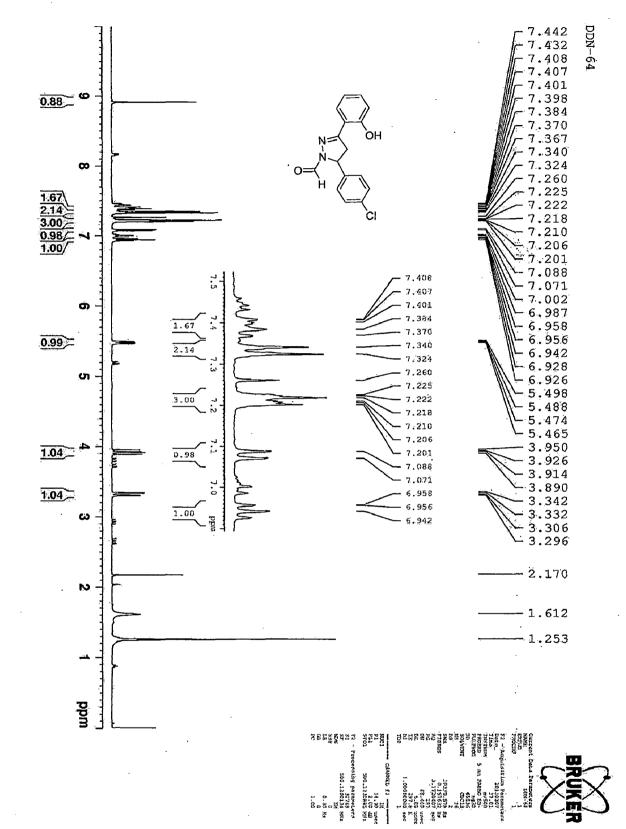
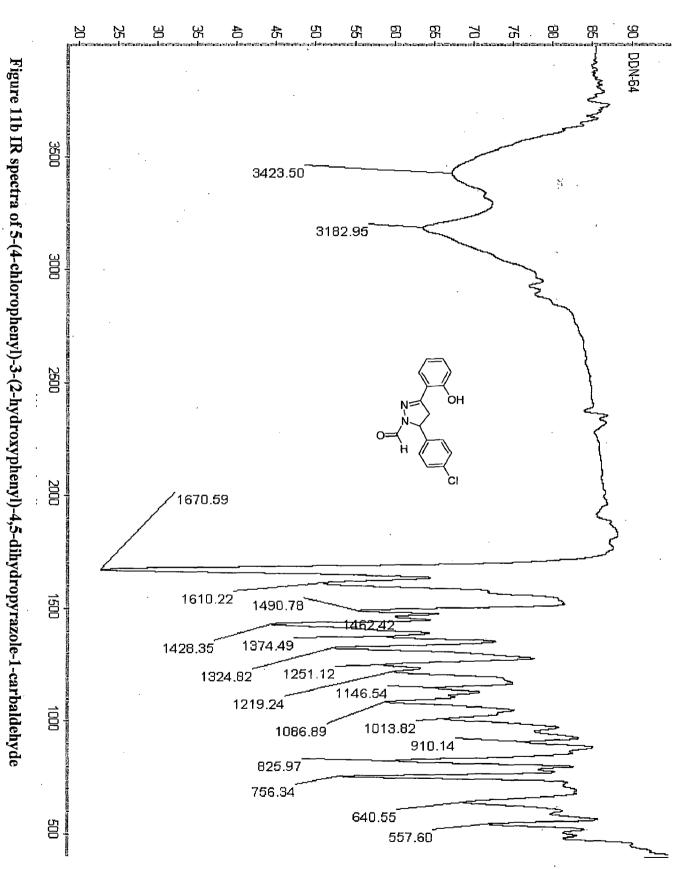
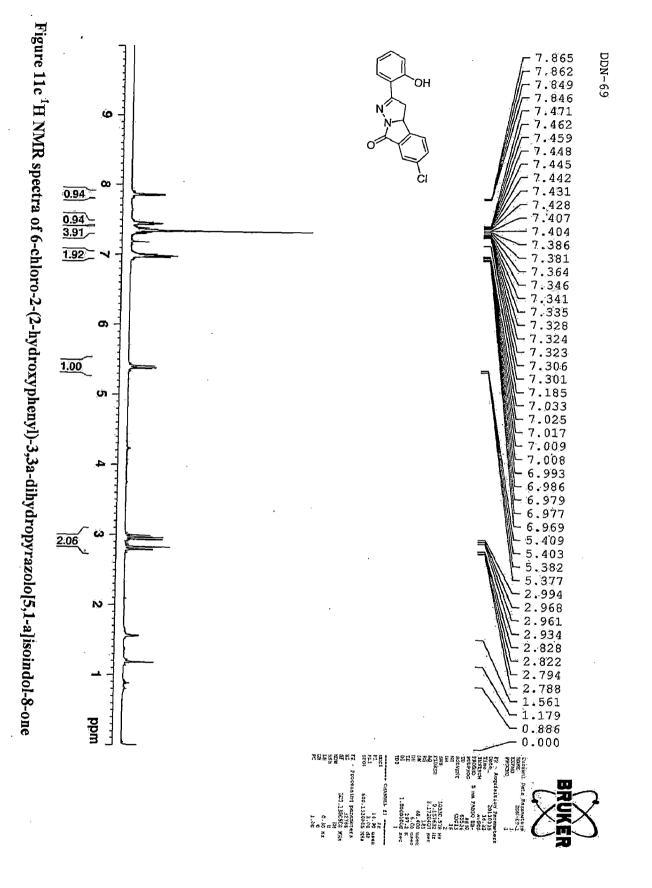


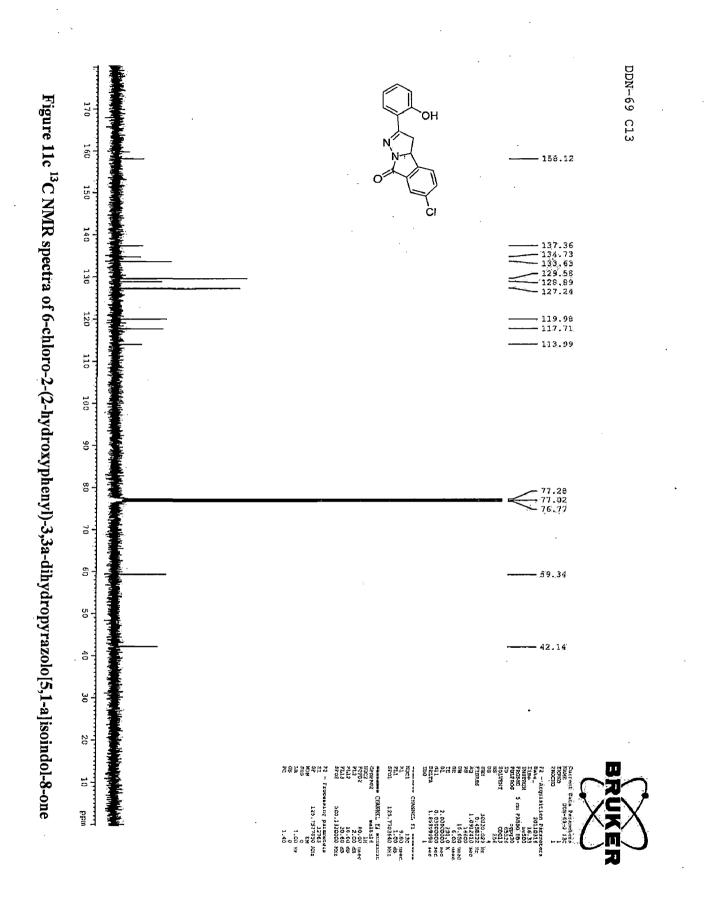
Figure 11b¹H NMR spectra of 5-(4-chlorophenyl)-3-(2-hydroxyphenyl)-4,5-dihydropyrazole-1-carbaldehyde

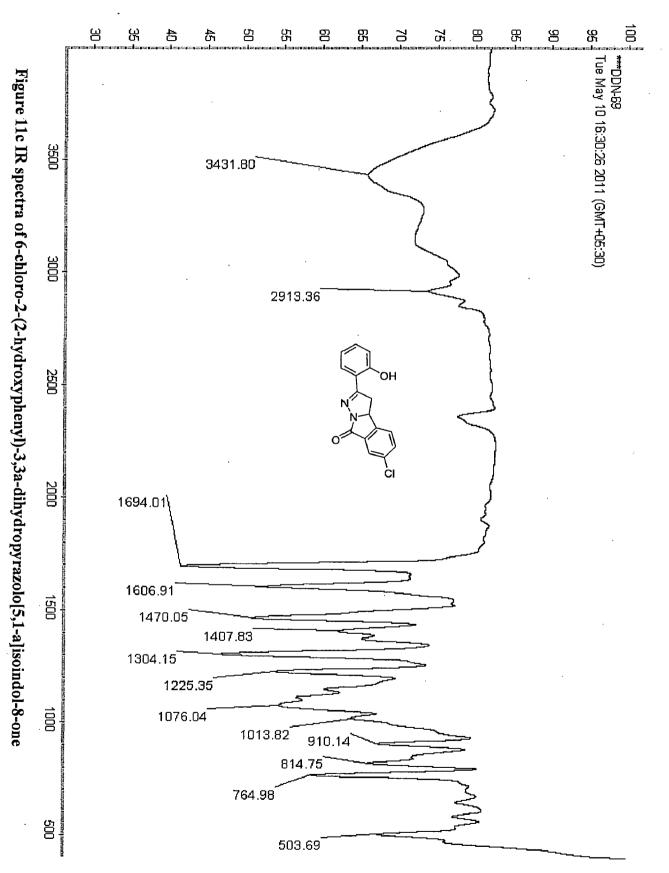


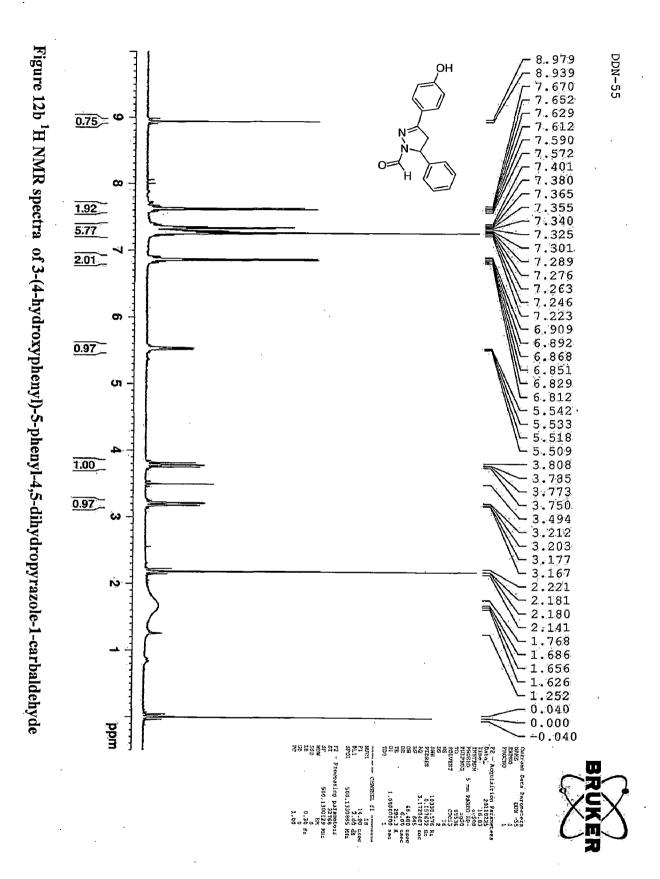


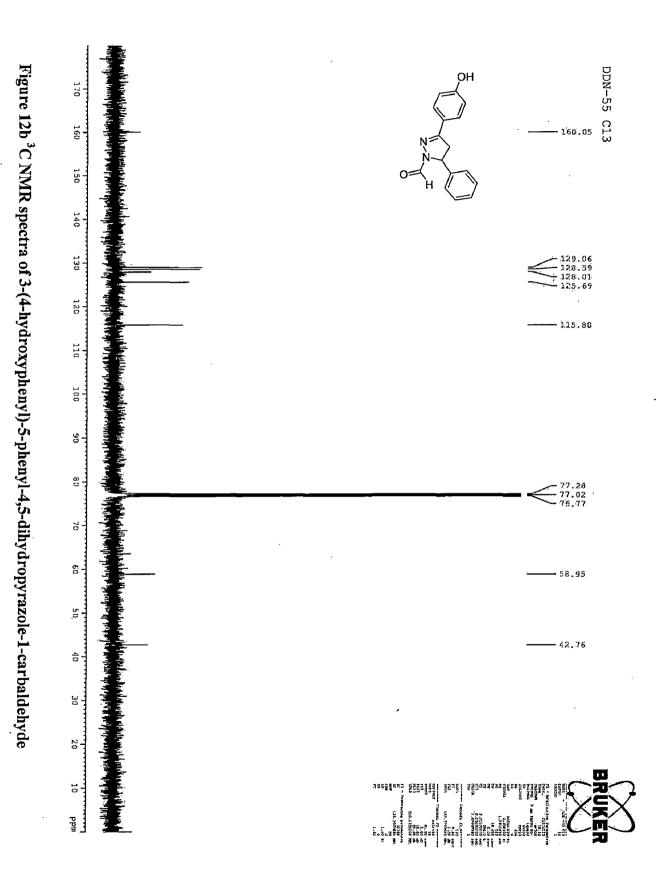


Ь



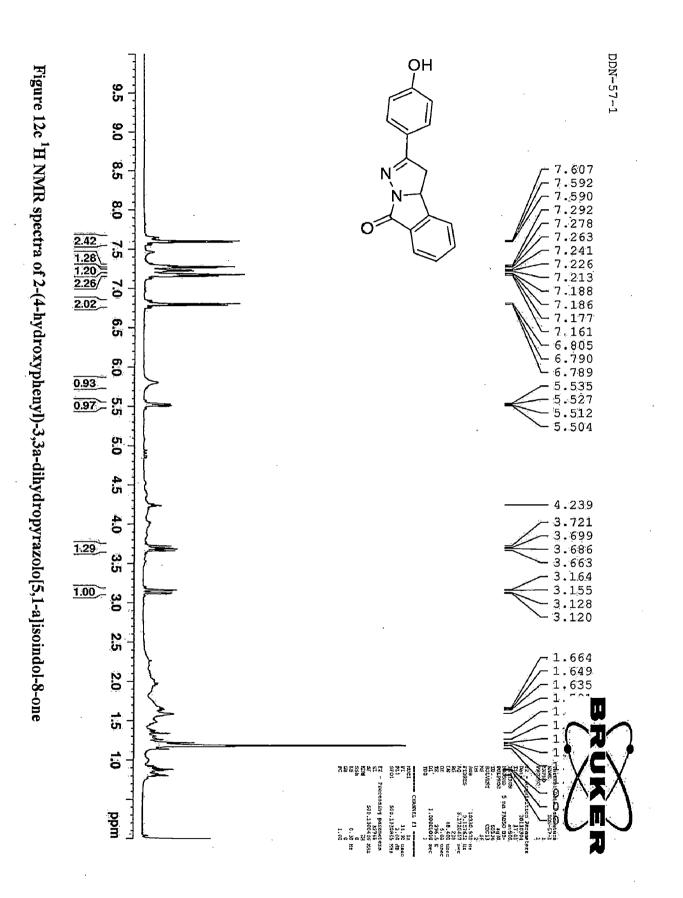


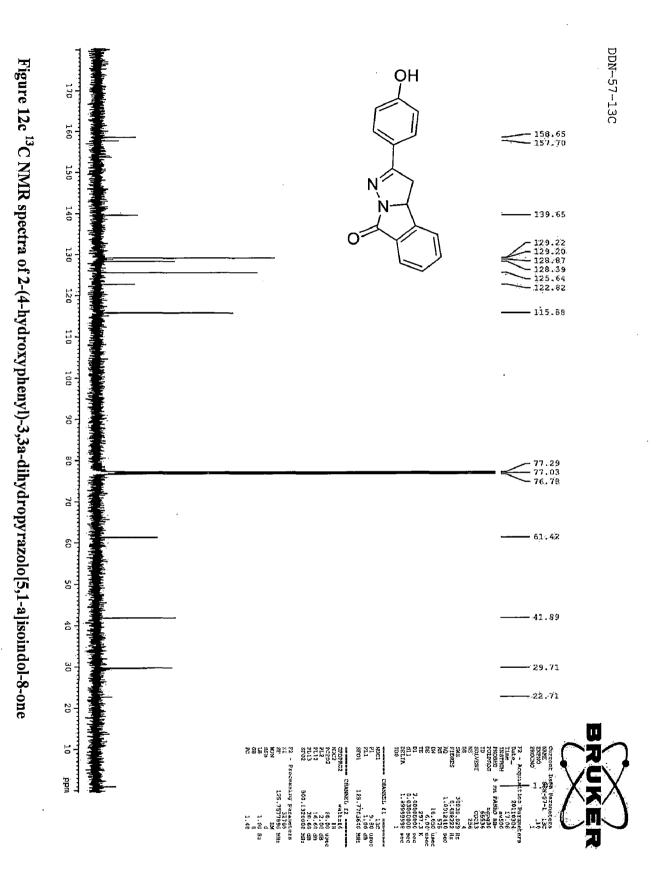




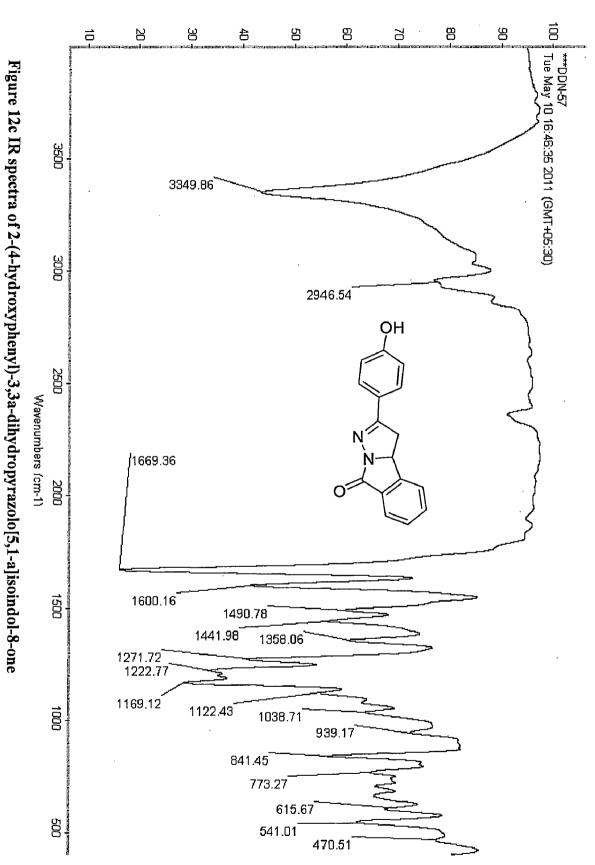


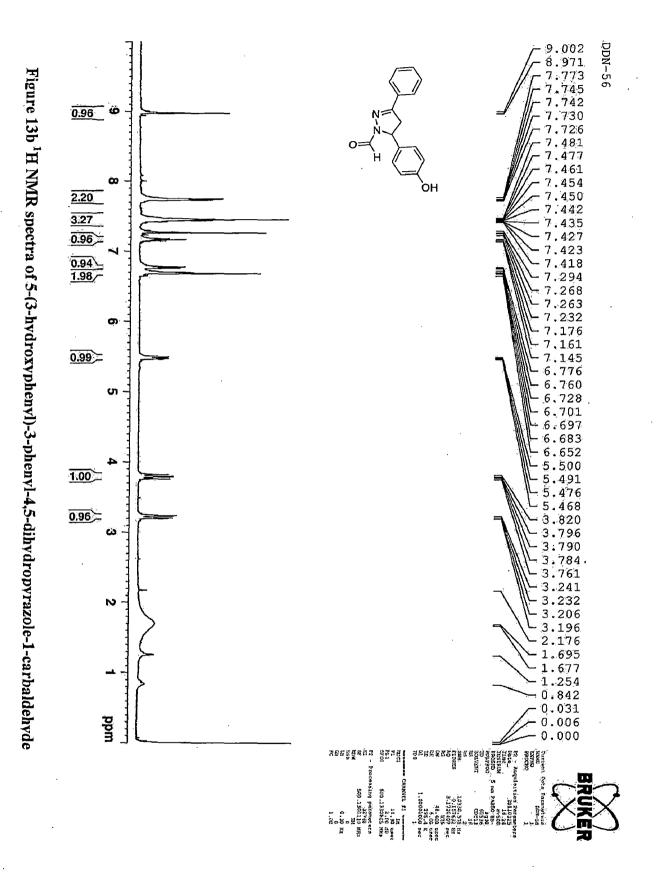
•

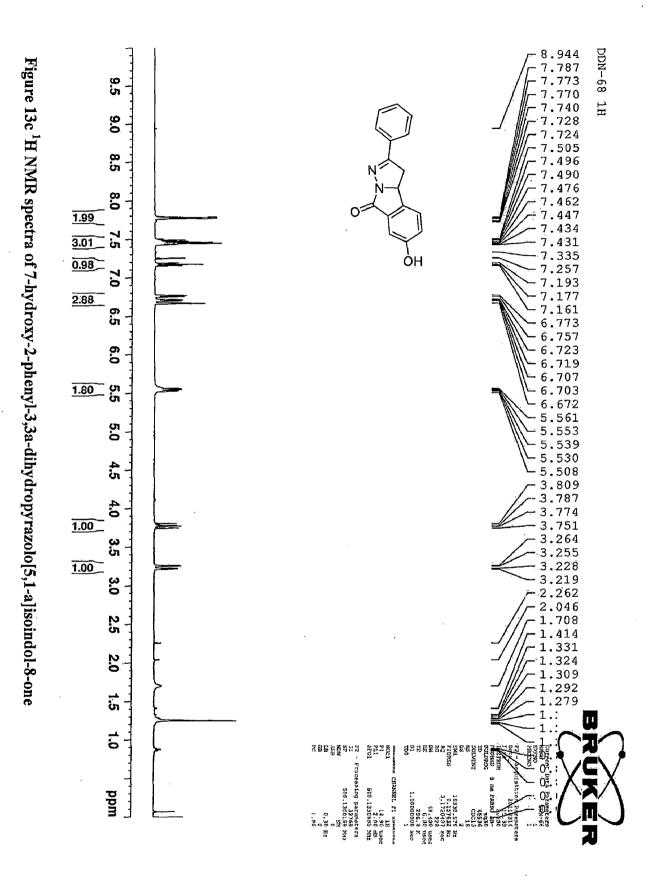


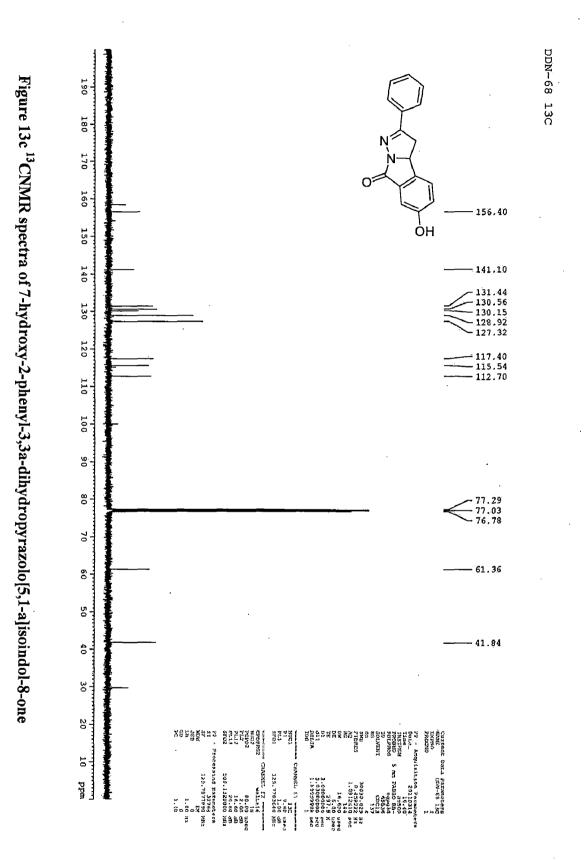


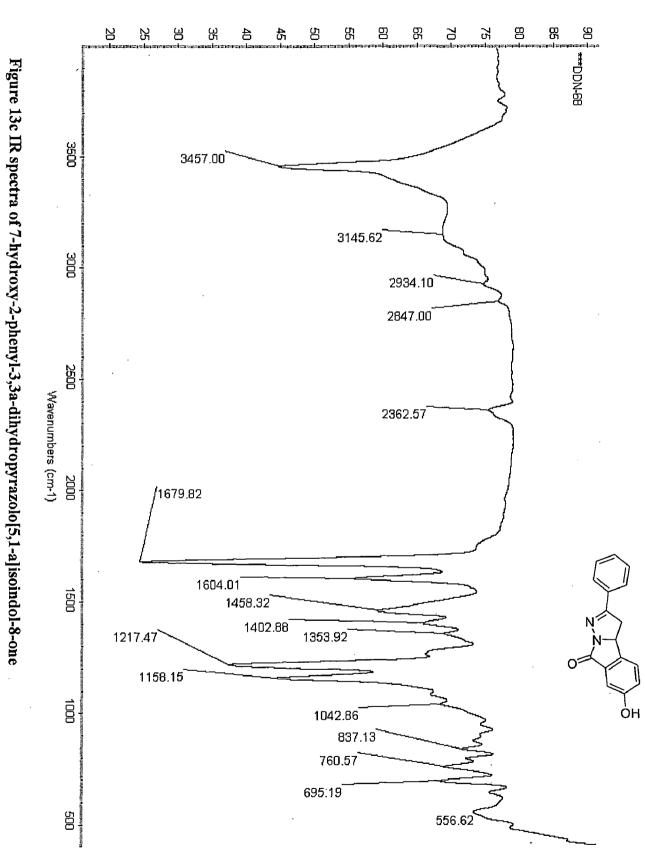




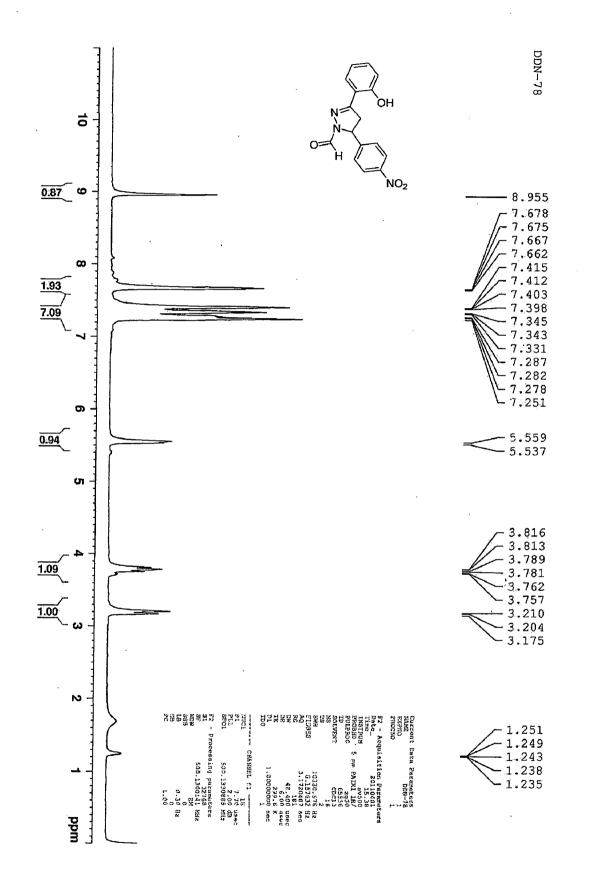


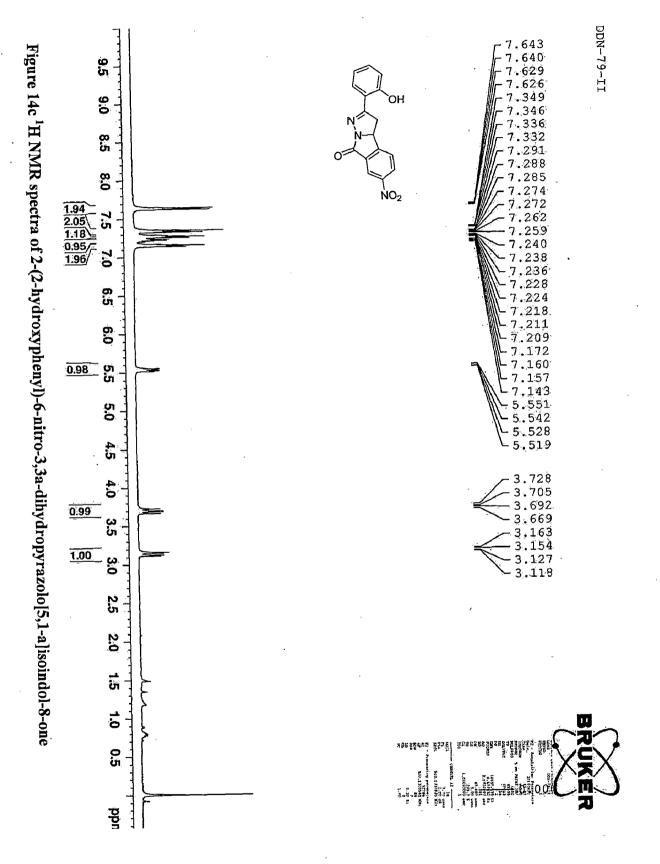




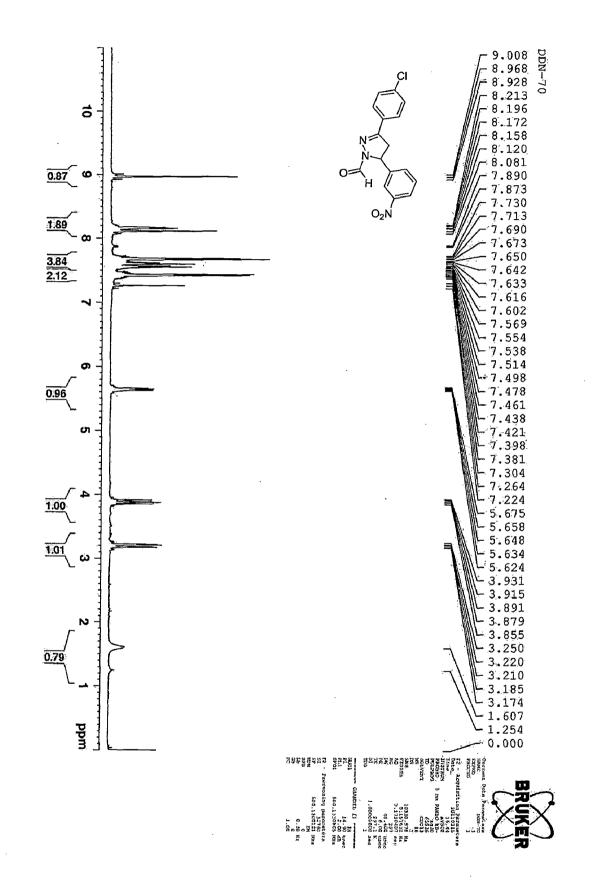












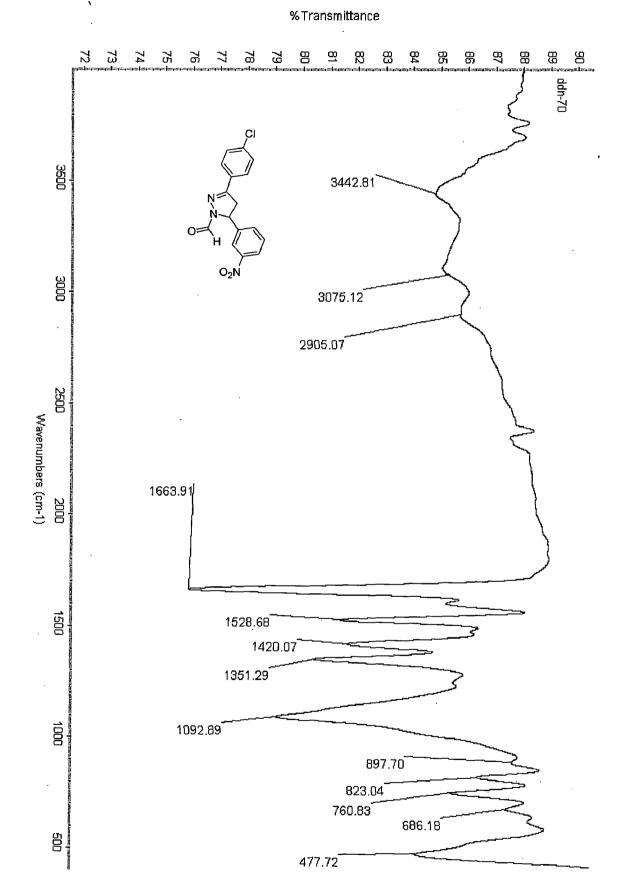
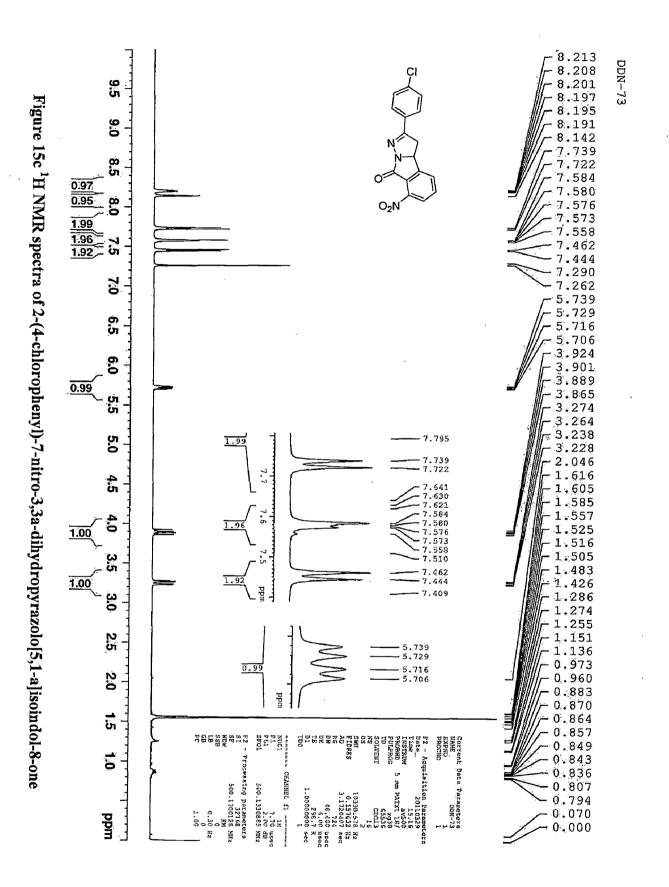


Figure 15b IR spectra of 3-(4-chlorophenyl)-5-(3-nitrophenyl)-4,5-dihydropyrazole-1-carbaldehyde



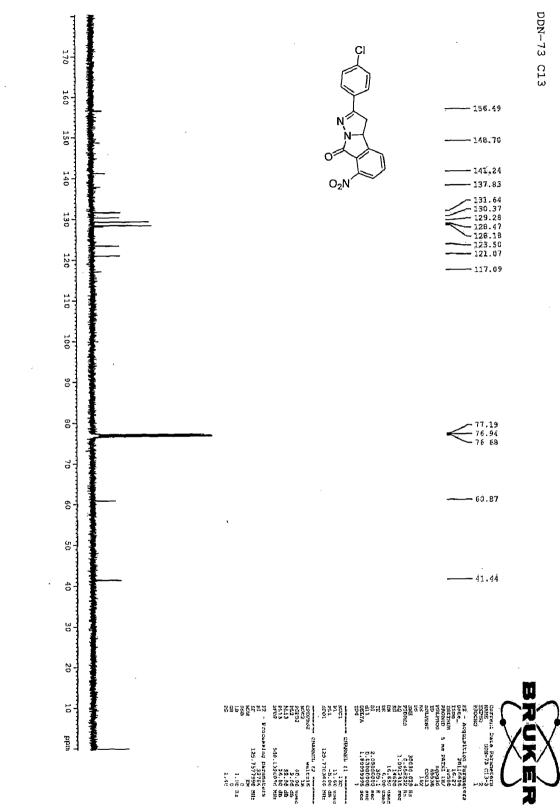


Figure 15c ¹³CNMR spectra of 2-(4-chlorophenyl)-7-nitro-3,3a-dihydropyrazolo[5,1-a]isoindol-8-one

)



